## Bio-Chemistry(Ph.D.)

1. What is the isoelectric point for phenylalanine given the $\mathrm{p} K_{\mathrm{a}}$ for the COOH group is 1.83 and the $\mathrm{NH}_{3}{ }^{+}$group is 9.13?
(A) 5.48
(B) 4.83
(C) 2.43
(D) 9.13
2. A sequence of amino acids in a certain protein is found to be-Ser-Gly-Pro-Gly-. The sequence is most probably part of a (n):
(A) Alpha helix
(B) Beta turn
(C) Parallel beta sheet
(D) Anti-parallel beta sheet
3. D-Glucose is called as a reducing sugar because it undergoes an oxidation-reduction reaction at the anomeric carbon. One of the products of this reaction is:
(A)
D-galactose
(B) D-D-glucurinate
(C) D-gluconate
(D) D-ribose
4. Side chains of tryptophan residues in proteins can interact with lectins. This is surprising because:
(A) The side chain of tryptophan is hydrophilic and lectins are hydrophobic
(B) The side chain of tryptophan is negatively charged and lectins are generally positively charged or are neutral
(C) The side chain of tryptophan can make hydrogen bonds and lectins cannot
(D) The side chain of tryptophan is hydrophobic and lectins are generally hydrophilic
5. The uncommon amino acid selenocysteine has an R group with the structure $-\mathrm{CH}_{2}-\mathrm{SeH}$ with a pKa of approximately 5 . In an aqueous solution, $\mathrm{pH}=7.0$, selenocysteine would:
(A) Be fully ionized zwitterion with no net charge
(B) Be nonionic
(C) Never be found in a protein
(D) Not be optically active
6. Prosthetic groups in proteins known as Glycoproteins are composed of:
(A) Carbohydrate (B)
Lipids
(C) Metals
(D) Phosphates
7. A major advance in the application of mass spectrometry to macromolecule came with the development of techniques to overcome which One of the following problems:
(A) Macromolecules were insoluble in solvents used in mass spectrometry
(B) Mass spectrometric analyses were too complex
(C) Mass spectrometric analysis involved molecules in gas phase
(D) Most macromolecules could not be purified to the degree needed for this technique
8. An individual molecular structure within an antigen to which an individual antibody binds is :
(A) Antigen
(B) Epitope
(C) Fab region
(D) Fc region
9. The biological role of restriction enzymes is:
(A) Restrict the damage to DNA by ultraviolet light
(B) Make bacteria resistant to antibiotics
(C) Degrade foreign DNA that enters a bacterium
(D) Restrict the size of DNA in certain bacterium
10. While in laboratory, you will introduce recombinant plasmids into bacterial cells by:
(A) Transformation- heat shock of the cells incubated with plasmid DNA in presence of $\mathrm{CaCl}_{2}$
(B) Mixing plasmids with an extract of broken cells
(C) Electrophoresis- a gentle low voltage gradient draws DNA into the cell
(D) Microinjection
11. Which one of the following analytical techniques does NOT help illuminate a gene's cellular function?
(A) DNA microarray analysis
(B) Protein chip analysis
(C) Southern blotting
(D) Two-hybrid analysis
12. Tay-Sachs disease is the result of a genetic defect in the metabolism of
(A) Sterols
(B) Gangliosides
(C) Triacylglycerols
(D) Vitamin D
13. Non steroidal anti-inflammatory drugs (NSAIDS) like aspirin act by blocking production of:
(A) Sphingolipids
(B) Ceramides
(C) Prostaglandins
(D) Waxes
14. Phosphatidylglycerol, bees wax and cholesterol were dissolved in chloroform and subjected to thin layer chromatography on silica gel using chloroform/methanol/water as developing solvent. Which statement is true for the given conditions:
(A) Bees wax will move fastest, followed by cholesterol; phophatidylglycerol will be slowest
(B) Cholesterol will move fastest, followed by Bees wax; phophatidylglycerol will be slowest
(C) Phophatidylglycerol will move fastest, followed by cholesterol; Bees wax will be slowest
(D) Bees wax will move fastest, followed by phophatidylglycerol; cholesterol will be slowest
15. Triple helical DNA structures can result from Hoogsteen (non Watson-Crick) interactions. These interactions are primarily:
(A) Covalent bonds involving deoxyribose
(B) Covalent bonds involving bases
(C) Hydrogen bonds involving the bases
(D) Hydrophobic interactions involving the bases
16. Which of the following statements about a plot of $\mathrm{V}_{\mathrm{o}}$ vs. [ S ] for an enzyme that follows Michelis Menten kinetics is false?
(A) At very high [S], initial velocity curve becomes a horizontal line that intersects
the y -axis at Km
(B) As [S] increases, initial velocity of reaction Vo also increases
(C) Km is the $[\mathrm{S}]$ at which $\mathrm{Vo}_{0}=1 / 2 \mathrm{Vmax}$
(D) The shape of the curve is a hyperbola
17. Penicillin and related drugs inhibit the enzyme____this enzyme is produced by $\qquad$ .
(A) Beta lactamase, bacteria
(B) Transpeptidase, human cells
(C) Transpeptidase, bacteria
(D) Lysozyme, human cells
18. Enzyme X exhibits maximum activity at $\mathrm{pH}=6.9$. X shows a fairly sharp decrease in its activity when the pH goes much lower than 6.4. One likely interpretation of this pH activity is that:
(A) A Glu residue on the enzyme is involved in the reaction
(B) A His residue on the enzyme is involved in the reaction
(C) The enzyme has a metallic cofactor
(D) The enzyme is found in gastric secretions
19. Which of the following is a common compound shared by TCA cycle and Urea cycle?
(A) Alpha ketoglutarate
(B) Argino succinic acid
(C) Arginine
(D) Fumarate
20. A patient reports with polyuria, polydipsia, polyphagia and blood glucose levels of 320 $\mathrm{mg} / \mathrm{dl}$; characteristic of Diabetes mellitus. Which of the following would occur in this patient?
(A) Increased conversion of fatty acids to acetyl CoA
(B) Decreased synthesis of cholesterol in liver
(C) Increased fatty acid synthesis from glucose in liver
(D) Increased stores of triacylglycerol in adipose tissue
21. Which of the following is a compound formed from both: a hydroxylation with an enzyme requiring Vitamin C and a subsequent methylation:
(A) Histamine
(B) Creatinine
(C) Epinephrine
(D) Serotonin
22. Which of the following dairy products could be recommended to an individual with Lactose intolerance:
(A) Condensed milk
(B) Yogurt
(C) Cheese
(D) Ice cream
23. Catabolism of 1 mol of glucose to lactate in the glycolytic pathway is accompanied by the reduction of how many moles of $\mathrm{O}_{2}$ ?
(A) 2
(B) 0
(C) 4
(D) 8
24. As electrons are received and passed down the transport chain, the electron carriers are first reduced with the acceptance of the electron and then oxidized with loss of the electron. An individual poisoned by which of the following compounds has the most highly reduced state of most of the respiratory chain carriers?
(A) Puromycin
(B) Rotenone
(C) Carbon monoxide
(D) Chloramphenicol
25. Which of the following nutrients is rich in short and medium chain fatty acids:
(A) Milk
(B) Peanut oil
(C) Sunflower oil (D) Almond oil
26. The pH of the body fluids is stabilized by buffer systems. Which of the following buffers is the most effective buffer system at physiological pH ?
(A) Phosphate buffer
(B) Protein buffer
(C) Deoxyhemoglobin
(D) Bicarbonate buffer
27. Replication of DNA genome in human cells occurs only at a specified time during its life span and this period is referred to as:
(A) $S$ phase
(B) $\quad \mathrm{M}$ phase
(C) G1
(D) G2
28. The derepression of lac operon in the presence of lactose is an example of:
(A) Double negative regulation
(B) Double positive regulation
(C) Positive regulation
(D) Negative regulation
29. Which of the following is NOT a stop codon in prokaryotes?
(A) UAA
(B) UAG
(C) UGG
(D) UGA
30. DNA is assembled into nucleosomes with the help of special proteins called
(A) Prolamines
(B) Histones
(C) Chaperones
(D) Protamines
31. Pasteur effect is due to inhibition of glycolysis by high concentration of :
(A) ATP
(B) ADP
(C) AMP
(D) Creatine phosphate
32. Which of the following tests is undertaken to differentiate between Glucose and Fructose:
(A) Benedict's test
(B) Seliwanoff's test
(C) Molisch test
(D) Starch test
33. Sphingosine is not present in:
(A) Gangliosides
(B) Cerebrosides
(C) Sphingomyelin
(D) Plasmalogen
34. Which essential nutrient participates as a coenzyme in the transport of high energy electrons and $\mathrm{H}+$ ions during oxidative phosphorylation in the mitochondria?
(A) Folic acid
(B) Thiamine
(C) Niacin
(D) Ascorbic acid
35. Cholesterol is a precursor of all except:
(A) Bile salts
(B) Bilirubin
(C) Steroids
(D) Vitamin D
36. All amino acids except one, participate in phase 2 reactions of detoxification
(A) Serine
(B) Glycine
(C) Glutamine
(D) Cysteine
37. Hill reaction is concerned with the production of
(A) Reducing agents like $\mathrm{TPNH}_{2}$ or $\mathrm{NADPH}_{2}$ and ATP in mitochondria
(B) Oxygen due to photolysis in chloroplasts
(C) Reducing agent like $\mathrm{NADPH}_{2}$ and $\mathrm{O}_{2}$ from $\mathrm{H}_{2} \mathrm{O}$ in chloroplasts
(D) Reducing agents like $\mathrm{TPNH}_{2}$ and ATP in chloroplasts
38. The C4 plants are photosynthetically more efficient than C3 plants because
(A) They have more chloroplasts
(B) $\quad \mathrm{CO}_{2}$ compensation point is high
(C) $\quad \mathrm{CO}_{2}$ efflux is not prevented
(D) $\quad \mathrm{CO}_{2}$ generated during photorespiration is trapped through PEP carboxylase
39. Artificial ripening of fruits is accomplished by treatment with
(A) IAA
(B) Ethylene gas
(C) Kinetin
(D) Sodium Chloride
40. Majority of Nitrogen fixation occurs by
(A) Lightening
(B) Volcanic reuptions
(C) Haber Bosch process
(D) Biological nitrogen fixing organisms
41. The major enzymes involved in Nitrogen fixation are
(A) Nitorgenase and hexokinase
(B) Nitrogenase and hydrogenase
(C) Nitrogenase and hydrolyase
(D) Nitrogenase and peptidase
42. Which part of the brain controls eating, drinking, body temperature \& provides a link between brain and endocrine system?
(A) Parietal lobes
(B) Temporal lobes
(C) Amygdala
(D) Hypothalamus
43. The technique used to locate specific genes in chromosomes is
(A) In situ hybridisation
(B) Colony hybridisation
(C) Dot blot technique
(D) Western blotting
44. Reverse transcriptase PCR uses
(A) DNA as a template to form ssDNA
(B) RNA as a template to form DNA
(C) mRNA as a template to form cDNA
(D) DNA as a template to form dsDNA
45. Minisatellites are
(A) Short coding repetitive regions on the eukaryotic genome
(B) Short non coding repetitive sequences present throughout the chromosome
(C) $\quad 10-40 \mathrm{bp}$ sized short sequences within the genes
(D) Regions of chromosomes after secondary constriction
46. The use of living microorganisms to degrade environmental pollutants is called
(A) Microremediation
(B) Nanoremediation
(C) Bioremediation
(D) Georemediation
47. Which of the following chelating agents is recommended for acute Lead poisoning with signs of encephalopathy?
(A) Dimercaprol and Calcium EDTA
(B) Succimer
(C) Penicillamine
(D) Calcium EDTA
48. Which of the following Mast cell products is not preformed and therefore has to be newly synthesized?
(A) Eosinophil chemotactic factor
(B) Heparin
(C) Histamine
(D) Prostaglandin
49. The circulation of a two month old breast fed baby will contain maternal:
(A) $\operatorname{Ig} \mathrm{A}$
(B) $\operatorname{IgG}$
(C) $\operatorname{IgM}$
(D) $\operatorname{IgD}$
50. For vaccination against mycobacterial diseases such as tuberculosis, the most important facet of the immune response to be stimulated is:
(A) High titer of antibody
(B) Cytotoxic T cells
(C) Macrophage activating cell mediated immunity
(D) Neutrophils

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## (BIOPHYSICS)

1. The most significant drawback in electron microscope is that
(A) It is very costly
(B) It requires high technical skill
(C) It is to be lodged in vacuum
(D) Living cells cannot be observed under it
2. Mycobacterium tuberculosis is an intra-cellular bacterium. It prefers to infect
(A) Macrophages
(B) $\quad \mathrm{B}$-cells
(C) T-cells
(D) Neutrophils
3. What does $\theta$ signifies in Brags equation: $2 \mathrm{~d} \operatorname{Sin} \theta=\mathrm{n} \lambda$
(A) Angle between incident x-ray and diffracted x-ray
(B) Angle between incident x -ray and plane of crystal
(C) Angle between incident $x$-ray and protein molecule
(D) Angle between incident x -ray and x -ray grid
4. Which gene transfer technique involves the use of a fatty bubble to carry a gene into a somatic cell?
(A) Electroporation
(B) Liposome transfer
(C) Microinjection
(D) Particle bombardment
5. What reagent is quantified when performing an indirect ELISA?
(A) Antibodies in patient's serum
(B) Fluorescent antibody
(C) Chromagen
(D) Complement
6. The vector for T-DNA is:
(A) Thermusaquaticus
(B) Agrobacterium tumefaciens
(C) Bacillus thuringiensis
(D) Salmonella typhimurium
7. Asprin, used as a common analgesic, antipyretic and anti-inflammatory agent, inhibits the synthesis of which one of the following?
(A) Arachidonic acid
(B) Prostaglandins
(C) Glucocorticoids
(D) Histamine
8. The $S$ wave form of normal human ECG originates due to:
(A) Septal and left ventricular depolarization
(B) Late depolarization of the ventricular walls moving back towards the AV junction
(C) Left to right septal depolarization
(D) Repolarization of atrium
9. Which one of the methods listed below is the most sensitive label free quantification methods for proteins?
(A) UV spectroscopy
(B) Infra-red spectroscopy
(C) Raman spectroscopy
(D) Southern blotting
10. Which of the following proteins acts as an energy transducer?
(A) G-protein
(B) Bacteriorhodopsin
(C) Haemoglobin
(D) Heat shock protein
11. Which of the following is unfavorable for protein folding?
(A) Hydrophobic Interaction
(B) Van DER Waals Interaction
(C) Conformational Entropy
(D) Hydrogen Bonding
12. Recombinant proteins are
(A) Proteins synthesized in animals
(B) Proteins synthesized by transgene in host cell by rDNA technology
(C) Proteinssynthesised in cells that are produced by protoplast fusion
(D) Proteins synthesized in mutated cell lines
13. Immunoprecipitation involve the purification of
(A) Antigen
(B) Antibody
(C) Antigen and antibodies
(D) Antigen-antibody complex
14. Bouguer's law relates
(A) Light reflection absorption
15. Which among the following is the simplest method to estimate the concentration of glycerol in an aqueous solution of glycerol?
(A) UV absorption spectroscopy
(B) Gas chromatography
(C) pH measurement
(D) Viscosity measurement
16. Choose the correct statement about peptides in Ramchandran plot.
(A) Peptides that are unstructured will have all the backbone dihedral angles in the disallowed regions
(B) It is not possible to conclude whether a peptide adopts entirely helix or entirely beta sheet conformation
(C) The occurrence of beta turn conformation in a peptide can be deduced
(D) The sequence of a peptide can be deduced
17. An optical measurement of protein is taken both before and after digestion of the protein by a protease. In which of the following spectroscopic measurement the signal change, i.e., before $\mathrm{v} / \mathrm{s}$ after protease treatment, could be the maximum?
(A) Absorbance at 280 nm
(B) Circular dichromism
(C) Absorbance at 340nm
(D) Fluorescence value
18. The presence and distribution of specific mRNAs within a cell can be detected by
(A) Northern blot analysis
(B) RNase protection assay
(C) In situ hybridization
(D) Real-time PCR
19. Which one of the following analytical techniques does NOT involve an optical measurement?
(A) ELISA
(B) Microarray
(C) Flow cytometry
(D) Differential Scanning Calorimetry
20. The secondary antibodies routinely used for the detection of primary antibodies in western blotting experiment are
(A) Anti-allotypic
(B) Anti-idiotypic
(C) Anti-isotypic
(D) Anti-paratypic
21. Histone deacytalase (HDAC) catalyses the removal of acetyl group from N -terminal of histones. Which amino acid of histone is involved in this process?
(A) Lysine
(B) Arginine
(C) Asparagine
(D) Histidine
22. Where are CD-45 cells responsible for signal transduction in the body located?
(A) B-cells
(B) T -cells
(C) All leukocytes except T-cells
(D) Haemopoietic cells
23. Which of the following antibiotics work by inhibiting protein synthesis in the target organism?
(A) Quinolones and Bacitracin
(B) Streptomycin and Tetracycline
(C) Rifampicin and Cephalosporin
(D) Tetracycline and Ampicillin
24. Which one of the following chemicals is a DNA intercalator?
(A) 5-Bromouracil
(B) Ethyl methane sulfonate
(C) UV
(D) Acridine orange
25. The appropriate match in Column 1 and 2 is

|  | COLUMN 1 |  | COLUMN 2 |
| :--- | :--- | :--- | :--- |
| a | Ferritin | i | Carboplatin |
| b | Chemotheraphy | ii | Iron storage |
| c | Metallothioneins | iii | Cystein rich protein |

(A) a-i, b-ii, c-iii
(B) a- ii, b-i, c-iii
(C) a-i, b-iii, c-ii
(D) a- iii, b-i, c-ii
26. Which of the following groups contain only aromatic Aminoacids?
(A) Histidine, Serine, Tryptophan
(B) Phenylalanine, Tyrosine, Valine
(C) Tryptophan, Tyrosine, Histidine
(D) Alanine, Tyrosine, Valine
27. Immediate hypersentivity reactions are associated with?
(A) $\operatorname{IgG}$
(B) $\operatorname{IgE}$
(C) $\quad \mathrm{IgM}$
(D) $\operatorname{Ig} \mathrm{A}$
28. The Biological role of Cytochrome P-450 is
(A) Nitrogen Fixation
(B) Oxidation of RH to ROH
(C) Amide hydrolysis
(D) Oxidation of aminoacids
29. If half-life of 100 g iodine (I131) is 8 days, how many grams will remain in 32 days
(A) 50 g
(B) 75 g
(C) $\quad 6.25 \mathrm{~g}$
(D) 25 g
30. Which of the following statements is/are correct?
(A) Temperature coefficient of thermistors is more than the thermoresistors
(B) Temperature coefficient of thermoresistors is more than the thermistor
(C) The resistance-temperature relationship of thermistors is exponential and thermoresistor is linear
(D) The resistance-temperature relationship of thermoresistor is exponential and thermistor is linear
31. A neoplasm is
(A) A population of cells growing out of control
(B) A cell organelle that begins the process of turning a cell cancerous
(C) The nucleus of a cancer cell
(D) Pain caused by cancer cells putting pressure on neurons
32. "A single ring-shaped DNA molecule" is a description of
(A) The only genetic material in a cancer cell
(B) The bacterial chromosome
(C) The viral chromosome
(D) A transposon
33. The tumor suppressor gene that shows up in approximately half of all U.S. cancers is called
(A) $\quad \mathrm{u} 2$
(B) p 51
(C) 6 pac
(D) p 53
34. Increased blood flow to a cancerous tumor is called
(A) Anaplasia
(B) Metastasis
(C) Malignancy
(D) Vascularization
35. A mutation that removes a base pair and thus offsets the reading frame of the genetic sequence by one letter is called
(A) A basal disruption
(B) A frameshift mutation
(C) A bump-and-run mutation
(D) A sliding sequence mutation
36. Lack of thymus leads to increased infection in humans such a condition called
(A) Thymectomy
(B) Nude mice
(C) DiGeorge's syndrome
(D) None of above
37. The proofreading of newly synthesized DNA, to excise incorrect nucleotides which have been inserted, is done by:
(A) A restriction endonucleases
(B)
DNA
gyrase
(C) DNA ligase
(D) DNA polymerase III
38. The codon is found in :
(A) DNA
(B) rRNA
(C) tRNA
(D) mRNA
39. The size of the E.coli genome is
(A) 4640 bp
(B) 4.64 Kbp
(C) 4.64 Mbp
(D) Not known with certainty
40. An enzyme that recognizes a specific (palindromic) sequence and cuts within a DNA molecule is called $a(n)$ :
(A) Exonuclease
(B) Methylase
(C) Modification enzyme
(D) Restriction endonuclease
41. Which polymerase made widespread use of PCR possible:
(A) DNA polymerase I
(B) Thermus aquaticus (Taq) polymerase
(C) DNA polymerase III
(D) None of the above
42. G-proteins are involved in relying signals through G-protein linked receptors. Which of the following forms of G-proteins considered in to be active state?
(A) G-protein -ADP
(B) G-protein-ATP
(C) G-protein-GDP
(D) G-protein-GTP
43. Which one of the following compounds does not act as second messenger during signaling process?
(A) cAMP
(B) Calcium ions
(C) Ionositol-3,4,5-trisphosphate
(D) Triacylglycerol
44. Which of the following biochemical reactions is most commonly utilized by living cells to propagate intracellular signals?
(A) Acylation
(B) Phosphorylation
(C) Methylation
(D) Decarboxylation
45. Tight junctions:
(A) Are essential for metabolic coupling
(B) Donot occur in vertebrates
(C) Have the closest approach of two plasma membranes of any junction
(D) Surround connexons
46. Which of the following is the component in the signaling pathway stimulated by receptor tyrosine kinases?
(A) Adenylate cyclase
(B) Adaptor proteins
(C) Autophosphorylating receptor
(D) Ras activating protein
47. Which of the following wavelength ranges is associated with UV spectroscopy?
(A) $\quad 0.8-500 \mu \mathrm{~m}$
(B) $400-100 \mathrm{~nm}$
(C) $380-750 \mathrm{~nm}$
(D) $0.01-10 \mathrm{~nm}$
48. Electron microscopes have higher resolution than any type of light microscope because of their:
(A) Higher magnification
(B) Lenses used are of high quality
(C) Very short wavelength of electrons
(D) Bulky structure of microscope
49. Lycopene $\left(\lambda_{\max }=469 \mathrm{~nm}\right)$ is present in tomatoes. What colour of light does lycopene absorb?
(A) Green
(B) Red
(C) Blue
(D) Orange
50. Which of the following is not a limitation of Beer Lambert's law, which gives the relation between absorption, thickness, and concentration?
(A) Concentration must be lower
(B) Radiation must have higher bandwidth
(C) Radiation source must be monochromatic
(D) Does not consider factors other than thickness and concentration that affect absorbance

## (BOTANY)

1) Basal elaterphore present in the capsule of Pellia develops from:
(A) Elater mother cells
(B) Sterile sporogenous cells
(C) Fertile sporogenous cells
(D) Amphithecium
2) Which of the following pair is an example of saprophytic liverwort and moss respectively?
(A) Cryptothallus mirabilis and Buxbaumia aphylla
(B) Buxbamia aphylla and Cryptothallus mirabilis
(C) Cryptothallus and Zoopsis
(D) Zoopsis and Cryptothallus
3) Setae are absent in the sporangium of:
(A) Anthoceros and Corsinia
(B) Corsinia and Tortula
(C) Tortula and Antoceros
(D) Sphagnum and Corsinia
4) Name the earliest land vascular plant discovered from Mid Silurian of Ireland.
(A) Cooksonia
(B) Rhynia
(C) Aglaophyton
(D) Psilophyton
5) Which of the following statement is not true about stele system of pteridophytes?
(A) Primitive type of stele is found in Lycopodium
(B) Stele in which centre is occupied by pith is called siphonostele
(C) A siphonostele perforated by several overlapping leaf gap is called Solenostele
(D) Amphiphloic siphonostele is the condition in which xylem is surrounded on both sides by phloem, pericycle and endodermis
6) In transverse sections of a young stem, if vallecular canals and cranial canals are present, then the plant belongs to:
(A) Lycopodiales (B) Isoetales
(C) Selaginellales (D) Equisetales
7) Which of the following families of the gymnosperms has $\mathbf{1}$ ovule per scale?
(A) Pinaceae
(B) Araucariaceae (C)
Cupressaceae (D) Ginkgoaceae
8) Sago is a starch mostly obtained from pith and cortex of the stem of Cycas species:
(A) Cycas cercinalis
(B) Cycas revoluta
(C) Cycas pectinata
(D) Cycas rumphi
9) A demulcent is a drug that:
(A) Calms the nerves and induces sleep
(B) Enhances appetite and digestion
(C) Soothes skin and mucous membrane
(D) Increases the discharges of urine
10) Which of the following plant extract has property to cure rheumatism?
(A) Azadirachta indica
(B) Cassia augustifolia
(C) Withania somniferum
(D) Aegle marmelos
11) In which of the following types, the embryo sac has 16 nuclei, with a 3-celled egg apparatus and two polar nuclei?
(A) Peperomia
(B) Penaea
(C) Drusa
(D) Fritillaria
12) Amborella trichopoda is now widely considered as:
(A) Oldest known fossils of an angiosperm
(B) Most primitive living angiosperm
(C) Most primitive living vascular plant
(D) Oldest known fossil of a seed plant
13) Which of the following algae have nonflagellate coenobia?
(A) Eudorina
(B) Pandorina
(C) Pediastrum
(D) Volvox
14) Formation of statospores is a characteristic feature of:
(A) Chlorophyta Division
(B) Chrysophyta Division
(C) Phaeophyta Division
(D) Pyrrophyta Division
15) The most important alga used as food in Japan is:
(A) Furcellaria fastiagata
(B) Gleopeltis furcata
(C) Microcystis aeruginosa
(D) Porphyra tenera
16) Which of the following shows secondary growth by successive cambia?
(A) Boerhaavia diffusa
(B) Aristolochia triangularis
(C) Thunbergia coceinea
(D) Serjania corrugate
17) Bicollateral vascular bundles are characteristic feature of which of the following genera:
(A) Cucurbita
(B) Helianthus
(C) Althaea
(D) Salvia
18) The type specimen is collected from the original material to serve as nomenclature type, when holotype is missing:
(A) Isotype
(B) Lectotype
(C) Neotype
(D) Topotype
19) ICN stands for:
(A) International code of nomenclature for plants
(B) International code of nomenclature for algae, fungi and plants
(C) International code of nomenclature for algae and fungi
(D) International code of nomenclature for angiosperm only
20) Which of the following is a bionomial in which genus and species names are identical in spelling?
(A) Autonym
(B) Tautonym
(C) Homonym
(D) Synonym
21) The Genome $D$ present in bread wheat is supposed to have been derived from:
(A) Aegilops Squarrosa
(B) Aegilops Speltoides
(C) Triticum Monococcum
(D) Triticum vulgare
22) The process involving inspection, fumigation and growing the introduced
plant material in isolation is known as
(A) Acclimatization
(B) Quarantine
(C) Adaptation
(D) Tolerance
23) Heterosis is:
(A) Superiority of hybrids over their parents
(B) Induction of mutation
(C) Mixture of two or more traits
(D) Spontaneous mutation
24) A type 1 survivorship curve is characteristic of the species with a rapid increase in mortality in old age. This type of curve is:
(A) Typical of many invertebrates that produce a large number of offspring
(B) Typical of humans and other large mammals
(C) Almost never found in nature
(D) Typical of all species of birds
25) According to McArthur and Wilson's equilibrium theory, which of the following is true?
(A) Larger islands and islands closer to continent are expected to have more species than smaller and isolated islands
(B) Smaller islands and islands far from the continent are expected to have more species than larger and isolated islands
(C) Smaller islands and islands closer to the continent are expected to have more species than far away and isolated islands
(D) More species are expected on all islands irrespective of their size and distance from continent
26) Wetlands are conserved internationally through an effort called:
(A) Basei convention
(B) Rio convention
(C) Montreal convention
(D) Ramsar convention
27) Which of the following fungal group belongs to phylum Straminopila?
(A) Ascomycota
(B) Basidiomycota
(C) Zygomycota
(D) Oomycota
28) Which of the following zoosporic fungi has nuclear cap in its zoospores?
(A) Blastocladiella emersonii
(B) Apodachlya brachynema
(C) Plasmodiophora brassicae
(D) Phytophthora infestans
29) Unicellular stalked teleutospores with papillar thickening is a characteristic feature of the following rust genera?
(A) Uromyces
(B) Puccinia
(C) Phragmidium (D) Ravenelia
30) Phenylalanine, a precursor of most of the phenolics in higher plants is a product of which one of the following pathways?
(A) Shikimic acid pathway
(B) Malonic acid pathway
(C) Mevalonic acid pathway
(D) Methylerythritol pathway
31) Gibberellic acid (GA) controls seed germination by directing breakdown of the stored starch. In which one of the following tissues of the barley seed, $\alpha$-amylase gene is induced in response to GA?
(A) Endosperm
(B) Coleoptile
(C) Aleurone layer (D) Embryo
32) How many ATP and $\mathrm{NADPH}_{2}$ are used in $\mathrm{C}_{3}$ cycle for net production of one molecule of 3-phosphoglyceraldehyde?
(A) 3 ATP and $2 \mathrm{NADPH}_{2}$
(B) 9 ATP and $6 \mathrm{NADPH}_{2}$
(C) 6ATP and $6 \mathrm{NADPH}_{2}$
(D) $\quad 6 \mathrm{ATP}$ and $9 \mathrm{NADPH}_{2}$
33) Magnesium, iron and Molybdenum metals are present in enzymes as activators, which are:
(A) Cytochrome, Peptidases and Phosphotases respectively
(B) Phosphatases, Cytochromes and Nitrogenase respectively
(C) Fructokinase, Cytochromes and Nitrate reductase respectively
(D) Dehydrogenase, Kinase and Nitrate reductase respectively
34) If the free energy change ( $\Delta \mathrm{G}$ ) in a reaction is a negative value, it indicates that the:
(A) Reaction releases energy
(B) Reaction absorbs energy
(C) Reaction is in positive direction
(D) Reaction is in negative direction
35) Enzyme acts as biological catalyst by increasing rate of reaction by:
(A) Increasing activation energy
(B) Decreasing activation energy
(C) Increasing free energy change
(D) Increasing entropy
36) During protein synthesis in prokaryotes, the peptidyl transferase activity required for pepetide bond formation is due to:
(A) Ribosomal proteins
(B) 16 S ribosomal RNA
(C) 23S ribosomal RNA
(D) Aminoacyl t-RNA
37) Major function of $\sigma$-Subunit of $E$. coli RNA polymerase during transcription is to:
(A) Initiate transcription and does not fall off during elongation
(B) Decreases affinity of core enzyme to promoter
(C) Binds to DNA independent of core enzyme
(D) Ensure recognition of promoter region by interacting with core enzyme
38) During replication, the RNA primer is degraded by the $5^{\prime}-3$ ' exonuclease activity of:
(A) RNase H1 (ribonuclease H1)
(B) FEN-1 (flap endonuclease 1)
(C) TopoisomeraseII B
(D) DNA polymerase $\varepsilon$
39) Species richness is much pronounced in which type of forests:
(A) Tropical rain forest
(B) Tropical deciduous forest
(C) Temperate forest
(D) Alpine forest
40) Average annual precipitation and temperature of temperate forest are respectively:
(A) 255 cm precipitation and $25^{\circ} \mathrm{C}$ temperature
(B) 300 cm precipitation and $15^{\circ} \mathrm{C}$ temperature
(C) $\quad 100 \mathrm{~cm}$ precipitation and $15^{\circ} \mathrm{C}$ temperature
(D) 300 cm precipitation and $25^{\circ} \mathrm{C}$ temperature
41) Subalpine forest and tropical thorn forest are dominated with following plant species:
(A) Dichanthium and Abies
(B) Ilex and Pinus
(C) Abies and Acacia
(D) Dipterocarpus and Acacia
42) The apple scab disease is caused by:
(A) Xanthomonas citri
(B) Venturia inaequalis
(C) Colletotrichum falcatum
(D) Curvularia prasadii
43) The 'Tundu' disease of wheat is caused by:
(A) Anguina triticia and Corynebacterium tritici
(B) Anguina tritici
(C) Anguina tritici and Erwinia dissolvens
(D) Globodera tritici
44) The "Great Bengal Famine" in 1943 was the result of which of the following:
(A) Bunt of wheat caused by Telletia
(B) Helmithosporium blight of rice
(C) Blast of rice caused by Pyricularia
(D) Late blight of potato caused by Phytophthora
45) Standard deviation of a sample is 240 and number of individuals of the sample are 64. Find out the Standard error of mean (SEM)?
(A) 30
(B) 50
(C) 20
(D) 60
46) Choose the correct relation among arithmetic mean (AM), geometric mean (GM) and harmonic mean (HM):
(A) $\mathrm{GM}>\mathrm{AM}>\mathrm{HM}$
(B) $\quad \mathrm{HM}>\mathrm{AM}<\mathrm{GM}$
(C) $\mathrm{AM}>\mathrm{GM}>\mathrm{HM}$
(D) $\quad \mathrm{HM}>\mathrm{AM}>\mathrm{GM}$
47) Which of the following is a non parametric test?
(A) Chi square test
(B) F-test
(C) T-test
(D) ANOVA test
48) The genetically modified tomato 'Flavr Savr' with delayed ripening was the result of:
(A) Over expression of Antisense gene for polygalacturonase
(B) Gene silencing by antisense RNA
(C) Over expression of gene for ethylene biosynthesis
(D) Over expression of gene for ACC synthase
49) Glyphosate-based herbicides, such as Roundup, target which pathway of the plants?
(A) Mevalonic acid pathway
(B) Shikimate pathway enzyme 5-enolpyruvylshikimate 3-phosphate (EPSP) synthase
(C) Methyl erythritol pathway
(D) Malonic acid pathway
50) Which of the following genetic engineering method is best suited for addition of gene into plants?
(A) Plasmid method
(C) Biolistic (gene gun) method
(B) Vector method
(D) Microinjection

## (CHEMISTRY)

1. The variable that is kept constant in an isobaric process is
(A) Volume
(B) Temperature
(C) Molarity
(D) Pressure
2. The following isotope is radioactive
(A) ${ }^{3} \mathrm{H}$
(B) ${ }^{12} \mathrm{C}$
(C) $\quad{ }^{2} \mathrm{H}$
(D) ${ }^{16} \mathrm{O}$
3. The drag on the central ion as discussed by Debye Huckel is known as
(A) Viscous effect
(B) Electrophoretic effect
(C) Asymmetry effect
(D) Wien effect
4. The well known radioisotope ${ }^{14} \mathrm{C}$ decays by emitting the following particle
(A) Alpha
(B) Negatron
(C) Positron
(D) Neutron
5. The volume occupied in Litres by 18.0 g of Oxygen gas made up of Oxygen atoms having 10 neutrons at normal temperature and pressure is
(A) 5.6
(B)
11.2
(C) 2.24
(D) 22.4
6. An experimenter tries to melt ice at 265 K as well as at 300 K . The sign of $\Delta \mathrm{G}$ for the melting of ice experiment at 265 K and 300 K respectively can be expressed as follows:
(A) +-
(B) ++
(C) - +
(D) --
7. It was observed that in a chemical reaction $A$ proceeding to $B$ the increase in the concentration of $A$ by 3 times increases the reaction rate by 9 times the order of the reaction would be
(A) 1
(B) 0
(C) 2
(D) 3
8. In Raman Spectroscopy the scattered radiation having energy more than the incident radiation is called
(A) Stokes
(B) Anti-Stokes
(C) Rayleigh
(D) Thermal
9. Given that $g_{N}=5.585$ and $\mu_{N}=5.05 \times 10^{-27} \mathrm{~J} \mathrm{~T}^{-1}$ the NMR frequency of a proton in a magnetic field of intensity 2.82 Tesla would in $\mathbf{M H z}$ be
(A) 60
(B) 120
(C) 90
(D) 30
10. Debye Falkenhegen Effect describes the conductance of an electrolyte of interest under the following condition
(A) High AC frequency
(B) High Voltage
(C) Low Concentration
(D) High Viscosity
11. The de Broglie wavelength of an electron moving with a velocity of $4.8 \times 10^{5} \mathrm{~m} \mathrm{sec}^{-1}$ is about
(A) $1.5 \times 10^{-9} \mathrm{~m}$
(B) $3.0 \times 10^{-10} \mathrm{~m}$
(C) $4.8 \times 10^{-5} \mathrm{~m}$
(D) $4.8 \times 10^{5} \mathrm{~m}$
12. The pair ${ }^{13} \mathrm{C}$ and ${ }^{13} \mathrm{~N}$ is an example of
(A) Isobar
(B) Isotone
(C) Isomer
(D) Isotope
13. The number of elements depicted in the most recent periodic table is
(A) 103
(B) 92
(C) 109
(D) 118
14. The unit cell parameter $a=b=c$ and $\alpha=\beta=\gamma=90^{\circ}$ represents crystal system
(A) Tetragonal
(B) Cubic
(C) Monoclinic
(D) Hexagonal
15. The lowest bond energy is depicted by the bond between following pair of atoms
(A) $\mathrm{H}-\mathrm{F}$
(B) $\mathrm{H}-\mathrm{Br}$
(C) $\mathrm{H}-\mathrm{H}$
(D) $\quad \mathrm{C}=\mathrm{C}$
16. If the number of ways of a molecular arrangement can be expressed by term $X$ then the relationship between entropy $S$ and the term $X$ can be represented by
(A) $\quad \mathrm{S}=\mathrm{X}$
(B) $\mathrm{S}=\mathrm{kX}$
(C) $\quad \mathrm{S}=\mathrm{Q} / \mathrm{T}$
(D) $\quad \mathrm{S}=\mathrm{k} \ln \mathrm{X}$
17. The neutron to proton ratio in the isotope of the element Boron having mass number 13 is
(A) 1.6
(B) 1.0
(C) 13
(D) 5
18. A compound $X$ on heating gives a colourless gas. The residue is dissolved in water to obtain $Y$. Excess $\mathrm{CO}_{2}$ is bubbled through aqueous solution of $Y$ which result in the formation of $\mathbf{Z} . \mathbf{Z}$ on gentle heating gives back $X$. The compound $X$ is:
(A) $\mathrm{CaCO}_{3}$
(B) $\quad \mathrm{Na} 2 \mathrm{CO}_{3}$
(C) $\mathrm{Ca}\left(\mathrm{HCO}_{3}\right)_{2}$
(D) $\quad \mathrm{K}_{2} \mathrm{CO}_{3}$
19. When orthoboric acid $\left(\mathrm{H}_{3} \mathrm{BO}_{3}\right)$ is heated, the residue left is:
(A) Boron
(B) Metaboric acid
(C) Boric anhydride
(D) Borax
20. Which of the following trihalides of nitrogen is least basic:
(A) $\quad \mathrm{NF}_{3}$
(B) $\quad \mathrm{NCl}_{3}$
(C) $\quad \mathrm{NBr}_{3}$
(D) $\quad \mathrm{NI}_{3}$
21. The number of $\mathbf{P}-\mathrm{O}-\mathrm{P}$ and $\mathrm{P}-\mathrm{O}-\mathrm{H}$ bonds present respectively in pyrophosphoric acid molecule are:
(A) 1,2
(B) 2,2
(C) 1,4
(D) 1,8
22. Which of the following does not form clatherates:
(A) Helium
(B) Argon
(C) Krypton
(D) Xenon
23. The aqueous solution containing which one of the following ions will be colourless:
(A) $\mathrm{Mn}^{2+}$
(B) $\mathrm{Fe}^{2+}$
(C) $\mathrm{Ti}^{3+}$
(D) $\mathrm{Sc}^{3+}$
24. Which of the following belongs to $\mathrm{C}_{3 \mathrm{v}}$ point group:
(A) $\quad \mathrm{SO}_{3}$
(B) $\quad \mathrm{BBr}_{3}$
(C) $\mathrm{NH}_{3}$
(D) $\quad \mathrm{AlCl}_{3}$
25. Which structures for $\mathrm{XeO}_{3}$ and $\mathrm{XeF}_{4}$ are consistent with the VSEPR model?
(A) Trigonal pyramidal, Square planar
(B) Trigonal planar, Square planar
(C) Trigonal pyramidal, Tetrahedral
(D) Trigonal planar, Tetrahedral
26. The compound which exhibits Jahn-Teller distortion is:
(A) $\left[\mathrm{Mn}\left(\mathrm{H}_{2} \mathrm{O}\right)_{6}\right]^{2+}$
(B) $\quad\left[\mathrm{Mn}\left(\mathrm{H}_{2} \mathrm{O}\right)_{6}\right]^{3+}$
(C) $\left[\mathrm{Cr}\left(\mathrm{H}_{2} \mathrm{O}\right)_{6}\right]^{3+}$
(D) $\left[\mathrm{Fe}(\mathrm{CN})_{6}\right]^{4-}$
27. The total no. of isomers in $\mathrm{Co}(\mathrm{en})_{2} \mathrm{Cl}_{2}$ is
(A) 4
(B) 3
(C) 6
(D) 5
28. The no. of lines that appear in the EPR spectra of $\left[\mathrm{C}_{6} \mathrm{H}_{6}\right]^{-}$is:
(A) 5
(B) 7
(C) 11
(D) 13
29. The hardness of water is measured by:
(A) Distillation
(B) Conductivity
(C) EDTA method
(D) Sublimation
30. Cyanide process is used to obtain:
(A) Cr
(B) Ag
(C) Cu
(D) Zn
31. Which of the following has largest bond angle:
(A) $\mathrm{NH}_{3}$
(B) $\mathrm{PH}_{3}$
(C) $\mathrm{AsH}_{3}$
(D) $\mathrm{SbH}_{3}$
32. Peroxo linkage is present in:
(A) $\quad \mathrm{H}_{2} \mathrm{SO}_{3}$
(B) $\mathrm{H}_{2} \mathrm{SO}_{5}$
(C) $\quad \mathrm{H}_{2} \mathrm{SO}_{4}$
(D) $\quad \mathrm{H}_{2} \mathrm{~S}_{2} \mathrm{O}_{7}$
33. WOC complex present in photosynthesis process contain
(A) Zn
(B) Fe
(C) Pu
(D) Mn
34. Assign $R$ / $S$ configuration at C-1, C-2 and C-5 in the following compounds.

(A) $1 R, 2 S, 4 R$
(B)
1R,2R,4R
(C) $1 \mathrm{~S}, 2 \mathrm{~S}, 4 \mathrm{R}$
(D) $\quad 1 \mathrm{~S}, 2 \mathrm{R}, 4 \mathrm{R}$
35. Mark the relationship ship between following structures $X$ and $Y$ :

(A) Enantiomers
(B)
Diastereomers (C) Meso
(D) Same
36. Which reactive intermediate is involved in the following reaction?

(A) Carbene
(B) Nitrene
(C) Benzyne
(D) Free radical
37. Reaction of 2-buten-1-ol with thionyl chloride results in formation of:
(A) 3-Chloro-1-butene
(B) 1-Chloro-2-butene
(C) 1,2-Dichlorobutane
(D) 1,3-Dichlorobutane
38. Solvolysis of neopentyl bromide result in formation of:
(A) 2-Methyl-2-butene
(B) 3-Methyl-1-butene
(C) 2-Butene
(D) 1-Butene
39. Nitration of $\mathbf{N}, \mathbf{N}$-dimethylaniline results in formation of:
(A) $\quad m$-Nitro-N,N-dimethylaniline
(B) $\quad o$-Nitro-N,N-dimethylaniline
(C) $\quad p$-Nitro-N,N-dimethylaniline
(D) N -Nitro-N,N-dimethylaniline
40. In sulphonation of benzene attacking species is:
(A) $\quad \mathrm{SO}_{2}$
(B) SO
(C) $\mathrm{SO}_{3}$
(D) $\mathrm{S}_{2} \mathrm{O}$
41. Addition elimination mechanism of aromatic nucleophilic substitution involves the intermediary of:
(A) Carbocation
(B) Carbanion
(C) Carbene
(D) Benzyne
42. Predict the product $X$ in the following reaction:

(A) 2-Phenyl ethyl amine
(B) 1-Phenyl ethyl amine
(C) Benzyl amine
(D) Ethyl benzene
43. The best reagent to achieve the following transformation is:

(A) $\quad \mathrm{BH}_{3} . \mathrm{THF}$
(B) $\mathrm{LiAlH}_{4}$
(C) $\mathrm{NaBH}_{4}$
(D) $\mathrm{NaBH}_{3} \mathrm{CN}$
44. Selective reduction of ketone in presence of aldehyde can be achieved by use of:
(A) $\mathrm{NaBH}_{4} / \mathrm{CH}_{3} \mathrm{OH}$
(B) $\mathrm{NaBH}_{3} \mathrm{CN} / \mathrm{CH}_{3} \mathrm{OH}$
(C) $\quad$ II. $\left.\mathrm{NaBH}_{4} / \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CeCl}_{3}\right]$
[II. $\mathrm{H}_{2} \mathrm{O} / \mathrm{H}^{+}$]
(D) $\quad \mathrm{NaBH}\left(\mathrm{OOCCH}_{3}\right)_{2}$
45. The best reagent for trans hydroxylation of alkenes is:
(A) Jones reagent
(B) Sarett reagent
(C) Collins reagent
(D) Prevost reagent
46. Predict the product $X$ in the following reaction:

(A) Benzoic acid
(B) Benzophenone
(C) Diphenyl amine
(D) Benzaldehyde
47. Oxidation of styrene to phenyl acetaldehyde can be achieved by use of:
(A) alk. $\mathrm{KMnO}_{4}$
(B) $\mathrm{MnO}_{2}$
(C) $\quad \mathrm{Tl}\left(\mathrm{NO}_{3}\right)_{3} /$ dil. $\mathrm{HNO}_{3}$
(D) $\quad \mathrm{HIO}_{4}$
48. Which intermediate is involved in Favorskii rearrangement?
(A) Cypclopropane
(B) Cyclopropene
(C) Cyclopropanol
(D) Cyclopropanone
49. Predict the product of the following reaction:

(A)

(B)

(C)

(D)

50. Acid catalysed hydration of propyne results in formation of:
(A) Propane oxide
(B) Propinoaldehyde
(C) Acetone
(D) Propane-1,2-diol

## (PHYSICS)

1) Let $u(x, y)=x+\frac{1}{2}\left(x^{2}-y^{2}\right)$ be the real part of analytic function $f(z)$ of the complex variable $Z=x+i y$, the Imaginary part of $f(z)$ is
(A)
$y+x y(\mathrm{~B})$
$x y$
(C) $y$
(D) $y^{2}-x^{2}$
2) If the fourth colour band of a given resistor is missing, what is the tolerance of a resistor
(A) $10 \%$
(B) $20 \%$
(C) $30 \%$
(D) $40 \%$
3) The value of the integral $I=\int_{0}^{\infty} \frac{\sin x}{x} d x$ is
(A) 0
(B) $\pi$
(C) $\frac{\pi}{2}$
(D) $2 \pi i$
4) In which of the following detector $p-n$ junction diode is used
(A) Surface barrier detector
(B) GM Counter
(C) Scintillation Counter
(D) Proportional Counter
5) Consider an anti-symmetric tensor $P_{i j}$ with indices $\boldsymbol{i}$ and $\boldsymbol{j}$ running from 1 to 5 . The number of independent components of the tensor is
(A) 3
(B) 10
(C) 9
(D) 6
6) The source of emission of electrons in a CRT is
(A) $\quad p-n$ junction diode
(B) Accelerating anode
(C) A barium and strontium oxide coated cathode
(D) Post-accelerating anode
7) Consider the linear differential equation $\frac{d y}{d x}=x y$, if $y=2$ at $x=0$, then the value of $y$ at $x=2$ is given by
(A) $\quad e^{-2}$
(B) $2 e^{-2}$
(C) $e^{2}$
(D) $2 e^{2}$
8) At the input an amplifier has a signal voltage level of $3 \mu V$ and noise voltage level of $1 \mu V$. If the voltage gain of the amplifier is 20 , the ratio of signal to noise $\left(\frac{S}{N}\right)$ at the out -put is
(A) 4
(B) 9
(C) 8
(D) 6
9) Laplac transform of $\left\{e^{-2 t}-e^{-3 t}\right\}$ is
(A) $\frac{1}{s^{2}+3 s+6}$ (B)
$\frac{1}{s^{2}+5 s+6}(\mathrm{C}) \quad \frac{1}{s-2}$
(D) $\frac{1}{s+2}$
10) The lifetime of an atomic state is 1 nanosecond. The natural line width of the spectral line in the emission spectrum of this state is of the order of
(A) $10^{-10} \mathrm{eV}$
(B) $10^{-9} \mathrm{eV}$
(C) $10^{-6} \mathrm{eV}$
(D) $\quad 10^{-4} \mathrm{eV}$
11) A Ge semiconductor is doped with acceptor impurity concentration of $10^{15} \mathrm{atoms} / \mathrm{cm}^{3}$. For the given hole mobility of $1800 \mathrm{~cm}^{2} / \mathrm{V} . \mathrm{s}$, The resistivity of the material is
(A) $0.288 \Omega \mathrm{~cm}$
(B) $0.694 \Omega \mathrm{~cm}$
(C) $3.472 \Omega \mathrm{~cm}$
(D) $6.944 \Omega \mathrm{~cm}$
12) A gate with only one input and one output is
(A) an OR gate
(B)
a NOT gate
(C) a NAND gate
(D) an AND gate
13) In a microprocessor, the resistor which holds the address of the next instruction to be fetched is
(A) Accumulator
(B) Program counter
(C) Stak counter
(D) Instructor register
14) The voltage resolution of a $\mathbf{1 2}$ bit digital to analog converter (DAC) whose out-put varies from -10 V to $+\mathbf{1 0} \mathrm{V}$, is approximately
(A) 1 mV
(B) 5 mV
(C) 20 mV
(D) 100 mV
15) If the memory chip size is $256 \times 1$ bits, what is the number of chips required to make up 1 Kbyte of memory?
(A) 32 chips
(B) 64 chips
(C) 128 chips
(D) 256 chips
16) The energy of the first excited quantum state of a particle in the two dimensional potential $V(x, y)=\frac{1}{2} m \omega^{2}\left(x^{2}+4 y^{2}\right)$ is
(A)
$2 \omega$
(B)
$3 \omega$
(C) $\frac{3}{2} \omega$
(D) $\frac{5}{2} \omega$
17) The quantum mechanical operator for the momentum of a particle moving in one dimension is given by
(A)
(B) $-i \frac{d}{d x}$
(C) $i \frac{d}{d t}$
(D) $\quad-\frac{2}{2 m} \frac{d^{2}}{d x^{2}}$
18) The energy eigen values of a particle in the potential $V(x)=\frac{1}{2} m \omega^{2} x^{2}-a x$ are
(A)
$E_{n}=\left(n+\frac{1}{2}\right) \omega-\frac{a^{2}}{2 m \omega^{2}}$
(B) $\quad E_{n}=\left(n+\frac{1}{2}\right) \omega+\frac{a^{2}}{2 m \omega^{2}}$
(C)
$E_{n}=\left(n+\frac{1}{2}\right)$
$\omega-\frac{a^{2}}{m \omega^{2}}$
(D) $\quad E_{n}=\left(n+\frac{1}{2}\right) \omega$
19) If $\boldsymbol{f}(\boldsymbol{\theta}, \varphi)$ is the scattering amplitude, the differential cross section $\boldsymbol{\sigma}(\boldsymbol{\theta}, \varphi)$ will be equal to
(A)
$|f(\theta, \varphi)|$
(B) $\frac{|f(\theta, \varphi)|^{2}}{2}$
(C) $\quad|f(\theta, \varphi)|^{2}$
(D) $\quad f(\theta, \varphi)$
20) Value of $\overrightarrow{\boldsymbol{s}_{\mathbf{1}}} \cdot \overrightarrow{\boldsymbol{s}_{\mathbf{2}}}$ for two spin half particle is
(A) $\frac{1}{4}$ for singlet
(B) $\quad-\frac{3}{4}$ for triplet
(C) $\quad \frac{1}{4}$ for triplet
(D) $\quad-\frac{3}{2}$ for singlet
21) The acceleration due to gravity ( $g$ ) on the surface of earth is approximately 2.6 times that on the surface of Mars. Given that the radium of Mars is about one half the radius of Earth, the ratio of the escape velocity on Earth to that on Mars is approximately
(A) 1.1
(B) 1.3
(C) 2.3
(D) 5.2
22) If the Lagrangian of a particle moving in one dimension is given by $L=\frac{\dot{x}^{2}}{2 x}-V(x)$, then the Hamiltonian is
(A) $\frac{1}{2} x p^{2}+V(x)$
(B) $\frac{\dot{x}^{2}}{2 x}+V(x)$
(C) $\frac{\dot{x}^{2}}{2}+V(x)$
(D) $\frac{p^{2}}{2 x}+V(x)$
23) A particle is moving under the action of a generalized potential $V(q, \dot{q})=\frac{1+\dot{q}}{q^{2}}$. The magnitude of the generalized force is
(A) $\frac{2(1+\dot{q})}{q^{3}}$
(B) $\frac{2(1-\dot{q})}{q^{3}}$
(C) $\frac{2}{q^{3}}$
(D) $\frac{\dot{q}}{q^{3}}$
24) The speed of a particle whose kinetic energy is equal to its rest mass energy is given by (c is the speed of light in vacuum)
(A) $\frac{c}{3}$
(B) $\quad \frac{\sqrt{ } 2}{3} c$
(C) $\frac{c}{2}$
(D) $\frac{\sqrt{3}}{2} c$
25) A particle is constrained to move along the inner surface of a hemisphere. The number of degrees if freedom of the particle is
(A) One
(B) Two
(C) Three
(D) Four
26) Four equal point charges are kept fixed at the four vertices of a square. How many neutral points (i.e points where the electric field vanishes) will be found inside the surface
(A) 1
(B) 4
(C) 0
(D) 7
27) The magnetic field corresponding to the vector potential $\vec{A}=\frac{1}{2} \vec{F} \times \vec{r}+\frac{10}{r^{3}} \vec{r}$ (where $\vec{F}$ is a constant vector) is
(A) $\vec{F}$
(B) $\quad-\vec{F}$
(C) $\vec{F}+\frac{30}{r^{4}} \vec{r}$
(D) $\vec{F}-\frac{30}{r^{4}} \vec{r}$
28) A plane electromagnetic wave travelling in free space is incident normally on a glass plate of refractive index $\frac{3}{2}$. If there is no absorption by the glass, its reflectivity is
(A) $04 \%$
(B) $16 \%$
(C) $20 \%$
(D) $50 \%$
29) The field of magnetic vector $\vec{B}$ is always
(A) Solenoidal
(B) Irrotational
(C) Non-solenoidal
(D) Sometime Irrotational
30) An infinitely long closely wound solenoid carries a sinusoidally varying current. The induced electric field is
(A) Zero inside and non-zero outside the solenoid
(B) Zero everywhere
(C) Non-zero inside and zero outside solenoid
(D) Non-zero inside as well as outside the solenoid
31) A metal with body centred cubic (bcc) show the first (i.e smallest angle) diffraction peak at a Bragg angle of $\theta=30^{0}$. The wavelength of X-ray used is $2.1 A^{0}$. The volume of the Primitive unit cell of the metal is
(A)
$26.2\left(A^{0}\right)^{3}$
(B)
$13.1\left(A^{0}\right)^{3}$
(C) $\quad 9.3\left(A^{0}\right)^{3}$
(D) $\quad 4.6\left(A^{0}\right)^{3}$
32) In the Debye model for a three dimensional crystal the internal energy $U$ at low temperature is represented by
(A)
$U \alpha T$
(B) $\quad U \alpha T^{2}$ (C)
$U \alpha T^{3}$ (D)
$U \propto T^{4}$
33) Sodium atoms crystallic in BCC metal. The atomic radius of sodium is $1.86 A^{0}$. The Fermi energy of sodium at $0 K$ is
(A)
5.11 eV
(B) $\quad 6.01 \mathrm{eV}(\mathrm{C})$
3.11 eV (D)
4.21 eV
34) For an ideal Fermi gas in three dimensions, the electron velocity $V_{F}$ at the Fermi surface is related to electron concentration ' $n$ ' as
(A) $\quad V_{F} \propto n^{\frac{2}{3}}$
(B) $\quad V_{F} \alpha n(\mathrm{C})$
$V_{F} \alpha n^{\frac{1}{2}}$
(D) $\quad V_{F} \alpha n^{\frac{1}{3}}$
35) Given that the ground state energy of the hydrogen atom is $\mathbf{- 1 3 . 6} \mathbf{e V}$. The ground state energy of the positronium (which is a bound state of an electron and positron) is
(A) $\quad+6.8 \mathrm{eV}$ (B)
-6.8 Ev
(C) $\quad-13.6 \mathrm{Ev}(\mathrm{D})$
$-27.2 \mathrm{eV}$
36) An electron collides with a hydrogen atom in its ground state and excites it to a state of $\boldsymbol{n}=$ 3. How much energy was given to the hydrogen atom in this inelastic collision?
(A)
13.6 eV (B)
6.8 eV
(C) $\quad 12.1 \mathrm{eV}$ (D)
1.51 eV
37) In deep inelastic scattering electrons are scattered off protons to determine if a proton has any internal structure. The energy of the electron for this must be at least
(A) $1.25 \times 10^{9} \mathrm{eV}$
(B) $1.25 \times 10^{12} \mathrm{eV}$
(C) $1.25 \times 10^{6} \mathrm{eV}$
(D) $1.25 \times 10^{8} \mathrm{eV}$
38) The reaction ${ }_{1}^{2} D+{ }_{1}^{2} D \rightarrow{ }_{2}^{4} \mathrm{He}+\pi^{0}$ cannot proceed via strong interaction, because it violates the conservation of
(A) Angular momentum
(B) Electric charge
(C) Baryon no
(D) Isospin
39) The reaction $e^{+}+e^{-} \rightarrow \mu^{+}+\pi^{-}$is forbidden because of
(A) Law of baryon number conservation
(B) Law of momentum energy conservation
(C) Law of muon number conservation
(D) Law of energy conservation
40) If the mass of a particle is three times of its rest mass, then the speed at which the particle is moving is (where cis the velocity of light)
(A) $\frac{\sqrt{ } 8}{3} c$
(B) $\frac{\sqrt{ } 8}{9} c$
(C) $\frac{8}{9} c$
(D) $3 c$
41) What is the approximate Fermi Kinetic energy of the nucleons (either proton or neutron) in case of a self-conjugate nuclei with $\mathrm{N}=\mathrm{Z}=\mathrm{A} / \mathbf{2}$ ?
(A)
33 MeV (B)
150 MeV
(C) $\quad 0.5 \mathrm{MeV}$
(D) 200 MeV
42) According to the shell model, the ground state spin of the ${ }_{6}^{13} C$ nucleus is
(A) $\frac{1}{2}$
(B) $\frac{3}{2}$
(C) $\frac{5}{2}$
(D) $\frac{7}{2}$
43) Which of the following statement is not correct for a compound nuclear reaction.
(A) Compound nuclear reaction is symmetric around 90 degree
(B) Mass distribution obtained in case of compound nuclear reaction is symmetric
(C) The reaction products peak around the grazing angle
(D) Light ion induced compound nuclear reaction follow Bohr's independence hypothesis
44) The neutron and proton form a deuteron bound state which is stable, while there is no bound state for two neutrons because
(A) Nuclear forces are saturated
(B) Nuclear forces are spin dependent
(C) Nuclear forces are charge dependent
(D) Nuclear forces depend upon magnetic moment
45) Let $E_{s}$ denotes the contribution of the surface energy per nucleon in the Liquid drop model. The ratio $E_{S}\left({ }_{13}^{27} A l\right): E_{s}\left({ }_{30}^{64} Z n\right)$ is
(A) $2: 3$
(B) $4: 3$
(C) $5: 3$
(D) $3: 1$
46) A system of $N$ non-interacting classical point particles is constrained to move on the two dimensional surface of a sphere. The internal energy of the system is
(A) $\frac{3}{2} N K_{B} T$
(B) $\frac{1}{2} N K_{B} T$
(C) $\quad N K_{B} T$
(D) $\frac{5}{2} N K_{B} T$
47) Consider a system of $\mathbf{3}$ fermions which can occupy any of the four available energy states with equal probability. The entropy of the system is
(A) $\quad K_{B} \ln 2$
(B) $2 K_{B} \ln 2$
(C) $2 K_{B} \ln 4$
(D) $3 K_{B} \ln 4$
48) Curie temperature of Iron is that temperature below which it is
(A) Ferromagnetic
(B) Electrically conducting
(C) Superconducting
(D) Radioactive
49) The number of ways in which $\mathbf{N}$ identical bosons can be distributed in two energy level is
(A) $\mathrm{N}+1$
(B) $\frac{N(N-1)^{2}}{2}$
(C) $\frac{N(N+1)}{2}$
(D) N
50) Which one of the following is a first order phase transition
(A) Vaporization of a liquid at its boiling point
(B) Ferromagnetic to paramagnetic
(C) Normal liquid He to super-fluid He
(D) Superconducting to normal state

## (GEOLOGY)

1. As per the Rule of V's the outcropping strata that intersects the contours in general will show a $V$ pointing upstream when the $\qquad$
(A) Strata dips upstream
(B) Strata is horizontal
(C) Strata is Vertical
(D) Strata dips downstream
2. One limb of a fold dips $45^{0}$ due $\mathrm{N} 45^{\circ} \mathrm{E}$ and the other dips $45^{\circ}$ due $\mathrm{N} 45^{\circ} \mathrm{W}$. What would be the approximate direction of plunge of the fold?
(A) East
(B) West
(C) South
(D) North
3. .is a strike slip fault with a vertical fault surface
(A) Pivotal Fault
(B) Detachment Fault
(C) Wrench Fault
(D) Gravity Fault
4. The absence of hydrous minerals and presence of pyroxenes characterises.
(A) Greenschist facies
(B) Blueshist facies
(C) Amphibolite facies
(D) Granulite facies
5. Bouguer anomaly (mgal) along oceanic ridges ranges between $\qquad$
(A) -20 and -30
(B) -200 and -300
(C) +200 and +250
(D) -10 and -50
6. Diamond bearing kimberlites are found in the tectonic setting of $\qquad$
(A) Continental rift
(B) Foreland Basin
(C) Oceanic ridges
(D) Cratons and Passive margins
7. Sensitive High Resolution Ion Micropobe (SHRIMP) is used for
(A) U-Pb Dating
(B) REE Geochemistry
(C) Isotope Geochemistry
(D) Fission Track Dating
8. Which among the following contains Thorium?
(A) Glauconite
(B) Perthite
(C) Monazite
(D) Coffinite
9. .is an ultramafic intrusive rock similar to peridotite
(A) Eurite
(B) Picrite
(C) Troctolite
(D) Allivalite
10. .is a melanocritic picritic basalt.
(A) Oceanite
(B) Saxonite
(C) Lherizolite
(D) Izolite
11. The stratigraphic Law of faunal succession was developed by. $\qquad$
(A) Niels Steno
(B) William Smith
(C) James Hutton
(D) Charles Lyell
12. Biotite is
(A) Uniaxial positive
(B) Biaxial positive
(C) Uniaxial negative
(D) Biaxial negative
13. Olivine shows
(A) $\quad 1^{\text {st }}$ order interference colours
(B) $\quad 2^{\text {nd }}$ order interference colours
(C) $3^{\text {rd }}$ order interference colours
(D) $4^{\text {th }}$ order interference colours
14. Which among the following contains Arsenic?
(A) Chalcopyrite
(B) Chalcocite
(C) Realgar
(D) Covelite
15. Khondalite rock contains which of the following assemblage
(A) Quartz-Orthoclase-Hypersthene
(B) Plagioclase-Orthoclase-Hypersthene
(C) Quartz-Garnet-Rhodonite
(D) Quartz-Haematite-Hypersthene
16. Which one is a metamorphic texture
(A) Lepidoblastic (B) Graphic
(C) Clastic
(D) Aphinitic
17. Which among the following orders survived the K-T boundary?
(A) Ceratitids
(B) Nautilids
(C) Goniatitids
(D) Ammonitids
18. Which one of the following belongs to the oldest trilobite order Redlichiida.
(A) Asaphus
(B) Olenellus
(C) Agnostus
(D) Phacops
19. .is a air breathing freshwater gastropod.
(A) Turbo
(B) Natica
(C) Physa
(D) Crepidula
20. .marks the beginning of Cambrian
(A) Agnostus
(B) Asaphus
(C) Treptichnus
(D) Olenus
21. is a cosmogenic isotope
(A) ${ }^{18} \mathrm{O}$
(B) ${ }^{26} \mathrm{Al}$
(C) ${ }^{12} \mathrm{C}$
(D) ${ }^{13} \mathrm{C}$
22. Earths obliquity cycle (axial tilt) occurs every
(A) 10,000 years
(B) 21,000 years
(C) 41,000 years
(D) 100,000 years
23. $\qquad$ .is considered one the first chordates.
(A) Lingula
(B) Pikaia
(C) Redlichia
(D) Terebratula
24. is a biostratigraphic zone that deals with high abundance of a taxa
(A) Lineage Zone
(B) Concurrent Zone
(C) Acme Zone
(D) Assemblage Zone
25. A marked lack of coal deposits all over the world is related to.....
(A) End Ordovician Extinctions
(B) End Devonian Extinctions
(C) Permian-Triassic Extinctions
(D) Cretaceous-Tertiary Extinctions
26. The Paleocene-Eocene Thermal Maximum (PETM) occurred around
(A) 35 Ma
(B) 45 Ma
(C) 55 Ma
(D) 65 Ma
27. Ediacara Biota found in India occur in the. $\qquad$
(A) Aravalis
(B) Dharwas
(C) Gondwanas
(D) Vindhyans
28. Age of the Deccan Traps is $\qquad$
(A) $\quad 100 \mathrm{Ma}$
(B) 66 Ma
(C) 45 Ma
(D) 150 Ma
29. As per Conodont Alteration Index the colour dark grey denotes temperature between
(A) $\quad 50-80^{\circ} \mathrm{C}$
(B) $\quad 60-140^{\circ} \mathrm{C}$
(C) $\quad 110-300^{0} \mathrm{C}$
(D) $\quad 190-300^{\circ} \mathrm{C}$
30. Spectral logging comes under
(A) Density Logging
(B) Gamma Ray Logging
(C) Resistivity Logging
(D) Nuclear Magnetic Resonance Logging
31. In a typical Bouma Sequence, the Bouma $D$ layer is deposited under.
(A) Very slight turbidity current
(B) No turbidity current
(C) High energy turbidity current
(D) Moderate energy turbidity current
32. The oldest oceanic crust found on the ocean floor today is around
(A) 100 Ma
(B) 450 Ma
(C) 180 Ma
(D) 55 Ma
33. The Galapagos Tripple Junction is an example of $\qquad$ .tripple junction
(A) Fault-Fault- Ridge
(B) Fault-Fault-Trench
(C) Ridge-Ridge-Ridge
(D) Ridge-Trench-Fault
34. Kerguelen Hotspot is the source of.
(A) Deccan Traps
(B) Panjal Traps
(C) Malani Igneous Suite
(D) Rajmahal Traps
35. The hyper-impact Lonar Crater is formed in.
(A) Granitic Rock
(B) Basaltic Rock
(C) Rhyolic Rock
(D) Gneissic Rock
36. .is a supercontinent that existed from $\sim 550 \mathrm{Ma}$ to $\sim 180 \mathrm{Ma}$.
(A) Laurasia
(B) Gondwanaland
(C) Columbia
(D) Rodinia
37. Evidence of Iridium anomaly related to K-T boundary has been recorded from the Indian state of. $\qquad$
(A) Madhya Pradesh
(B) Meghalaya
(C) Karnataka
(D) Andhra Pradesh
38. are the joints that are approximately perpendicular to the fold axis.
(A) Pinnate joints
(B) Conjugate joints
(C) Cross joints
(D) Sigmoidal joints
39. The first appearance of the horse Equus in the Siwaliks is at.....
(A) 5.5 Ma
(B) 2.6 Ma
(C) 8.4 Ma
(D) 3.5 Ma
40. The Global Stratotype Section and Point of Neogene-Quaternary Boundary is located in $\qquad$
(A) Guryul Ravine, Kashmir Valley, India
(B) Chicxulub, Mexico
(C) El-Kef, Tunisia
(D) Monte San Nicola Section, Italy
41. $\qquad$ single celled algae with cell wall made up of silica.
(A) Acritarchs
(B) Dinocysts
(C) Diatoms
(D) Chitinozoans
42. $\qquad$ materials are non-magnetic when a magnetic field is absent and magnetic when a magnetic field is applied.
(A) Ferromagnetic
(B) Paramagnetic
(C) Ferrimagnetic
(D) Antiferrimagnetic
43. Dharamsala Formation underlies.
(A) Subathu Formation
(B) Dagshai Formation
(C) Kasauli Formation
(D) Siwalik Group
44. The oldest dates that can be reliably measured using Radiocarbon method is.
(A) $100,000 \mathrm{yrs}$
(B) $500,000 \mathrm{yrs}$
(C) $50,000 \mathrm{yrs}$
(D) 500 yrs
45. Base and precious metal deposits are usually associated with.........
(A) Divergent Plate boundaries
(B) Convergent Plate boundaries
(C) Transform faults
(D) Transcurrent Faults
46. Calcium, Magnesium, Iron, Manganese and Aluminium silicates are commonly associated with
(A) Skarn deposits
(B) Gossan deposits
(C) Placer deposits
(D) Evaporite deposits
47. Pterosaurs were flying
(A) Birds
(B) Mammals
(C) Reptiles
(D) Dinosaurs
48. The southernmost Fault that runs along the Himalayas is called......
(A) Main Boundary Thrust
(B) Main central Thrust
(C) Himalayan Frontal Thrust
(D) Nahan Thrust
49. $\qquad$ is bounded by maximum flooding surface at the top and maximum regressive surface at the base.
(A) High Stand System Tract
(B) Low Stand System Tract
(C) Transgressive System Tract
(D) Regressive System Tract
50. The collision of India with Asia took place sometime between......
(A) $\quad 20-15 \mathrm{Ma}$
(B) $\quad 85-75 \mathrm{Ma}$
(C) $\quad 65-35 \mathrm{Ma}$
(D) $\quad 200-100 \mathrm{Ma}$

## $x-x-x$

## Home Science(Ph.D.)

1. Which of the following statements is not true?
(A) Growth is a biological process
(B) Development is a quantitative process
(C) Education is a goal-oriented process
(D) Learning is a process of behavioural changes
2. Identify the term related to flower arrangement.
(A) Feng Shui
(B) Ying-Yang
(C) Ikebana
(D) Alpana
3. Researcher wants to test the association of female literacy on infant feeding practices. Which of the following test of significance is the most appropriate for this?
(A) F-test
(B) Chi-square test
(C) ' $t$ ' test
(D) Mann-Whitney test
4. Which of the following is not an Embroidery Stitch?
(A) Dabka
(B) Menthi
(C) Phanda
(D) Murri
5. $\qquad$ is claimed to be father of modern management
(A) Henry Fayol
(B) Frederick Taylor
(C) Gross \& Crandall
(D) Nickel \& Dorsey
6. Recommended dietary allowances for Indians are given by $\qquad$
(A) ICAR
(B) ICSSR
(C) CSIR
(D) ICMR
7. The greatest resource in extension work is
(A) Money
(B) Local material
(C) Local people
(D) Contacts with high officials
8. Mean, Median and Mode are:
(A) Deviation
(B) Ways of sampling
(C) Measures of Central tendency
(D) Hypothesis
9. Which of the following is a double pointed dart?
(A) Flange
(B) French
(C) Fish
(D) Dressmakers
10. Rhythm is created through
(A) Proportion
(B) Repetition
(C) Harmony
(D) Balance
11. A young child's inability to distinguish her own cognitive perspective from others' perspectives is
(A) Egocentrism
(B) Centration
(C) Irreversibility of thought
(D) Identity constancy
12. Which of the following does not represent Kasuti Embroidery?
(A) Menthe
(B) Aari
(C) Ganti
(D) Murgi
13. NCPCR is an autonomous body which stands for
(A) National Commission for Protection of Child Rights
(B) National Council for Prevention and Control of Reproduction
(C) National Commission for Poverty control and Relief
(D) National Council for Preschool, Creche and Research
14. If a researcher wants to predict with $99 \%$ accuracy he would set the level of significance at
(A) .05
(B) .95
(C) . 01
(D) . 10
15. A reasoning where we start with certain particular statements and conclude with a universal statement is called
(A) Deductive Reasoning
(B) Inductive Reasoning
(C) Abnormal Reasoning
(D) Transcendental Reasoning
16. Sociometry is a tool to measure child's
(A) Intelligence level
(B) Behaviour problems
(C) Relationship with peers
(D) Relationship with parents
17. Live microorganisms similar to beneficial microorganisms found in the human gut are called as ---------.
(A) Prebiotics
(B) Probiotics
(C) Antibiotics
(D) Symbiotics
18. 'AGMARK is a certificate mark employed on
(A) Poultry \& fisheries products
(B) Agriculture Products
(C) Vegetables
(D) Fruits
19. Which of the following is not a plain weave?
(A) Percale
(B) Calico
(C) Drill
(D) Chintz
20. Compared with mail questionnaires the principal advantage of the personal interview is that it is
(A) Low cost
(B) Depth of information collected
(C) Objectivity
(D) High reliability of information collected
21. A child with average intelligence has an IQ of
(A) $\quad 70-79$
(B) $\quad 80-89$
(C) 90-109
(D) $110-119$
22. Extension Education is the education for the betterment of people for changing their
(A) Behaviour
(B) Status
(C) Income
(D) Values
23. Which of the following is not a hand printing technique?
(A) Duplex
(B) Screen
(C) Block
(D) Stencil
24. Gluten free diet is given to patients with
(A) Crohn's Disease
(B) Celiac Disease
(C) Irritable Bowel Disease
(D) Liver Disease
25. The period of human embryo is
(A) $2-4$ weeks
(B) $1-8$ weeks
(C) $2-8$ weeks
(D) $2-6$ weeks
26. Material used for insulation, radio cabinets and handles etc. is $\qquad$
(A) Bakelite
(B) Plastic
(C) Rubber
(D) Fiber
27. Which of the following garment finishers is used for finishing number of garments together?
(A) Form Press
(B) Tunnel
(C) Buck Press
(D) Die Press
28. One of the most effective methods of imparting skill is
(A) Field Visit
(B) Group Discussion
(C) Demonstration
(D) Project
29. 'Allin' is a compound found in
(A) Garlic
(B) Turmeric
(C) Cinnamon
(D) Cloves
30. A representative sample is used so that the results of a study are
(A) Reliable
(B) Generalized
(C) Convenient
(D) Limited
31. The fact that motor control of the arms precedes control of the legs is an example of
(A) Secular trend
(B) Cephalo-caudal trend
(C) Proximal-distal trend
(D) Evocative trend
32. Which of the following is not a cool colour?
(A) Bluish green
(B) Yellow orange(C)
Blue
(D) Green
33. Area, line, pictoral, pie are the types of -
(A) Charts
(B) Posters
(C) Graphs
(D) Diagrams
34. Following fat is the richest source of MUFA:
(A) Sunflower Oil (B) Soyabean Oil
(C) Coconut Oil
(D) Olive Oil
35. Qualitative Research is used in the situations
(A) Where all sample units of study are homogenous
(B) Where phenomenon under study is cardinal
(C) Where phenomenon under study is nominal
(D) Where it is not possible to measure the phenomenon and responses are subjective
36. $\qquad$ is an intense educational activity for motivating and mobilizing a community to action.
(A) Brainstorming (B)
Colloquium
(C) Campaign
(D) Symposium
37. Anti-ageing vitamin is
(A) Vitamin E
(B) Vitamin K
(C) Vitamin $\mathrm{B}_{12}$
(D) Vitamin A
38. DRDA is functioning at
(A) District level
Block level
(C) Village level
(D) State level
39. Urie Bronfenbrenner's theory is
(A) Humanistic perspective
(B) Sociobiology
(C) Ecological Systems Theory
(D) Socio-cultural Theory
40. Type-I Error occurs if
(A) Null hypothesis is rejected even though it is true
(B) Null hypothesis is accepted even though it is false
(C) Both the null hypothesis as well as alternative hypothesis are rejected
(D) Both the null hypothesis as well as alternative hypothesis are accepted
41. The occurrence of the first menstrual period is known as
(A) Menopause
(B) Menogenesis
(C) Puberty
(D) Menarche
42. Hallmark is a standardized certification of
(A) Jewellery
(B) Eco friendly products
(C) Electrical appliances
(D) Canned Food
43. This nutrient is needed for a healthy immune system and strong connective tissue-----
(A) Fiber
(B) Vitamin K
(C) Vitamin C
(D) Fluoride
44. While washing cotton fabrics, colour can be prevented from bleaching by using
(A) Vinegar
(B) Common Salt
(C) Sodium Carbonate
(D) Lissapol
45. Bread,vinegar, wine and beer are produced with the help of
(A) Mould
(B) Bacteria
(C) Yeast
(D) Enzymes
46. $\qquad$ is a traditional luxury ornamental handicraft of Kashmir
(A) Madhubani
(B) Papier-mache (C)
Modak
(D) Alpana
47. The Act which governs Consumer Disputes Redressal Forum is
(A) The Bureau of indian Standard Act
(B) The consumer Protection Act
(C) The Trade Merchandise Mark Act
(D) Restrictive Trade Practises Act
48. 8th March is celebrated as
(A) International Environment Day
(B) International Sanitation Day
(C) International Women's Day
(D) International Energy Day
49. Which of the following is not a stretch fiber?
(A) Neoprene
(B) Spandex
(C) Elastane
(D) Viscose
50. In a normally distributed population, Mean + 1.96 S.D. will cover $\qquad$ of population.
(A) $90 \%$
(B) $95 \%$
(C) $98 \%$
(D) $99 \%$

## Human Genomics( Ph.D.)

1. Which of the following is the correct reason why liquid media is favoured for culturing thermophilicarchaea?
(A) Liquid media can be heated to higher temperatures
(B) Liquid media is easier to store
(C) Solid media is usually unstable at optimum growing temperatures
(D) Solid media becomes glass-like at high temperatures
2. Which of the following microscopy techniques relies on the specimen interfering with the wavelength of light to produce a high contrast image without the need for dyes or any damage to the sample?
(A) Conventional bright field light microscopy
(B) Phase contrast microscopy
(C) Electron microscopy
(D) Fluorescence microscopy
3. With respect to their surrounding membrane system, which is the odd one out?
(A) Nucleus
(B) Endoplasmic reticulum
(C) Mitochondria
(D) Chloroplasts
4. Within the nucleus, individual chromosomes are thought to occupy discrete territories. Which of the following is most likely to promote this segregation?
(A) Nuclear lamina
(B) Nuclear pore complexes
(C) Nuclear matrix/scaffold
(D) Intermediate fibres
5. The base adenine (A. of the DNA of the fungus Neurosporacrassa forms $23.3 \%$ of its composition. Which of the following is the most likely base composition of the genome, given C is cytosine, G guanine and T thymine?
(A) $\mathrm{A}=\mathrm{T}=23.3 \%$ and $\mathrm{G}=\mathrm{C}=23.3 \%$
(B) $\mathrm{A}=\mathrm{C}=26.7 \%$ and $\mathrm{G}=\mathrm{T}=26.7 \%$
(C) $\mathrm{A}=\mathrm{T}=23.3 \%$ and $\mathrm{G}=\mathrm{C}=26.7 \%$
(D) $\mathrm{A}=\mathrm{T}=26.7 \%$ and $\mathrm{G}=\mathrm{C}=23.3 \%$
6. A population of cells grown in adherent culture contains 0.4 mg protein per 106 cells. Actin comprizes $4.5 \%$ of the total protein. Given the Mr of actin is 42000 and Avogadro's number is $6.02 \times 1023$, which of the following equals the mean number of actin molecules per cell?
(A) $2.58 \times 1014$ actin molecules
(B) $2.58 \times 1011$ actin molecules
(C) $2.58 \times 108$ actin molecules
(D) $2.58 \times 1010$ actin molecules
7. What would the generally expected effect on the PCR reaction be of adjustments that increase the temperature of the annealing phase and the length of the elongation phase?
(A) Precision and yield will be reduced
(B) Precision will be reduced, but yield will be increased
(C) Precision will be increased, but yield will be reduced
(D) Precision and yield will be increased
8. In principle, what outcome would be least expected in a failure to separate pre-PCR and post-PCR activities?
(A) False positive reactions
(B) False negative reactions
(C) Mixed or non-specific PCR products (D)
Increased reliability of PCR results
9. What would the expected effect be on a PCR reaction if the primers used were slightly shorter and more variable than the intended oligonucleotide sequences?
(A) The PCR reaction would not commence
(B) The PCR reaction would end after one cycle
(C) The reaction would generate a single short PCR product
(D) The reaction would yield a mixture of non-specific products
10. You need to use a first generation sequencing method for de novo sequencing, which template should give optimum results for this project?
(A) Genomic DNA
(B) PCR product
(C) Bacterial artificial chromosome
(D) Plasmid DNA
11. Once the sequences are obtained from your Next Generation Sequencing experiment what is the first thing you should do?
(A) Perform a bioinformatics analysis of your data
(B) Check your data using a different method
(C) Publish your results
(D) Further investigate the sequences of interest
12. Which of the following DNA binding proteins interacts with DNA in a sequence specific manner?
(A) Histone H3
(B) DNA polymerase
(C) NF-kB
(D) RNA polymerase
13. Which of the following is an equilibrium method that can be used to accurately determine DNA-protein dissociation constants?
(A) Site directed mutagenesis
(B) Chromatin Immunoprecipitation
(C) ELISA
(D) Footprinting
14. In an EMSA experiment free DNA is separated from protein-DNA complexes in a native gel by which following principle?
(A) Charge
(B) Molecular weight
(C) DNA digestion with DNAse
(D) Antibody immunoprecipitation
15. Which statement best describes the main distinction between the origin of the two classes of small regulatory RNAs: siRNA and miRNA?
(A) siRNAs originate within the cell cytoplasm; miRNAs originate from the cell genome
(B) siRNAs originate from predominantly exogenous dsRNA; miRNAs originate from
the cell genome
(C) miRNAs are expressed whenever siRNAs are unable to appropriately degrade

RNA
sequences
(D) miRNAs are processed from dsRNA viruses, siRNAs are processed from ssRNA viruses
16. For an application where you require a sample of your target protein at high purity, what would be a good purification strategy? Assume that your starting point is E. coli cells in which the target protein fused to an affinity tag has been over-expressed.
(A) Affinity chromatography (AC. followed by size exclusion chromatography (SEC.
(B) AC only
(C) AC followed by ion-exchange (IEX) followed by SEC
(D) AC followed by IEX, followed by hydrophobic interaction (HIC. and then SEC
17. Which of these techniques is often considered a suitable "polishing" step in a protein purification strategy?
(A) Affinity chromatography (AC)
(B) Ion-exchange chromatography (IEX)
(C) Hydrophobic interaction chromatography (HIC)
(D) Size-exclusion chromatography (SEC)
18. What properties of a protein does hydrophobic interaction chromatography exploit for purification?
(A) Charged amino acids
(B) Hydrophobic amino acids on the protein surface
(C) Molecular weight
(D) Enzyme activity
19. In the yeast two-hybrid system, which of the following statements is accurate: A reporter gene
(A) Is fused to the activation domain of a transcription factor
(B) Is fused to the DNA binding domain of a transcription factor
(C) Requires the presence of Histidine in the growth medium for its expression
(D) Is expressed only if the tested protein interaction occurs
20. Which of the following types of genetic manipulations allow a researcher to experimentally increase gene expression in a mouse model?
(A) Knockin
(B) Conditional knockout
(C) Transgenic
(D) Knockout
21. Which one of the following elements is least likely to participate in a hydrogen bond?
(A) O
(B) F
(C) S
(D) N
22. Which one of the following terms describes a positive and negative charge, which are separated in space within a molecule?
(A) Salt bridge
(B) Polar bond
(C) Dipole
(D) Van der Waals interaction
23. The molar mass of an element is equal to which of the following?
(A) Avogadro's number
(B) Atomic number
(C) Mass number
(D) Relative atomic mass
24. If we begin with 8 mL of a glucose solution with a concentration of $0.25 \mathrm{~mol} \mathrm{~L}-1$ and dilute it to a volume of 0.5 L , what is the concentration of the final, diluted solution?
(A) $250 \mathrm{~mol} \mathrm{~L}-1$
(B) $4 \mathrm{~mol} \mathrm{~L}-1$
(C) $0.004 \mathrm{~mol} \mathrm{L-1}(\mathrm{D}) \quad 0.0025 \mathrm{~mol} \mathrm{~L}$ -
25. Only one amino acid has two identical groups attached to its central carbon atom. Which of the following amino acids is it?
(A) Alanine
(B) Valine
(C) Glycine
(D) Cysteine
26. A beta-barrel is an example of what level of structure?
(A) Primary structure
(B) Secondary structure
(C) Tertiary structure
(D) Quaternary structure
27. Zinc finger motifs are a particular characteristic of proteins with which one of the following functions?
(A) Biochemical catalysis (The proteins are enzymes)
(B) Formation of the cell cytoskeleton (The proteins are structural proteins)
(C) Gene regulation (The proteins are DNA-binding proteins)
(D) Signal transduction across the cell membrane (The proteins are transmembrane proteins)
28. NAD+ associates with the enzyme lactate dehydrogenase to catalyse the oxidation of malate. What term is used to describe NAD+ in this context?
(A) Prosthetic group
(B) Coenzyme
(C) Functional group
(D) Intermediate
29. Which of the following statements regarding the Gibbs free energy change for a reaction is false?
(A) The Gibbs free energy change is the proportion of the enthalpy change of a reaction that is used to increase the entropy
(B) If the Gibbs free energy change for a reaction is negative, the reaction happens spontaneously
(C) The Gibbs free energy is represented by the symbol G
(D) A reaction with a negative Gibbs free energy change of reaction is called an exergonic reaction
30. Consider the binding of a protein and its ligand, represented by the expression $\mathrm{P}+\mathrm{L} \leftrightarrows \mathrm{PL}$. Which of the following statements in relation to this binding process is true?
(A) The $K_{d}=\frac{[\mathrm{PL}]}{[\mathrm{P}][\mathrm{L}]}$
(A) The expression for the dissociation constant, Kd , is
(B) A small value of Kd tells us that the protein and ligand bind tightly
(C) If binding is strong, the association reaction lies to the left
(D) If binding is weak, the dissociation reaction lies to the left
31. Which of the following statements regarding enzymes is false?
(A) A given enzyme catalyses just one type of reaction
(B) While most enzymes are proteins, some are composed of RNA
(C) The activity of enzymes is typically impaired at high temperatures
(D) Enzymes act to lower the activation energy of a reaction by stabilising the transition state, but do not participate chemically in the reaction
32. Which one of the following statements regarding Vmax and $K_{M}$ is false?
(A) Vmax is the maximum rate at which a particular enzyme-catalysed reaction can proceed
(B) $\quad \mathrm{K}_{\mathrm{M}}$ is the concentration of substrate at which the rate of the reaction reaches

Vmax
(C) A small value of $K_{M}$ tells us that an enzyme binds strongly to its substrate
(D) A large value of $K_{M}$ tells us that an enzyme shows little specificity for a given substrate
33. Which of the following techniques is used to study the three-dimensional structure of a molecule?
(A) Infra-red spectroscopy
(B) Mass spectrometry
(C) UV-visible spectroscopy
(D) X-ray crystallography
34. A mouse, true breeding for normal gait, was crossed with a mouse, true breeding for an odd gait called 'dancing'. The F1 animals all showed normal gait. What is the most likely proportion of dancing mice in the F2 generation?
(A) $1 / 3$
(B) $1 / 4$
(C) $3 / 4$
(D) 0
35. In goats, being polled (hornless) is dominant to being horned. What are the genotypes of polled parents that give birth to a kid which subsequently grows horns?
(A) Homozygous dominant for polled allele
(B) Homozygous recessive for polled allele
(C) Heterozygous for polled allele
(D) Impossible to know
36. What would be the probability that the five children in a family are all boys?
(A) $1 / 32$
(B) $1 / 64$
(C) $1 / 25$
(D) $1 / 16$
37. The association of histone H 1 with a nucleosome indicates which of the following?
(A) Transcription is occurring
(B) DNA replication is occurring
(C) The DNA is condensed into a 30 nm fibre
(D) The DNA double helix is exposed
38. When calculating a LOD score why do the values vary?
(A) They vary according to both the level of recombination and the level of independent assortment used in the calculation
(B) They vary according to the level of independent assortment used in the calculation
(C) They vary according to the level of recombination used in the calculation
(D) The values never vary
39. Which of the following is the main base methylated in mammalian DNA by the action of DNA methylase?
(A) 7-methyl guanine
(B) 5-methyl cytosine
(C) Methyl adenine
(D) Thymine
40. To which of the following types of sequence does most of the human genome belong?
(A) Pseudogenes
(B) Genes
(C) Tandem repeat sequences
(D) Interspersed repeat sequences
41. Which inherited human disorder results from mutations in the nucleotide excision repair system?
(A) Huntingdon disease
(B) Myotonic dystrophy
(C) Hypermutability syndrome
(D) Xerodermapigmentosum
42. What is the basis of the mutagenic action of the base analogue bromouracil?
(A) It replaces T and binds with G
(B) It replaces G and binds with T
(C) It replaces T and binds with C
(D) It replaces A and binds with C
43. In tropical areas where malaria is prevalent, people who have one allele for beta thalassemia have a selective advantage over people who have no or two alleles for beta thalassemia. What is this an example of?
(A) Frequency dependent selection
(B) Diversifying selection
(C) Hybrid vigour
(D) Heterozygous advantage
44. On what does natural selection act?
(A) Phenotype
(B) Genotype
(C) A population's gene pool
(D) Homozygous dominant and heterozygous individuals
45. Which of the following strategies can ensure production of a cloned human gene in a bacterium?
(A) Use of a fusion plasmid/human viral vector
(B) Additional insertion of a human origin of replication
(C) Cloning into a RNA phage
(D) Insertion of the cDNA sequence
46. Why do molecular biologists sometimes compare cytochrome oxidase I sequences from different sources?
(A) To investigate gene function
(B) To identify stem cells
(C) To investigate evolutionary relationships
(D) To map genes
47. Which of the following occurs during pyrosequencing?
(A) Dideoxynucleotides release fluorescent bases
(B) Dideoxynucleotides are incorporated and terminate DNA synthesis
(C) A released dideoxynucletode generates a fluorescent signal
(D) A released pyrophosphate generates a fluorescent signal
48. Which of the following occurs when a knockout mouse is produced?
(A) A mutant gene is replaced by a functional allele
(B) A functional gene is replace by a mutant allele
(C) A functional gene is inserted in addition to the mutant allele
(D) A mutant gene is inserted in addition to the functional allele
49. For which of the following is PCR not used?
(A) Site specific mutagenesis
(B) To generate double stranded DNA for DNA sequencing
(C) To generate copies of microsatellites for DNA fingerprinting
(D) To generate cDNA from mRNA
50. Which mode of natural selection is most important in maintaining genetic variation at single loci?
(A) Overdominance
(B) Positive frequency-dependent selection
(C) Negative frequency-dependent selection
(D) Underdominance

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## Medical Physics(Ph.D.)

1. Which characteristic increases with increasing photon energy?
(A) Wavelength
(B) Frequency
(C) Mass
(D) Charge
2. Electron capture can result in emission of
(A) Antineutrinos
(B) High-LET radiation
(C) Characteristic x-rays
(D) Positrons
3. If the distance from a radiation source is halved, the radiation intensity will
(A) Increase by $2 \%$
(B) Increase by $50 \%$
(C) Double
(D) Quadruple
4. After $\mathbf{2 4}$ hours, the activity of a $\mathbf{1 0 0} \mathbf{~ M B q}{ }^{\mathbf{1 2 3}} \mathbf{I}\left(\mathbf{T}_{1 / 2}=\mathbf{1 3}\right.$ hours) source will be about
(A) 50 MBq
(B) 25 MBq
(C) 10 MBq
(D) 5 MBq
5. Which of the following is measured in newtons?
(A) Electrons flowing through a medium
(B) Attraction or repulsion between two bodies
(C) Mass
(D) Electric resistance
6. Which of the following decay modes changes the mass number (A) of an unstable nucleus?
(A) Electron capture
(B) Beta positive Decay
(C) Alpha decay
(D) Isomeric transition
7. The maximum photon energy in a x-ray beam is determined by
(A) Voltage across the x-ray tube
(B) Atomic number of the x-ray beam filter
(C) Current flowing through the x-ray tube
(D) Total exposure time (seconds)
8. The heel effect is more pronounced
(A) At larger distances from the focal spot
(B) With a larger target (anode) angle
(C) With a smaller anode angle
(D) At the cathode edge of the $x$-ray field
9. Targets for production of $x$-rays have
(A) Low atomic numbers (Z)
(B) Air cooling
(C) Beryllium covering
(D) High heat capacities
10. The ratio of heat to $x$-rays produced in a x-ray tube is about
(A) 1:99
(B) $99: 1$
(C) $50: 50$
(D) $90: 10$
11. All of the following could affect the HVL of an x-ray beam except
(A) Tube voltage
(B) Voltage ripple
(C) Tube current
(D) Anode angle
12. Which of the following is not a component of an image intensifier?
(A) Anode
(B) Input phosphor
(C) Photocathode
(D) Photomultiplier tube
13. High ratio grids increase all the following except
(A) Screen/film speed
(B) Image contrast
(C) Patient dose
(D) Removal of scatter
14. Which of the following factors would have the least effect on image sharpness?
(A) Film type
(B) Focal spot size
(C) Screen/film contact
(D) Screen thickness
15. The patient integral dose does not depend on the
(A) Skin dose
(B) Beam area
(C) Organ sensitivity
(D) Patient thickness
16. Which of the following is not true for Poisson distributions?
(A) They are used to describe radioactive decay
(B) They are used to describe quantum mottle
(C) The variance is equal to the mean
(D) They are always symmetrical
17. A ROC curve is used to measure diagnostic imaging
(A) Performance
(B) Accuracy
(C) Specificity
(D) Sensitivity
18. How many bits are required to store 512 shades of gray?
(A) 6
(B) 8
(C) 9
(D) 10
19. Input devices for a computer do not include
(A) Keyboard
(B) Trackball
(C) Touch screen
(D) Array processor
20. The Nyquist frequency for a 1 k digital photospot image ( $\mathbf{2 5} \mathrm{cm}$ image intensifier size) is
(A) $1 \mathrm{lp} / \mathrm{mm}$
(B) $2 \mathrm{lp} / \mathrm{mm}$
(C) $4 \mathrm{lp} / \mathrm{mm}$
(D) $81 \mathrm{p} / \mathrm{mm}$

## 21. Breast compression in mammography

(A) Improves image contrast
(B) Eliminates the need for a grid
(C) Requires the use of a wide-latitude film
(D) Increases radiation dose
22. Breast imaging using MRI would not use
(A) Fat-suppression techniques
(B) Special breast coils
(C) lodine contrast
(D) Three-dimensional imaging techniques
23. The use of thermography to detect breast cancer
(A) Involves ionizing radiation
(B) Uses thermoluminescent dosimeters
(C) Is most effective near the chest wall
(D) Is deemed by the ACR to be ineffective
24. The fundamental measurement made by a CT scanner is the
(A) Sorting of CT numbers
(B) Determination of gray scale
(C) Pixel density
(D) Relative x-ray attenuation
25. Which of the following is not a source of CT artifacts?
(A) Patient motion
(B) Metal implants
(C) Beam hardening
(D) Low tube current
26. CT scanner spatial resolution could improve with an increase of
(A) Reconstruction matrix
(B) Detector elements size
(C) Focal spot size
(D) Scan time
27. The pulse height analyzer in NM imaging increases
(A) Detector efficiency
(B) Scattered photons
(C) Contrast-to-noise ratio
(D) Count rate
28. Following administration of ${ }^{131}$ It a patient, the dose rate near the patient does not depend on
(A) Administered activity
(B) Patient age
(C) Effective half-life
(D) Distance to patient
29. The variance of a NM image pixel with a 100 count would be
(A) 10
(B) 20
(C) 50
(D) 100
30. Which of the following does not concern itself with radiation risk estimates?
(A) ICRP
(B) UNSCEAR
(C) BEIR
(D) ICRU
31. An ultrasound beam traveling through tissue cannot be
(A) Absorbed
(B) Amplified
(C) Scattered
(D) Reflected
32. Contrast in MR can be due to all the following differences except
(A) Presence of flow
(B) Proton density
(C) T 1
(D) Atomic number
33. MR spectroscopy is used to detect all the following except:
(A) ${ }^{31} \mathrm{P}$
(B) ${ }^{32} \mathrm{P}$
(C) Inorganic phosphate
(D) Phosphocreatinine
34. The semi-interquartile range is most closely related to the
(A) Median
(B) Mean
(C) Mode
(D) None of the above
35. Parity is not conserved in
(A) Alpha-decay
(B) Beta-decay
(C) Gamma-decay
(D) None of the above
36. Which of the following factors would be most appropriate to produce a T1-weighted image in MRI?
(A) $\mathrm{TR}=500, \mathrm{TE}=20$
(B) $\mathrm{TR}=2,000, \mathrm{TE}=20$
(C) $\mathrm{TR}=2,000, \mathrm{TE}=100$
(D) $\mathrm{TR}=500, \mathrm{TE}=100$
37. Which enzyme is activated during double stranded break in DNA?
(A) DNA polymerase
(B) Klenow fragment
(C) RNA polymerase
(D) Translesional polymerase
38. The main source of solar energy is
(A) Nuclear fission
(B) Nuclear fusion
(C) Gravitational contraction
(D) Combination of coal and hydrogen
39. Transverse electric (TE) waves have
(A) Magnetic field component H in the direction of propagation
(B) Electric field component E in the direction of propagation
(C) Magnetic field component H in the direction of propagation and no component of electric field E in this direction
(D) Electric field component E in the direction of propagation and no component of magnetic field H in this direction
40. In GM counter experiment the measured data is 3600 , the statistical error quoted with $95 \%$ confidence level will be will be
(A) 180
(B) 60
(C) 120
(D) 104
41. $\overline{\overline{\mathrm{AB}}+\overline{\mathrm{AC}}}$ is equivalent to
(A) $\mathrm{A}+\mathrm{B}+\mathrm{C}$
(B) ABC
(C) $A \overline{B C}$
(D) $A B \bar{C}$
42. The correct order of increasing wavelength is
(A) X-rays, IR, Microwave, Visible
(B) UV, IR, Microwave, X-rays
(C) Microwave, X-rays, UV, IR
(D) X-rays, UV, IR, Microwave
43. In an n-p-n transistor biased for operation in forward active region
(A) emitter is positive with respect to base
(B) collector is positive with respect to base
(C) base is positive with respect to emitter and collector is positive with respect to base
(D) none of the above
44. A uniform plane wave is one in which
(A) $\overrightarrow{\mathrm{E}} \times \overrightarrow{\mathrm{H}}=0$
(B) $\overrightarrow{\mathrm{E}} \cdot \overrightarrow{\mathrm{H}}=0$
(C) $\overrightarrow{\mathrm{E}}$ and $\overrightarrow{\mathrm{H}}$ are perpendicular
(D) $\overrightarrow{\mathrm{E}}$ and $\overrightarrow{\mathrm{H}}$ lie in a plane and are perpendicular to each other
45. Which of the following is true as regards photo emission?
(A) Rate of photo emission is inversely proportional to light intensity
(B) Maximum velocity of electron increases with decreasing wave length
(C) Both holes and electrons are produced
(D) Velocity of emitted electrons is dependent on light intensity
46. Given $\vec{A}, \vec{B}$ and $\vec{C}$ are the translational vectors in case of unit cell of a lattice in solid.The volume of the unit cell is
(A) $|(\vec{A} \times \vec{B}) \cdot \vec{C}|$
(B) $|\vec{A} \cdot \vec{B} \cdot \vec{C}|$
(C) $|(\vec{A} \times \vec{B})|$
(D) $|(\vec{A} \times \vec{B}) \times \vec{C}|$
47. A Zener diode
(A) has a constant current in the breakdown region
(B) has a constant voltage in the breakdown region
(C) has a constant current in the forward region
(D) has a constant voltage and constant current in the breakdown region
48. The parent radionuclide of the $A=4 n+2$ radioactive series is
(A) ${ }^{238} \mathrm{U}$
(B) ${ }^{232} \mathrm{Th}$
(C) ${ }^{232} \mathrm{U}$
(D) ${ }^{238} \mathrm{Th}$
49. In common-base configuration, the output resistance is given by
(A) $\left[\frac{\Delta V_{B E}}{\Delta I_{E}}\right]_{\mathrm{I}_{\mathrm{C}}=\text { constant }}$
(B) $\left[\frac{\Delta \mathrm{V}_{\mathrm{BE}}}{\Delta \mathrm{I}_{\mathrm{E}}}\right]_{\mathrm{V}_{\mathrm{CE}}=\text { constant }}$
(C) $\left[\frac{\Delta V_{C E}}{\Delta \mathrm{I}_{\mathrm{C}}}\right]_{\mathrm{I}_{\mathrm{B}}=\text { constant }}$
(D) $\left[\frac{\Delta \mathrm{V}_{\mathrm{CB}}}{\Delta \mathrm{I}_{\mathrm{C}}}\right]_{\mathrm{I}_{\mathrm{E}}=\text { constant }}$
50. The unit of magnetic flux in SI system is
(A) Maxwell
(B) Tesla
(C) Weber
(D) Gauss

## Nuclear Medicine( Ph.D.)

1. 1 terabecquerel is equal to
(A) $10^{9}$ disintegrations/Second
(B) $10^{12}$ disintegrations/Second
(C) $10^{15}$ disintegrations/Second
(D) $10^{18}$ disintegrations/Second
2. Radiation weighting factor (Wr) for fast neutrons
(A) 5
(B) 20
(C) $\quad 1-20$
(D) $\quad 5-20$
3. The artificial radioactivity was discovered by
(A) Henri Becquerel
(B) Irene Curie and F. Joliot
(C) Blumgart
(D) Rutherford
4. The ejection fraction of the gall bladder can be evaluated using
(A) Cimitidine
(B) Dipyridamole
(C) Cholecystokinin
(D) Dobutamine
5. Radiocolloids are cleared from the circulation by
(A) Liver parenchymal cells
(B) Kupffer cells
(C) Hepatocytes
(D) Hemangiomas
6. Colloid shift refers to
(A) Small colloid particles clumped together to form large particles which localizes in the lungs
(B) 99 mTc sulphur colloid changing into albumin colloid
(C) Increased uptake of the colloid in the spleen and bone marrow relative to liver
(D) The redistribution of colloid within the liver over time
7. The energy of beta particle for strontium-90 is
(A) 346.2 Kev
(B) 546.2 Kev
(C) 746.2 Kev
(D) 946 Kev
8. The energy of gamma photon from samarium-153 is
(A) $\quad 73.2 \mathrm{Kev}(B) \quad 932 \mathrm{Kev}$ (C) $\quad 103.2 \mathrm{Kev}$
(D) $\quad 113.2 \mathrm{Kev}$
9. The $P$ wave of electrocardiogram is associated with
(A) Contraction of Atria
(B) Depolarisation of atrial muscular tissue
(C) Contraction of ventricles
(D) Repolarisation of SA node
10. What is the usual particle size of sulphur colloid?
(A) $\quad 0.3-1.0 \mu \mathrm{~m}$
(B) $\quad 0.03-0.1 \mu \mathrm{~m}$
(C) $\quad 2.0-10 \mu \mathrm{~m}$
(D) $\quad 4.0-15 \mu \mathrm{~m}$
11. How many MBq will correspond to $50 \mu \mathrm{ci}$ of activity?
(A) $\quad 1.85 \mathrm{MBq}$
(B) $\quad 0.85 \mathrm{MBq}$
(C) $\quad 18.5 \mathrm{MBq}$
(D) $\quad 0.185 \mathrm{MBq}$
12. Which of the following cannot be evaluated by a MUGA scan?
(A) Wall motion
(B) Aneurismal flow
(C) Vessel patency
(D) Wall thickness
13. The normal half-time of gastric emptying of a a labelled solid meal is
(A) $\quad 45-60$ minutes
(B) $\quad$ 75-90 minutes
(C) $\quad 90-120$ minutes
(D) $\quad$ 120-180 minutes
14. What is the mechanism of ${ }^{201} \mathrm{TI}$ decay?
(A) Electron capture
(B) Internal conversion
(C) Positron decay
(D) Isomeric transition
15. If a radiation worker has to spend a full 40-hrs working week in a particular area, what is the maximum dose rate which can be allowed?
(A) $5 \mu \mathrm{~Sv} / \mathrm{hr}$
(B) $10 \mu \mathrm{~Sv} / \mathrm{hr}$
(C) $12 \mu \mathrm{~Sv} / \mathrm{hr}$
(D) $\quad 15 \mu \mathrm{~Sv} / \mathrm{hr}$
16. The dose rate at 2 m from a particular gamma source is $400 \mu \mathrm{~Sv} / \mathrm{hr}$. At what distance will it give a dose rate of $25 \mu \mathrm{~Sv} / \mathrm{hr}$ ?
(A) 6 m
(B) 7 m
(C) 8 m
(D) 10 m
17. What is the biological half life of $99 \mathrm{mTc}-\mathrm{MAA}$ in lungs?
(A) 30 minutes
(B) $\quad 1-1.5 \mathrm{hrs}$
(C) $\quad 2-3 \mathrm{hrs}$
(D) $\quad 3-4 \mathrm{hrs}$
18. Normal gallium scan findings at 72 hours may include uptake in all of the following except
(A) Salivary glands
(B) Lacrimal glands
(C) Liver
(D) Kidney
19. What fraction of an intravenous dose of 99 mTc -sulphur colloid localise in the liver, spleen and bone marrow, respectively?
(A) $85 \% ; 10 \%, 5 \%$
(B) $75 \% ; 15 \%, 10 \%$
(C) $50 \% ; 40 \%, 10 \%$
(D) $45 \% ; 50 \%, 5 \%$
20. If excessive aluminium is present in ${ }^{99 m} \mathrm{Tc}$ eluate, which one of the following would be expected on a bone scan
(A) Lung uptake
(B) Liver uptake
(C) Thyroid uptake
(D) Gastric uptake
21. If the lead HVL for ${ }^{99 \mathrm{~m}} \mathrm{Tc}$ is 2.6 mm and a lead shield containing ${ }^{99 \mathrm{~m}} \mathrm{Tc}$ eluate is 13 mm thick, what will the exposure rate be from the shielded vial, if the unshielded vial had a rate of $100 \mathrm{mR} / \mathrm{hr}$ ?
(A) $\quad 1.6 \mathrm{mR} / \mathrm{hr}$
(B) $\quad 3.1 \mathrm{mR} / \mathrm{hr}$
(C) $\quad 6.3 \mathrm{mR} / \mathrm{hr}$
(D) $\quad 12.5 \mathrm{mR} / \mathrm{hr}$
22. What is the distance required to reduce the radiation level from a $20 \mathrm{MBq}{ }^{60} \mathrm{Co}$ source to 0.025 $\mathrm{mGy} / \mathrm{h}$ ? (exposure rate constant for ${ }^{60} \mathrm{Co}=3.1 \mathrm{mGy} / \mathrm{h} / \mathrm{MBq}$ at 1 cm )
(A) 49.8 cm (B)
40.8 cm (C)
14.8 cm (D)
33.8 cm
23. The tissue weighing factor for brain is
(A) 0.12
(B) 0.08
(C) 0.04
(D) 0.01
24. Proteins are separated by SDS-electrophoresis on the basis of their
(A) Size
(B) Charge
(C) Amino acid composition
(D) Charge and shape
25. In Scanning Electron Microscope (SEM), to form an image of the specimen
(A) Electron should pass through the specimen
(B) Electrons are scattered from the surface
(C) A thin film of heavy metal is evaporated
(D) Specimens are stained
26. The tertiary structure of protein is detected by
(A) X-ray crystallography
(C) Electrophoresis
(D) Chromatography
27. ELISA assay uses
(A) An enzyme which can react with a secondary antibody
(B) An enzyme which can react with the antigen
(C) A substrate which gets converted to a colored product
(D) A radiolabelled secondary antibody
28. Which spectroscopy is used to detect -SH group and disulphide linkages in proteins
(A) CD spectroscopy
(B) Fluorescence spectroscopy
(C) NMR spectroscopy
(D) FTIR spectroscopy
29. Maximum limit on total discharge per day in sanitary sewage system for ${ }^{125}$ Is
(A) 0.37 MBq
(B)
3.7MBq(C)
0.037 MBq
(D) 37 MBq
30. The exposure rate at the surface of a package to be shipped is $50 \mathrm{mrem} / \mathrm{hr}$. What label is required?
(A) DOT Radioactive White I
(B) DOT Radioactive Yellow II
(C) DOT Radioactive Yellow III
(D) no radioactive label is required
31. The most sensitive stage for the lethal effects of radiation is:
(A) Preimplantation
(B) Early organogenesis
(C) Late organogenesis
(D) The fetal period
32. The enzyme responsible for continuing DNA replication in prokaryotes, once it is initiated is
(A) DNA polymerase I
(B) DNA polymerase III
(C) Polymerase beta
(D) DNA Gyrase
33. Northern blotting is used for separation of
(A) DNA
(B) Mrna
(C) Protein
(D) Plasmids
34. In isoelectric focusing, proteins are separated
(A) In a pH gradient
(B) In a salt gradient
(C) In a density gradient
(D) In a temperature gradient
35. The enzyme used to form a phosphodiester bond in a nick between a 3'end of one DNA chain and a 5 end of another is
(A) DNA polymerase
(B) Restriction Endonuclease
(C) S1 nuclease
(D) DNA ligase
36. Which of the following techniques is primarily undertaken to amplify DNA?
(A) Polymerase chain reaction
(B) Microarrays
(C) Northern Blotting
(D) Southern Blotting
37. Electrodessication is used to destroy tissue by
(A) High frequency positrons
(B) Low frequency positrons
(C) High frequency photoelectrons (D)
High frequency electric current
38. Malondialdehyde is a degradation product of
(A) Peroxidised lipids
(B) Peroxidised proteins
(C) Glucose metabolism
(D) Carbohydrate metabolism
39. Which of the following statement is not true for $X$-rays
(A) $\quad X$-rays are electromagnetic radiations
(B) $\quad X$-rays have a speed of sound
(C) Produced by stopping high energy electrons in a tungsten target
(D) Intensity falls of in accordance with the inverse square law
40. Which of the following heals the quickest after injury:
(A) Bone
(B) Epithelium
(C) Cartilage
(D) Muscle
41. Which of the following suffixes implies "growth" or "formation":
(A) -blast
(B) -lemma (C)
-stasis (D) -cyte
42. If the concentration of glucose in the water outside of a cell is higher than the concentration inside,
(A) Water will tend to enter the cell by osmosis
(B) Water will tend to leave the cell by osmosis
(C) Glucose will tend to enter the cell by osmosis
(D) Glucose will tend to leave the cell by osmosis
43. Times a proton is heavier than an electron is
(A) 1827
(B) 1876
(C) 1836
(D) 1789
44. What is the resting membrane potential of a neuron?
(A) -70 Mv
(B) -65 Mv
(C) $\quad-80 \mathrm{Mv}$
(D) $\quad-55 \mathrm{mV}$
45. Which of the following is not a chemical radiosensitizer?
(A) Nucleotide analogues
(B) Electronic affinic compounds
(C) Nitroimidazoles
(D) Aminothiols
46. The material used for absorbing excess neutrons in a nuclear reactor is
(A) Cadmium
(B) Neodymium
(C) Vanadium
(D) Indium
47. Energy of thermal neutrons is
(A) 0.50 ev
(B) 0.05 ev
(C) 0.25 ev
(D) 0.025 ev
48. $\operatorname{In} \beta^{+}$decay, nucleon number is
(A) Conserved
(B) Not conserved
(C)
Unstable
(D) Stable
49. Phenomena of radioactivity was discovered in
(A) 1893
(B) 1894
(C) 1895
(D) 1896
50. Heavy nuclei have
(A) More protons than neutrons
(B) More electrons than neutrons
(C) More neutrons than electrons
(D) More neutrons than protons

## Statistics(Ph.D. \& M.Phil.)

1. Let $A=\left(\begin{array}{ll}a & b \\ c & d\end{array}\right)$ be a matrix such that $A^{3}=O_{2 \times 2}$, but $A \neq O_{2 \times 2}$, then
(A) $A^{2}=O_{2 \times 2}$
(B) $A^{2}=A$
(C) $A^{2}=I-A$
(D) $A^{2}=I+A$
2. The inverse of a skew symmetric matrix of odd order is
(A) A symmetric matrix
(B) A skew symmetric matrix
(C) Diagonal matrix
(D) Does not exist
3. If $A=\left(\begin{array}{lll}a_{1} & b_{1} & c_{1} \\ a_{2} & b_{2} & c_{2} \\ a_{3} & b_{3} & c_{3}\end{array}\right)$ and $|A| \neq 0$, then the system of equations

$$
a_{1} x+b_{1} y+c_{1} z=0, a_{2} x+b_{2} y+c_{2} z=0 \text { and } a_{3} x+b_{3} y+c_{3} z=0 \text { has }
$$

(A) Only one solution
(B) Infinite number of solutions
(C) No solution
(D) More than one but finite number of solutions
4. One hundred identical coins, each with probability $p$ of showing heads are tossed once. If $0<p<1$ and the probability of heads showing on 50 coins is equal to that of heads showing on 51 coins, the value of $p$ is
(A) $1 / 2$
(B) $51 / 101$
(C) 49/101
(D) $3 / 101$
5. Let $A, B$ and $C$ be three mutually independent events. Consider the two statements $S_{1}$ and $S_{2}$ :

$$
\begin{aligned}
& S_{1}: A \text { and } B \cup C \text { are independent } \\
& S_{2}: A \text { and } B \cap C \text { are independent }
\end{aligned}
$$

Then
(A) Both $S_{1}$ and $S_{2}$ are true
(B) Only $S_{1}$ is true
(C) Only $S_{2}$ is true
(D) Neither $S_{1}$ nor $S_{2}$ is true
6. A letter is taken at random from the letters of the word 'STATISTICS' and another letter is taken at random from the letters of the word 'ASSISTANT'. The probability that they are the same letter is
(A) $1 / 45$
(B) $5 / 18$
(C) $13 / 90$
(D) $19 / 90$
7. Each of two persons $A$ and $B$ toss three fair coins. The probability that both get the same number of heads is
(A) $3 / 8$
(B) $1 / 9$
(C) $5 / 16$
(D) $7 / 16$
8. A letter is known to have come from either TATANAGAR or CALCUTTA. On the envelope, just two consecutive letters, TA, are visible. The probability that the letter has come from CALCUTTA is
(A) $4 / 11$
(B) $1 / 3$
(C) $5 / 12$
(D) $1 / 7$
9. If the regression line of $Y$ on $X$ is $Y+0.8 X=25$ and the standard deviations of $X$ and $Y$ are respectively 3 and 8 , then the value of correlation coefficient $r$ is
(A) $\quad-0.3$
(B) $\quad-0.4$
(C) 0.3
(D) 0.4
10. Suppose $r$ is the correlation coefficient between two variables $X$ and $Y$ where standard deviation of $X$ and $Y$ are equal. If $\theta$ is the angle between the regression lines, then
(A) $\quad \tan \theta=\frac{1+r^{2}}{2 r}$
(B) $\quad \tan \theta=\frac{1-r^{2}}{2 r}$
(C) $\tan \theta=\frac{1+r^{2}}{r}$
(D) $\tan \theta=\frac{1-r^{2}}{r}$
11. The joint probability density function of $(X, Y)$ is $f(x, y)=\exp \{-(x+y)\}$, for $0<x<\infty$ and $0<y<\infty$.

Assertion(A): $P(X<Y / X<2 Y)=P(X<Y)$

Reason $(R): X$ and $Y$ are independently exponentially distributed which possesses 'lack of memory property'

Select your answer from the following codes:
(A) Both $(A)$ and $(R)$ are true and $(R)$ is correct explanation of (A)
(B) Both $(A)$ and $(R)$ are true but $(R)$ is not correct explanation of $(A)$
(C) $\quad(A)$ is true but $(R)$ is false
(D) (A)is false but $(R)$ is true
12. The joint probability mass function of random variables $X$ and $Y$ is

$$
f(x, y)=\frac{\lambda^{x} e^{-\lambda} p^{y}(1-p)^{x-y}}{y!(x-y)!}, y=0,1, \ldots, x ; x=0,1, \ldots .
$$

The marginal distribution of
(A) X and Y both are Poisson
(B) X and Y are binomial
(C) X is binomial and that of Y is Poisson
(D) X is Poisson and that of Y is binomial
13. Let X be a random variable

Assertion $(A): E\left(X^{2}\right) \geq(E(X))^{2}$

Reason ( $R$ ): $X^{2}$ is convex function of $X$

Select your answer from the following codes:
(A) (A) is true but ( R ) is false
(B) (A) is false but (R) is true
(C) Both (A) and (R) are true and (R) is correct explanation of (A)
(D) Both (A) and (R) are true but (R) is not correct explanation of (A)
14. For normal distribution, quartile deviation, the mean deviation and standard deviation are approximately
(A) 1:2:3
(B) $\frac{1}{2}: 3: 5$
(C) 10:12:15
(D) $1: 1: 1$
15. The hypergeometric distribution with parameters $\mathrm{N}, \mathrm{M}$ and n , for $N \rightarrow \infty, \frac{M}{N} \rightarrow p, 0<p<1$, reduces to the following distribution
(A)
Gamma(B) Geometric
(C) Binomial
(D) Normal
16. If X and Y are two independent Poisson variates such that $X \sim P(1)$ and $Y \sim P(2)$, the probability, $P(X+Y<3)$ is
(A) $8.5 e^{-3}$
(B) $4 e^{-3}$
(C) $e^{-3}$
(D) $3 e^{-3}$
17. If $\phi(t)$ is characteristic function, which of the following is incorrect?
(A) $\quad|\phi(t)| \leq 1$
(B) $\quad \phi(t)$ is continuous everywhere on real line
(C) $\quad \phi(0)=1$
(D) $|\phi(t)|>1$
18. If X and Y are independent exponential random variable with the same mean $\lambda$, then the distribution of $\min (X, Y)$ is
(A) Exponential with mean $\lambda / 2$
(B) Exponential with mean $2 \lambda$
(C) Exponential with mean $\lambda$
(D) Laplace distribution with mean $\lambda$
19. Mean and variance of standard Logistic distribution are
(A) $\quad$ Mean $=0$, Variance $=\frac{\pi^{2}}{3}$
(B) $\quad$ Mean $=0$, Variance $=\frac{\pi^{2}}{4}$
(C) Mean $=0$, Variance $=\frac{2 \pi^{2}}{3}$
(D) $\quad$ Mean $=0$, Variance $=\frac{3 \pi^{2}}{2}$
20. If the coefficient of the study variable Y and auxiliary variable X in a population are 18 and 32 respectively, then for what range of the coefficient of correlation $(\rho)$ between X and Y , the ratio method of estimation will be preferable over sample mean (assuming that $\mathrm{R}>0$ )?
(A) $\quad \rho<0.63$
(B) $0.33<\rho<0.80$
(C) $\quad \rho>0.88$
(D) $0.63<\rho<0.85$
21. Let X be a geometric distribution with parameter $\theta$ and the experimenter is to decide between $H_{0}: \theta=\theta_{0}=0.5$ against $\theta=\theta_{1}=0.1$. We accept $H_{0}$ if $\{X \leq 2\}$ and reject it if $\{X>2\}$. Then, the probability of Type-II error is
(A) 0.81
(B) 0.19
(C) 0.25
(D) 0.75
22. Let $X_{n}$ be a sequence of random variables and $X$ is a random variable. Further let $k$ be a constant.

Assertion (A): $X_{n}$ converges in probability to $X \Rightarrow X_{n}$ converges in distribution to $X$
Assertion ( $B$ ): $X_{n}$ converges in distribution to $k \Rightarrow X_{n}$ converges in probability to $k$
Then
(A) Both $A$ and $B$ are true
(B) Only $A$ is true
(C) Only $B$ is true
(D) Neither $A$ nor $B$ is true
23. If the population standard deviation is known to be 5 for a population containing 80 units, then the standard error of sample mean for a simple random sample of size 25 without replacement is
(A) 0.55
(B) 0.20
(C) 0.35
(D) 0.85
24. In a railway-marshaling yard, goods trains arrive at a rate of 30 trains per day. Assume that the inter-arrival time follows an exponential distribution and the service time (the time taken to hump a train) distribution is also exponential with an average of 36 minute. Then the average number of trains in the queue is
(A) 2
(B) 3
(C) 4
(D) 5
25. The error degrees of freedom for $4 \times 4$ Graeco-Latin square design is
(A) 6
(B) 5
(C) 4
(D) 3
26. Let $X_{1}, \ldots, X_{n}$ be a random sample of size $n$ from Uniform $(0, \theta)$ distribution, where $\theta$ is unknown. The maximum likelihood estimator of $\theta$ based on this sample is
(A) $\min \left(X_{1}, \ldots, X_{n}\right)$
(B) $\max \left(X_{1}, \ldots, X_{n}\right)$
(C) $\frac{\min \left(X_{1}, \ldots, X_{n}\right)+\max \left(X_{1}, \ldots, X_{n}\right)}{2}$
(D) $\frac{\max \left(X_{1}, \ldots, X_{n}\right)-\min \left(X_{1}, \ldots, X_{n}\right)}{2}$
27. Let $X_{1}, \ldots, X_{n}$ be independent and identically distributed as Poisson distribution with parameter $\theta$. Let $(\theta)=\frac{e^{-\theta} \theta^{3}}{3!}, T=\sum_{i=1}^{n} X_{i}$ and $\binom{T}{3}<0$ for $T<3$. Then, the unique minimum variance unbiased estimator of $\tau(\theta)$ is
(A) $\quad \phi(T)=\binom{T}{3}\left(\frac{1}{n}\right)^{3}\left(1-\frac{1}{n}\right)^{T-3}$
(B) $\quad \phi(T)=\binom{T}{2}\left(\frac{1}{n}\right)^{2}\left(1-\frac{1}{n}\right)^{T-2}$
(C) $\quad \phi(T)=\binom{T}{1}\left(\frac{1}{n}\right)^{1}\left(1-\frac{1}{n}\right)^{T-1}$
(D) $\quad \phi(T)=\binom{T}{0}\left(\frac{1}{n}\right)^{0}\left(1-\frac{1}{n}\right)^{T-0}$
28. Consider the linear programming (LP) problem - maximize $x_{1}+x_{2}$ subject to

$$
\begin{aligned}
& x_{1}-2 x_{2} \leq 10 \\
& x_{2}-2 x_{1} \leq 10 \\
& x_{1} \geq 0, x_{2} \geq 0
\end{aligned}
$$

Then
(A) The LP problem admits an optimal solution
(B) The LP problem is unbounded
(C) The LP problem admits no feasible solution
(D) The LP problem admits a unique feasible solution
29. Which of the following relations is incorrect in the context of Cost of Living Index Number (CLIN)
(A) $\quad$ Real Wages $=\frac{\text { Money Wages }}{\text { CLIN }} \times 100$
(B) Purchasing Power of Money $=\frac{1}{C L I N}$
(C) Real Wages $=\frac{\text { CLIN }}{\text { Money Wages }} \times 100$
(D) CLIN $=\frac{\text { Total Expenditure in current year with base year quantities as weights }}{\text { Total expenditure in base year }} \times 100$
30. Suppose $X_{1}, X_{2}, \ldots$. is a sequence of $i . i . d$. random variables with common variance $\sigma^{2}>0$. Let $Y_{n}=\frac{1}{n} \sum_{i=1}^{n} X_{2 i-1} \operatorname{and} Z_{n}=\frac{1}{n} \sum_{i=1}^{n} X_{2 i}$. Then, the asymptotic distribution of $\sqrt{n}\left(Y_{n}-Z_{n}\right)$ as $n \rightarrow \infty$ is
(A) $\quad N(0,1)$
(B) $\quad N\left(0, \sigma^{2}\right)$
(C) $\quad N\left(0,2 \sigma^{2}\right)$
(D) $\quad N\left(0,3 \sigma^{2}\right)$
31. Let X be a discrete random variable taking non-negative integer values in a set E . Let $P(X>a+$ $b / X>a)=P(X>b)$ for any two positive integers $a, b \in E$. Then, which of the following is a possible distribution of X ?
(A) Negative Binomial
(B) Geometric
(C) Binomial
(D) Possion
32. Let $Y_{1}, Y_{2}, Y_{3}$ and $Y_{4}$ be four random variables such that $E\left(Y_{1}\right)=\theta_{1}-\theta_{3} ; E\left(Y_{2}\right)=\theta_{1}+\theta_{2}-$ $\theta_{3} ; E\left(Y_{3}\right)=\theta_{1}-\theta_{3} ; E\left(Y_{4}\right)=\theta_{1}-\theta_{2}-\theta_{3}$, where $\theta_{1}, \theta_{2}, \theta_{3}$ are unknown parameters. Also assume that $\operatorname{Var}\left(Y_{i}\right)=\sigma^{2}, i=1,2,3,4$. Then
(A) $\quad \theta_{1}$ is estimable
(B) $\quad \theta_{2}$ is estimable
(C) $\quad \theta_{3}$ is estimable
(D) $\theta_{1}, \theta_{2}$ and $\theta_{3}$ are estimable
33. The hazard rates of two life time random variables $T_{1}$ and $T_{2}$ with respective cumulative distribution functions $F_{1}(t)$ and $F_{2}(t)$ and probability density functions $f_{1}(t)$ and $f_{2}(t)$, are $\square_{1}(t)=3 t^{2}$ and $\square_{2}(t)=4 t^{3}, t>0$, respectively. Then,
(A) $\quad E\left(T_{1}\right)<E\left(T_{2}\right)$
(B) $\quad f_{1}(t)<f_{2}(t) \forall t>0$
(C) $\quad F_{1}(t) \geq F_{2}(t) \forall t>0$
(D) $\quad F_{1}(t)<F_{2}(t) \forall t>1$
34. Let $\left(X_{1}, Y_{1}\right), \ldots,\left(X_{n}, Y_{n}\right)$ be a bivariate set of n independent observations from $(X, Y) \sim B V N\left(\zeta, \eta, \sigma_{1}^{2}, \sigma_{2}^{2}, \rho\right)$, then the distribution of $(\bar{X}, \bar{Y})$ is
(A) $\quad B V N\left(\zeta, \eta, \sigma_{1}^{2}, \sigma_{2}^{2}, \rho\right)$
(B) $\quad B V N\left(\zeta, \eta, \frac{\sigma_{1}^{2}}{n}, \frac{\sigma_{2}^{2}}{n}, \rho\right)$
(C) $\quad B V N\left(\frac{\zeta}{n}, \frac{\eta}{n}, \frac{\sigma_{1}^{2}}{n}, \frac{\sigma_{2}^{2}}{n}, \rho\right)$
(D) $\quad B V N\left(\frac{\zeta}{n}, \frac{\eta}{n}, \frac{\sigma_{1}^{2}}{n}, \frac{\sigma_{2}^{2}}{n}, \frac{\rho}{n}\right)$
35. Let $g(F)$ be an estimable parameter of degree $m$, and let $X_{1}, \ldots, X_{n}$ be a sample of size $n, n \geq$ $m$. Corresponding to any symmetric kernel $T\left(X_{i_{1}}, \ldots, X_{i_{m}}\right)$ of $g(F)$, the one sample U-statistic for the sample is given by $U\left(X_{1}, \ldots, X_{n}\right)=\binom{n}{m}^{-1} \sum_{C} T\left(X_{i_{1}}, \ldots, X_{i_{m}}\right)$, where summationC is over all $\binom{n}{m}$ combinations of $m$ integers $\left(i_{1}, i_{2}, \ldots, i_{m}\right)$ chosen from $(1,2, \ldots, n)$. The variance of $U\left(X_{1}, \ldots, X_{n}\right)$ is
(A) $\quad \frac{1}{\binom{n}{m}} \sum_{c=1}^{m}\binom{m}{n}\binom{n-m}{m-c} \zeta_{c}$
(B) $\quad \frac{1}{\binom{n}{m}} \sum_{c=1}^{n}\binom{m}{n}\binom{n-m}{m-c} \zeta_{c}$
(C) $\quad\binom{n}{m} \sum_{c=1}^{m}\binom{m}{n}\binom{n-m}{m-c} \zeta_{c}$
(D) $\quad\binom{n}{m} \sum_{c=1}^{n}\binom{m}{n}\binom{n-m}{m-c} \zeta_{c}$
$\operatorname{Here}_{c}=\operatorname{cov}_{F}\left(T\left(X_{i_{1}}, \ldots, X_{i_{m}}\right), T\left(X_{j_{1}}, \ldots, X_{j_{m}}\right)\right)$
36. For the bivariate normally distributed random variable $(X, Y)$, Kendall's tau measure of association between X and Y is zero
(A) If and only if X and Y are independent
(B) If and only if X and Y are perfectly positive correlated.
(C) If and only if X and Y are perfectly negative correlated
(D) If and only if X and Y are perfectly correlated
37. A persistent state (say $k$ ) is said to be null, if its mean recurrence time is
(A) Finite
(B) Unity
(C) Infinite
(D) Zero
38. Let $X_{1}, X_{2}, X_{3}$ be three random variables such that $\rho_{12}=\rho_{13}=\rho_{23}=\rho \neq 1$, then the square of multiple correlation coefficient $R_{1.23}^{2}$ is
(A) $\frac{\rho^{2}}{1+\rho}$
(B) $\frac{1+2 \rho^{2}}{1+\rho}$
(C) $\frac{2 \rho^{2}}{1+\rho^{2}}$
(D) $\frac{2 \rho^{2}}{1+\rho}$
39. Let X be a continuous random variable with mean 2 and variance 9. Then $P\{|X-2| \geq 6\}$ is
(A) Bounded above by $1 / 4$
(B) Bounded below by $1 / 4$
(C) Bounded above by $1 / 2$
(D) Bounded below by $1 / 2$
40. The minimum number of equal length subintervals needed to approximate $\int_{1}^{2} x e^{-x} d x$ to an accuracy of at least $\frac{1}{3} \times 10^{-6}$ using the trapezoidal rule is
(A) 100
(B) $100 e$
(C) 1000
(D) $1000 e$
41. Relative efficiency of cluster sampling with cluster of equal size $M$ with simple random sampling without replacement, assuming that number of clusters $N$ is sufficiently large, is
(A) $1+\rho(M-1)$
(B) $1+2 \rho(M-1)$
(C) $\quad(1+\rho(M-1))^{-1}$
(D) $\quad(1+2 \rho(M-1))^{-1}$
42. In one classification model, consider the following analysis of variance table

| Sources of variation | d.f. | Sum of squares | F-statistic |
| :---: | :---: | :---: | :---: |
| Treatments | 2 | 250 | $\mathrm{~F}=7.5$ |
| Error | 12 | 200 |  |
| Total | 14 | 450 |  |

The coefficient of determination $R^{2}$ is
(A) $5 / 2$
(B) $5 / 9$
(C) $9 / 5$
(D) $2 / 5$
43. Let $f$ be a measurable function defined over a measurable set $E$. Then the function $-f$ (the negative of $f$ ) is
(A) Measurable
(B) May or may not be measurable
(C) Measurable if function $f$ vanishes no where on $E$
(D) Outer measurable
44. For normal underlying distribution, the asymptotic relative efficiency of Mann-Whitney U-test relative to two-sample $t$-test is
(A) 0.655
(B) 0.755
(C) 0.855
(D) 0.955
45. In a $3^{3}$ design, the error degress of freedom with 5 replicates are
(A) 108
(B) 106
(C) 104
(D) 102
46. The study state control limits for exponentially weighted moving average control chart with parameters, $\mu_{0}=0, L=3, \sigma=1$ and $\lambda=1$, are in the interval
(A) $\quad-4$ to 4 (B) $\quad-3$ to 3
(C) -2 to 2
(D) -1 to 1
47. A necessary condition for a symmetrical BIBD, assuming the treatments as even, is that $(r-\lambda)$ must be
(A) Positive integer
(B) Negative integer
(C) Either positive or negative integer
(D) Perfect square
48. If $A \sim W_{m}(n, \Sigma), n>0$ is a positive integer and $\underline{Y}$ is an $m \times 1$ random vector independently distributed of A with $P(\underline{Y}=\underline{0})=0$, then
(A) $\frac{\underline{Y^{\prime}} \operatorname{l} \underline{\underline{Y}}}{\underline{Y}^{\prime} \Sigma \underline{Y}} \sim \chi_{n}^{2}$ and is independent of $\underline{Y}$
(B) $\frac{\underline{Y}^{\prime} A \underline{Y}}{\underline{Y}^{\prime} \Sigma \underline{Y}} \sim \chi_{2 n}^{2}$ and is independent of $\underline{Y}$
(C) $\frac{\underline{Y^{\prime}} \boldsymbol{\underline { Y ^ { \prime } } \underline { \underline { Y } }}}{\underline{Y}^{\prime}} \sim \chi_{\frac{n}{2}}^{2}$ and is independent of $\underline{Y}$
(D) $\frac{\underline{Y}^{\prime} A \underline{Y}}{\underline{Y^{\prime} E \underline{Y}}} \sim \chi_{3 n}^{2}$ and is independent of $\underline{Y}$
49. For the Gauss Markoff model $\left(\underline{Y}, A \theta, \sigma^{2} I\right)$, with $A=\left(\begin{array}{lll}1 & 1 & 0 \\ 1 & 0 & 1 \\ 1 & 1 & 0\end{array}\right)$ and $\underline{\theta}=\left(\begin{array}{l}\theta_{1} \\ \theta_{2} \\ \theta_{3}\end{array}\right), l_{1} \theta_{1}+l_{2} \theta_{2}+$ $l_{3} \theta_{3}$ is estimable if
(A) $l_{1}=0, l_{2}=1, l_{3}=0$
(B) $l_{1}=0, l_{2}=0, l_{3}=1$
(C) $\quad l_{1}=1, l_{2}=0, l_{3}=0$
(D) $l_{1}=1, l_{2}=1, l_{3}=0$
50. The ability of a sampling plan to discriminate between lots of high quality and lots of low quality is described as
(A) An average outgoing quality curve
(B) A process control chart
(C) An operating characteristic curve
(D) A range chart

## Stem Cell Tissue Engineering \& Biomedical Excellence(Ph.D.)

1. Downregulation of different G-Protein receptor signalling by $\beta$-adrenergic receptor kinase represents which one of the following process:
(A) Homologous Desensitization
(B) Homologous Sensitization
(C) Heterologous Desensitization
(D) Heterologous Sensitization
2. Which one of the following does not represent as a growth medium?
(A) Leibovitz
(B) RPMI 1640
(C) Histopac
(D) DMEM
3. In order to perform cell line authentication which of the following methods is not used:
(A) DNA barcoding
(B) in situ hybridization
(C) Isoenzyme analysis
(D) Cytogenetic analysis
4. One of the following does not represent second messenger:
(A) cAMP
(B) cGMP
(C) $\quad \mathrm{cCMP}$
(D) Diacylglycerol
5. Surface immunoglobulin (Ig) is stripped from the surface of B-lymphocytes by all the proteolytic enzymes, except one of the following:
(A) Collagenase
(B) Pronase
(C) Chymotrypsin
(D) Trypsin
6. Overexpressing the niche milieu will produce which of the following:
(A) No effect in the local environment
(B) Apoptosis of the cells
(C) Cancer cell phenotype
(D) Differentiation of cells
7. Differentiation of one differentiated cell to other differentiated cell type is known as:
(A) Primitive cell differentiation
(B) Ontogeny
(C) Metaplasia
(D) Dysplasia
8. During drosophila development, number of nuclei bathing in the common cytoplasm is represented by which of the following:
(A) Cap cell
(B) Syntial blastoderm
(C) Follicle cell
(D) Trophoblast cell
9. Richest reservoir of mesenchymal stem cells is represented by which of the following:
(A) Bone Marrow
(B) Cord blood cells
(C) Adipose Tissue
(D) Wharton Jelly
10. One of the following can be used to generate the whole organism:
(A) Blastomere
(B) Dental pulp stem cells
(C) Blastocyst cells
(D) Bone marrow cells
11. The embryoid bodies can be generated from which type of the cells:
(A) Induced Pluripotent Stem Cells
(B) Mesenchymal stem cells
(C) Hemangioblast
(D) Niche support cells
12. All the following, except one is not related to feeder cells used during stem cell culture:
(A) Provide Extracellular Secretions for Growth
(B) Source of antitrypsin
(C) Layer of cells unable to divide
(D) Irradiated cell types
13. Pancreatic cell progenitor can be identified from the pool of population of stem cells by which of the following master regulator:
(A) PAX 1
(B) PDX
(C) GPCR
(D) RxR
14. When studying the antibody reaction on the cells grown on matrix which of the following is exercised:
(A) Use blocking reagent after addition of specific antibody
(B) Use blocking reagent before addition of specific antibody
(C) A combination of specific and non-specific antibody
(D) Only single antibody specific protein in question
15. Which one of the following can be exploited to trace the stem cells and separate them from rest of the nucleated cells?
(A) Fluorescent cell sorting analysis
(B) BrdU labeling studies
(C) ${ }^{14} \mathrm{C}$ labeling studies
(D) ${ }^{35}$ S labeling studies
16. One of the following characteristics that both mouse Embryonic stem cells (ES) and Embryonic germ cells (EG) possess is:
(A) High Alkaline phosphatase activity
(B) Low Alkaline phosphatase activity
(C) High Acid phosphatase activity
(D) Low Acid phosphatase activity
17. Addition of leukaemia inhibitory factor to the embryonic cell culture performs which of the following function:
(A) Augmentation of Apoptosis
(B) Augmentation of Differentiation
(C) Inhibition of differentiation
(D) Inhibition of Apoptosis
18. In order to find the interacting partner of Sox 2 protein for stem cell pluripotency, Sox2 vector was tagged with sequence coding for His-2 tag, its expression in stem cell was used to identify the interacting protein. Pinpoint the methodology that you would use to achieve it:
(A) Immuno-hybridization followed by southern blotting
(B) Immuno-precipitation and followed by Northern blotting
(C) South-Western blotting
(D) Immuo-pulldown followed by immuo-blotting
19. The deliberate creation of embryo for production of stem cells line that will not provoke any immune response and rejection is termed as:
(A) Positional cloning
(B) Genetic cloning
(C) Reproduction cloning
(D) Therapeutic cloning
20. Matrigel is a commonly used material for growing cell. Which of the following is not the component of Matrigel:
(A) Heparan sulfate (B) Laminin
(C) Collagen
(D) $\beta$-Catenin
21. Scaffold or materials that are permissive to bone formation but do not attract the osteo-progenitors that initiate bone formation is termed as:
(A) Osteo-conduction
(B) Osteo-induction
(C) Osteo-inhibition
(D) Osteo-porosis
22. Mesangioblasts are the stem cells associated with differentiation into:
(A) Mesothelial cells
(B) Vessel forming cells
(C) Bone forming cells
(D) Neural cell
23. One of the following process is used to generate all female derived embryonic stem cells by tricking the non-fertilized egg into duplication of DNA:
(A) Artificial insemination
(B) in vitro fertilization
(C) Fertilization
(D) Parthenogenesis
24. Which of the following biological assay was used first to detect hematopoietic stem cells:
(A) Cell migration assay
(B) Clonal subtraction assay
(C) Soft agar colony forming assay
(D) Spleen colony forming assay
25. Which one of the following best explains the progenitor cells:
(A) An immature cell committed to a particular differentiation lineage with limited proliferation
(B) A cell that is terminally differentiated with very low proliferation
(C) Cell isolated from inner cell mass of the embryo with very high proliferation rate
(D) Totipotent cell isolated at morula stage having very high proliferation
26. The term epigenetic is commonly used during stem cell imprinting, this term is best explained as:
(A) Mitotically /meiotically heritable change involving covalent modification of existing protein
(B) Mitotically /meiotically heritable change not involving changes in DNA sequence
(C) Mitotically /meiotically heritable change involving events at 5' upstream region of a gene
(D) Mitotically/meiotically heritable change due to mutation in promoter region
27. Method of removing trophectoderm of a blastocyst, using antibodies bound to surface antigen of trophectoderm and complement is represented as:
(A) Immuno-surgery
(B) Immuno-precipitation assay
(C) Immuno-rejection
(D) Immuno-sorbent assay
28. The Stem cells isolated from bone marrow were differentiated into myogenic lineage. The cells were then washed and loaded with fresh medium with no serum for a period of 8 hrs. When this medium was added on the fresh stem cells these cells also got differentiated to myogenic lineage. Such a medium is termed as:
(A) Condition medium of pluripotency
(B) Condition medium of myogenic cells
(C) Condition medium of stem cells
(D) Condition medium of bone marrow
29. Which of the following represents cocktail for generating induced pluripotent stem cell and are named as original Yamanaka factors:
(A) Oct4, Sox2, cMyc, and Klf4
(B) Oct4, Sox2, cMyc, and nanog
(C) Oct4, Sox2, nanog, and Klf4
(D) Oct4, nanog, cMyc, and Klf4
30. Which one of the following stem cell marker can be exploited to differentiate mouse embryonic stem cells from human embryonic stem cells:
(A) Nanog.
(B) SSEA-1
(C) Oct $3 / 4$
(D) Sox2
31. All the following except one, represent progenitor cells:
(A) Alveolar type 2 cells for lungs
(B) Blast cells for T \& B lymphocytes
(C) Satellite cells for muscle
(D) Oval Cells for liver
32. All the following differentiated cells could be derived from embryonic stem cells except one of the following:
(A) Adipocytic Cells
(B) Chondrocytic Cells
(C) Myocytic Cells
(D) Trophoectodermal Cells
33. The Stem cell sorting and identification required use of CD markers during FACS analysis, what does CD stands for:
(A) Combinatorial differentiation
(B) Cluster of differentiation
(C) Combined designation
(D) Chemotactic designation
34. During Flowcytometry, commonly used cell types called side population cell is named due to which of the following character:
(A) Cells lying on the side of Hoechst dye retaining cells
(B) Cells lying on the side of differentiated cell lacking Hoechst dye
(C) Cells displaced on the side of a culture plate before addition of Hoechst dye
(D) Cells positive for Hoechst dye obtained after trypsinization
35. Which of the following event occurs following mammalian fertilization:
(A) Inhibition of heterokaryon formation
(B) Loss of follicular cells
(C) Exit from the meiosis
(D) Entry to the meiosis
36. In newt, following amputation of forelimb the cells around the amputated area undergo mitosis, trans-differentiation and de-differentiation. Which of the following event precedes all these:
(A) Formation of blastema
(B) Formation of embryoid body
(C) Formation of trophoectoderm
(D) Formation of teratoma
37. One of the major difference between osteoblast and osteoclast cells is:
(A) Differentiation to osteocytes by osteoblast and chondrocytes differentiation by osteoclast cell
(B) Differentiation to myocytes by osteoblast and chondrocytes differentiation by osteoclast cell
(C) Bone resorption by osteoblast and bone formation by osteoclast cell
(D) Bone formation by osteoblast and bone resorption by osteoclast cell
38. The concentration of magnesium sulfate higher than its optimum, during PCR amplification, can lead to which of the following outcome:
(A) Increase in nonspecific amplification
(B) Increase in the specific amplification
(C) Increases the DNA denaturation and hasten the speed of amplification
(D) Increases the DNA denaturation and slowing down speed of amplification
39. For the identification of sequence that a transcription factor binds for the gene regulatory activity can be ascertained by which of the following technique:
(A) Mutagenic assay
(B) DNA finger printing assay
(C) Gel retardation assay
(D) Gene silencing assay
40. The procedure for selecting hybridoma cells requires HAT medium containing aminopterin. This is included in the medium because:
(A) It provides a precursor for purine metabolism
(B) It inhibits purine metabolism
(C) It provides a precursor for dihydrofolate reductase activity
(D) It inhibits dihydrofolate reductase activity
41. The location of Hematopoiesis during development in mouse is:
(A) Mesonephric to Metanephric axis
(B) Neural tube site
(C) Extraembryonic yolk sac
(D) Extraembryonic aorta-gonad-metanephric
42. Abzyme, is an important molecule designated as:
(A) RNA with catalytic activity
(B) Catalytical antibody
(C) Catalytic carbohydrate
(D) Nano-based Fe with catalytic activity
43. All except one of the following cannot be used to differentiate the cancer cell out of mixed cell population:
(A) Loss of telomerase function
(B) Loss of contact inhibition
(C) High nuclear to cytosol ratio
(D) Polyploidy in cells
44. Dideoxy nucleotides are used to achieve which of the following outcome:
(A) Sequencing of DNA
(B) Mapping of DNA
(C) Generation of clone contigs
(D) Amplification of DNA
45. The major source of the collagen synthesizing cells used in the tissue engineering studies is:
(A) Neural cells
(B) Endothelial cells
(C) Fibroblast cell
(D) Epithelial cells
46. Bisulfite sequencing methodology is used to ascertain which of the following:
(A) Methylation of CpG islands
(B) Acetylation of histones
(C) Phosphorylation of serine/threonine
(D) Deacetylation of histones
47. The inactivation of complement system of serum to be used during cell culture is achieved by which of the following method:
(A) Filtration through 0.2 micron filter
(B) Heat inactivation
(C) Charcoal treatment
(D) Autoclaving
48. The action of the antibody involves its specificity towards a particular antigen, which of the following contributes to antibody specificity:
(A) The variable region of the heavy and light chain
(B) The constant region of antibody
(C) The heavy chain of the antibody
(D) Hinge region of antibody
49. Non-adherent clusters of neural progenitor cells grown under in vitro and used for propagation of neural stem cells is represented by which of the following:
(A) Glial cell clusters
(B) Axonic cell clusters
(C) Neurosphere
(D) Neural tube
50. Which one of the following stem cell type escapes immunological barrier during transplantation in mouse:
(A) Mesenchymal stem cell
(B) Embryonic stem cell
(C) Hematopoietic stem cells
(D) Neural stem cells

## System Biology \& Bioinformatics(Ph.D.)

1. What is full form of RCSB?
(A) Research collaborator for structural bioinformatics
(B) Research collaboration for structural bioinformatics
(C) Research contributory for structural bioinformatics
(D) Research contribution for structural bioinformatics
2. What makes BLAST faster than FASTA?
(A) Processor speed of the computer
(B) Hash table lookup
(C) Database size
(D) E value
3. Gen bank file format is DNA centric report because of
(A) Coding region
(B) Decoded region
(C) Qualifier
(D) CDS
4. What is the difference between RefSeq and GenBank?
(A) RefSeq includes publicly available DNA sequences
(B) GenBank includes non redundant curated data
(C) GenBank sequences are derived from RefSeq
(D) RefSeq sequences are derived from GenBank
5. ASN. 1 is computer language which is?
(A) Cross-platform dependent
(B) Machine-readable only
(C) Human-readable and machine-readable
(D) Human-readable only
6. GI number in NCBI is
(A) Gen Info Identifier
(B) Gene Info Indent
(C) Gene information Identity
(D) Genome Identifier
7. The Pitch/turn of helix observed generally in B-DNA structures is
(A) 33.3
(B) 33.2
(C) 33.1
(D) 33.4
8. The major groove of B-DNA structures is
(A) Narrow
(B) Shallow
(C) Deep, narrow (D) Wide, shallow
9. What is VRML
(A) Visual Real Modeling Language
(B) Visual Reality Model Language
(C) Virtual Real Modeling Language
(D) Virtual Reality Modeling Language
10. Which of the following most accurately determined using molecular modeling?
(A) Molecular orbital energies
(B) Minimum energy conformation
(C) Electrostatic potentials
(D) Energy
11. The Phi-Psi scatter diagram is a
(A) Ramachandran plot
(B) Chi plot
(C) Scatter plot
(D) Prediction plot
12. Which factor distinguish between sensitivity in BLAST
(A) P value only
(B) E value only
(C) P \& E value
(D) P or E value
13. GENSCAN developed by
(A) Christopher Burge
(B) Christopher rode Samuel
(C) Christopher rode
(D) Christopher karlin
14. How Sequence filtering in BLAST help in fast database searching
(A) Searches for LCR
(B) Searches for Repeats
(C) Both A \& B
(D) None of the above
15. Using SEG and PSEG in BLAST increase
(A) Sensitivity
(B) Specificity
(C) Search result
(D) E value
16. Many genes involved in pathogenicity are located in defined continuous regions ("islands") of the genome and have different GC content from other portions of the genome, suggesting that they have arisen through
(A) Horizontal gene transfer
(B) Gene duplication and subsequent mutation
(C) Protection from methylation
(D) Increase in the rate of mutation of successive GC pairs
17. Parametric bootstrap differ from non parametric because
(A) It uses simulated replicate rather than pseudo replicate
(B) It uses simulated pseudo replicate rather than replicate
(C) It uses simulation parameter for search
(D) It do not uses simulation parameter for search
18. Mention the type of the following reaction
$\mathrm{C}_{12} \mathrm{H}_{22} \mathrm{O}_{11}+\mathrm{H}_{2} \mathrm{O} \rightarrow \mathrm{C}_{6} \mathrm{H}_{12} \mathrm{O}_{6}+\mathrm{C}_{6} \mathrm{H}_{12} \mathrm{O}_{6}$
(A) Synthesis
(B) Hydrolysis
(C) Dehydration
(D) Hydrogenation
19. Value of coefficient of kurtosis for a Normal distribution is
(A) Positive
(B) Negative
(C) Equal to one
(D) Equal to three
20. The Northern blotting technique depends on
(A)
Similarities between the sequences of probe DNA and experimental DNA
(B) Similarities between the sequences of probe RNA and experimental RNA
(C) Similarities between the sequences of probe protein and experimental protein
(D) The molecular mass of proteins
21. In an airline reservation system, the entities are date, flight number, place of departure, destination, type of plane and seats available. The primary key is
(A) Flight number
(B) Flight number + place of departure
(C) Flight number + date
(D) Flight number + destination
22. MEGA phylogenetic software uses
(A) Comupute synonymous and not nonsynonymous sites
(B) Comupute both synonymous and nonsynonymous sites
(C) Comupute only synonymous and nonsynonymous sites
(D) Comupute only nonsynonymous sites
23. PSI-BLAST \& BLAST?
(A) Different
(B) Same
(C) Search engine
(D) Certain Difference
24. What are the advantages of computer in CADD?
(A) Simple \& fast
(B) Simple, short \& fast
(C) Complex \& slow
(D) $\mathrm{A} \& \mathrm{~B}$
25. Kyte-Doolittle hydropathy plot are?
(A) Used by TGREASE
(B) Short Prediction
(C) Signaling
(D) 1 dimension plot
26. Which one is correct?
(A) 8 bit $=10$ byte
(B) $1024 \mathrm{~KB}=1 \mathrm{MB}$
(C) $1024 \mathrm{~GB}=10$ Terabyte
(D) $1024 \mathrm{~GB}=1$ Mega Byte
27. Starch content of potatoes can be increased by using a bacterial gene, known as
(A) Sucrose phosphate synthase gene
(B) ADP glucose pyrophosphorylase gene
(C) Polygalactouranase gene
(D) None of the above
28. Kimura 2-parameter for Mutational models for DNA determine:
(A) Transversions more likely than transitions
(B) Transitions more likely than transversions
(C) Transversions vs. transitions
(D) Identify all transitions
29. DOT PLOT uses
(A) Low window and high stringencies
(B) Low stringencies and high window
(C) Low window and low stringencies
(D) High window and high stringencies
30. Repeat and inverted repeats in DNA sequences are significantly identify by DOT PLOT
(A) Agree
(B) May be
(C) Strongly agree
(D) Not possible
31. Normal distribution require for statistically signification?
(A) 1572 change amino acid changes in PAM
(B) 71 groups of PAM
(C) Entropy of PAM
(D) None of the above
32. Full form of SAGE
(A) Serial analysis of gene expression
(B) Series analysis of gene expression
(C) Sorted analysis of gene expression
(D) Segregated analysis of gene expression
33. Significant of pair sequence alignment is not possible without scoring matrix.
(A) Statement is correct
(B) Statement is incomplete
(C) Statement is no universal
(D) Statement is justified
34. In prokaryotes, just before the cell divides, the two daughter genomes are attached side by side to the
(A) Cell membrane
(B) Replication origin
(C) Centromeres
(D) Equatorial plate
35. Charge-charge relationship of noncovalent interactions to the distance separating the interaction molecules is
(A) $1 / r$
(B) $1 / \mathrm{r}^{2}$
(C) $1 / r^{3}$
(D) $1 / \mathrm{r}^{4}$
36. LOD score
(A) Statistical estimation
(B) Establish linkage between two loci
(C) None
(D) $\mathrm{A} \& \mathrm{~B}$
37. What is the range of sample size used in SDS-PAGE methods.
(A) Micrograms
(B) Naonograms
(C) Picograms
(D) Milligrams
38. Who coined the term mitosis
(A) Strasburger
(B) Flemming
(C) Strasburger \& Flemming
(D) Clamming
39. Rotameric structures are?
(A) Different phi \& psi but differ in side chain
(B) Same psi but differ in side chain
(C) Same phi \& psi but differ in side chain
(D) Same phi but differ in side chain
40. Beta Globin fold consist of
(A) All alpha helices
(B) Eight alpha helices
(C) Four alpha \& four beta helices
(D) Four alpha \& three beta helices
41. Macro molecular thermodynamics do not deals with
(A) Pressure
(B) Entropy
(C) Free energy
(D) Surface
42. DH5 alpha Plasmid DNA is
(A) Positively supercoiled
(B) Negatively supercoiled
(C) Early supercoiled
(D) Supercoiled
43. Linking number of DNA describe
(A) Topology
(B) Conformation (C)
Model
(D) Flexibility
44. Phylogenetics analysis are not based on
(A) Homology
(B) Para logy
(C) Orthology
(D) Xenology
45. Which of these amino acids is highly conserved \& abounded in nature?
(A) Glycine
(B) Alanine
(C) Both
(D) None
46. Triad tools in molecular modeling consist of
(A) Force field, Parameter sets, molecular mechanism
(B) Force field, Minimization algorithm, Parameter sets
(C) Force field, molecular dynamics, parameter sets
(D) All of the above
47. Which is second generation force field
(A) MMFF
(B) MM Family
(C) AMBER
(D) AMBER 2.0
48. How many copies of mitochondria is present in an eukaryotic cell
(A) $\quad 10^{3}-10^{4}$
(B) $10^{2}-10^{3}$
(C) 1000
(D) $10^{6}-10^{7}$
49. Minimum energy conformation is always an active conformation of the model
(A) Statement is correct
(B) State is incomplete
(C) Statement is not justify
(D) None of the above
50. Which of them are rare amino acid
(A) Citrulune
(B) Selenoacedities
(C) 3,4-dihydroxy-phenylalanine
(D) 3,8-dihydroxy-phenylalanine

## (ZOOLOGY)

1. Which of the following is not part of the triad of epidemiology?
(A) Host
(B) Agent
(C) Environment
(D) Time
2. Cydippid larva is a notable feature of phylum
(A) Echinodermata
(B) Mollusca
(C) Porifera
(D) Ctenophora
3. Which tracheal system is present in dipteran pupae?
(A) Propneuistic
(B) Metapneuistic
(C) Holopneuistic
(D) Hemipneuistic
4. Which of the following nucleotide bases is never present in genetic code?
(A) Adenine
(B) Uracil
(C) Thymine
(D) Cytosine
5. Grass carp is
(A) Ctenopharyngodon idellus
(B) Cyprinus carpio
(C) Hypophthalmichthys molitrix
(D) Oreochromis mossambicus
6. In which sequence the following proteins are activated as prerequisite for ventralisation in Drosophila
(A) Snake, Spätzle, Easter
(B) Snake, Easter, Spätzle
(C) Spätzle, Snake , Easter
(D) Easter, Snake, Spatzle
7. Which of the following is not a G-Protein coupled receptor?
(A) Rhodopsin
(C) Nicotinic acetylcholine receptor
(B) Muscarinic acetylcholine receptor
(D) Thyrotropin receptor
8. The honey bee drones
(A) Have a mother and a father
(B) Have sisters but no brother
(C) Have no father but have a grandfather
(D) Have only grandmother and grandfather
9. In C. elegans, pharynx is generated by two sets of cells derived from
(A) ABp and E cells
(B) ABa and E cells
(C) ABa and EMS cells
(D) ABp and EMSs
10. The median dorsal arm of epicranial suture is termed
(A) Metopic suture
(B) Epistomal suture
(C) Fulturae
(D) Gula
11. Which of the following isoenzymes is the first indicator of myocardial infarction?
(A) $\quad \mathrm{CPK}_{1}$
(B) $\quad \mathrm{CPK}_{2}$
(C) $\mathrm{CPK}_{3}$
(D) $\quad \mathrm{CPK}_{4}$
12. Which of the following provides exact distance between two loci on a chromosome?
(A) Genetic map
(B) Physical map
(C) Linkage analysis between two loci
(D) Results of crossing over between two loci
13. Which of the following technique is used for determining the molecular weight of the biomolecules?
(A) High Pressure Liquid Chromatography
(B) Affinity Chromatography
(C) Size Exclusion Chromatography
(D) Ion Exchange Chromatography
14. Which of the following are all game fish?
(A) Salmo trutta fario, Schizothorax richardsonii and Tor putitora
(B) Salmo trutta fario, Schizothorax richardsonii and Cyprinus carpio
(C) Catla catla, Labeo rohita and Cirrhinus mrigala
(D) Ctenopharyngodon idellus, Cyprinus carpio and Barilus barila
15. Who is considered as the father of epidemiology?
(A) John Wisnar
(B)
Louis Pasteur (C)
John Snow
(D) John Smith
16. In vertebrate visual receptors which of the following secondary messengers plays a key role
(A) AMP
(B) $\mathrm{Ca}++$
(C) cGMP
(D) Phospholipids
17. Which of the following is a temporary endocrine gland in human body?
(A) Corpus cardiacum
(B) Corpus luteum
(C) Corpus striatum
(D) Corpus albicans
18. The naive $B$ cell contains which type of antibodies
(A) $\operatorname{IgE}$ and $\operatorname{Ig} A$
(B) $\operatorname{IgM}$ and $\operatorname{IgE}$
(C) IgM
(D) IgG1 and IgG2
19. What is thought to be the origin of the Alu RNA transposon
(A) It is thought to be derived from a retrovirus
(B) It is thought to be derived from a protein coding gene
(C) It is thought to be derived from a cellular non coding RNA molecule
(D) It is thought to be derived from a DNA virus
20. Gill nets are mostly used in
(A) In rivers where the water current is very fast
(B) In shallow waters where depth is not more than a feet
(C) In reservoirs and lakes where the water is stationary
(D) In torrential streams
21. Which of the following is an example of constitutive heterochromatin?
(A) Barr body
(B) Hox genes of insects
(C) rRNA transcribing region of genome
(D) Centromeric region of chromosome
22. Foetus belong to which type of transplant
(A) Xenograft
(B) Isograft
(C) Autograft
(D) Allograft
23. Edman degradation is used for the
(A) Determination of amino acid sequence from the N -terminal of a protein
(B) Determination of nucleotide sequence of the DNA
(C) Determination of amino acid sequence from the C-terminal of a protein
(D) Determination of nucleotide sequence of the RNA
24. The volume of blood left in the ventricle after ventricular systole is called
(A) Stroke's volume
(B) End systolic volume
(C) Cardiac volume
(D) End diastolic volume
25. In which of the following animals germ cells are not determined by material in egg cytoplasm?
(A) Nematodes
(B) Frogs
(C) Mammals
(D) Insects
26. In an analytical procedure, a mixture of proteins is subjected to electrophoresis, transferred to nitrocellulose and then probed by using a labeled antibody probe. This technique is
(A) Southern blotting
(B) Northern blotting
(C) Western blotting
(D) RT-PCR
27. The syndrome in humans in which an individual's somatic cells contain the XXX chromosomes is called
(A) Klinefelter's
(B) Turner's
(C) Down's
(D) Superfemale
28. Best site for taking biopsy for diagnosis of Trichinellosis is
(A) Deltoid muscle (B) Diaphragm
(C) Pectoralis major (D) Liver
29. Contraction of the external intercostals results in:
(A) Downward movement of the ribs and sternum
(B) Elevation of the ribs and sternum
(C) Downward movement of diaphragm
(D) Abdominal contraction
30. In humans major cause of Burkitt's lymphoma is:
(A) Translocation of part of chromosome 8 to chromosome 14
(B) Translocation of part of chromosome 9 to chromosome 22
(C) Reciprocal translocation of parts of chromosome 15 and 17
(D) Translocation of part of chromosome 18 to chromosome 14
31. Berlese filter chamber in the midgut of insects is a modification for
(A) Conserving water
(B) Rapidly removing water
(C) Sieving solid food particles
(D) Separating digestible and non-digestible materials
32. In new born infants, haemolytic disease can be limited by
(A) Administration of specific allergens
(B) Allergens of specific shots
(C) Administration of antihistamines
(D) Administration of anti-Rh antibodies
33. In vertebrate genes, transcription regulatory regions that contain CpG islands are inactivated by which CpG modification?
(A) Methylation
(B) Myristylation
(C) Phosphorylation
(D) Acetylation
34. Pad-like structures on ventral surface of the tarsal subsegments are
(A) Euplantulae
(B)
Basitarsus
(C) Empodium
(D) Basicoxite
35. The presence of the pharyngeal teeth is a key character of
(A) Catfishes
(B) Carps
(C) Eels
(D) Snakehead fish
36. A high concentration of fructose is present in the secretion of
(A) Prostate
(B) Seminal vesicle
(C) Epididymis
(D) Vas deferens
37. Karnovsky's fixative used in electron microscopy is
(A) Glutaraldehyde
(B) Paraformaldehyde
(C) Glutaraldehyde and Paraformaldehyde
(D) Osmium Tetraoxide and Uranyl acetate
38. In Epidemiology "Quarantine" refers to:
(A) Separation of the diseased individual
(B) Separation of the individual exposed to a disease
(C) Separation of the diseased individual and its family
(D) Treatment of the diseased individual
39. Which of the following gene is required for SOS DNA repair in prokaryotes?
(A) MutS
(B) $\quad U v r A$
(C) UmиD
(D) $\quad$ MutH
40. Detection of odour is mediated by
(A) Ungated Na channels
(B) Gated Na- channels
(C) Gated cation channels
(D) G-protein coupled receptors
41. Notopterus notopterus and Chitala chitala are commonly referred as
(A) Eels
(B) Carps
(C) Featherbacks
(D) Catfishes
42. Which of the following is not an actin binding protein?
(A) Formin
(B) Filamin
(C) Profilin
(D) Laminin
43. All of the following characteristics are seen in the stools in amoebic dysentery of
(A) RBCs in clumps
(B) Charcot-Leyden crystals
(C) Pyknotic bodies
(D) Ghost Cells
44. The best way to demonstrate whether nuclear DNA is being replicated in cell is by:
(A) Phase contrast microscopy
(B) Thin layer chromatography
(C) Autoradiography
(D) Gel electrophoresis
45. The situation in which the RBC infection with Plasmodium species is not eliminated by immune system or by therapy and their numbers in the RBCs begin to increase again with subsequent clinical symptoms is called
(A) Gametogony
(B) Secondary exoerythrocytic schizogony
(C) Replase
(D) Recrudescence
46. A type of egg tube in insects in which vitellarium contains eggs only
(A) Acrotropic
(B) Polytropic
(C) Meroistic
(D) Panoistic
47. Basic Keratins are type of:
(A) Type II intermediate filaments
(B) Type III intermediate filaments
(C) Type IV intermediate filaments
(D) Type V intermediate filaments
48. The Mitosis Promoting Factor (MPF), which propels an animal cell into division stage, has two essential sub units: one is catalytic and the other is regulatory sub unit of MPF. Which one of the following is the regulatory sub unit of MPF:
(A) Cyclin
(B) Cdk
(C) Proline
(D) Ubiquitin
49. What are cryptic splice sites?
(A) These are splice sites that are used in some cells but not in others
(B) These are splice sites that are always used
(C) These are splice sites that are involved in alternative splicing resulting in removal of exons from some RNA molecules
(D) These are the sequences within exons or introns that resemble consensus splicing signals, but are not true splice sites
50. Trochophore larva is present in both these classes
(A) Amphineura \& Cephalopoda
(B) Pelecypoda \& Scaphopoda
(C) Pelecypoda \& Amphineura
(D) Amphineura \& Scaphopoda

## Public Health(Ph.D)

1. What is Gini Index:
(A) Measure of distribution of Income among households between countries
(B) Measure of distribution of Income among households within a country
(C) Measure of distribution of Income among households between States
(D) Measure of distribution of Income among households within States
2. Corpulence index means
(A) Measurement of Obesity
(B) Measurement of BP
(C) Measurement of Cholesterol
(D) Measurement of Depression
3. AIDS-causing HIV identifies its target cell by recognizing
(A) Gated Channels in the membranes
(B) Carbohydrates of Glycocalyx
(C) Apoproteins in the coated pits of membranes
(D) Low density lipoproteins in cell membranes
4. What is the use of Johari Window
(A) Studying health effects due to pollutants
(B) Studying the interpersonal relations
(C) Studying epidemiology
(D) Studying common health problems
5. Salutogenesis is
(A) Is a type of disease
(B) Is a term used in Health System
(C) Is related with immunization
(D) Is a process of improvement in health
6. Which of the following would usually require the smallest sample size because of its efficiency?
(A) One stage cluster sampling
(B) Simple random sampling
(C) Two stage cluster sampling
(D) Quota sampling
7. Who introduced the term 'stateless society'?
(A) Paul Bohannan
(B) E. Evans-Pritchard
(C) A. Powell
(D) A. Giddens
8. What is the marriage of a widow to her deceased husband's brother termed as?
(A) Widow marriage
(B) Sororate
(C) Polyandry
(D) Levirate
9. What are the minimum number of newborns should be examined for calculating percentage of LBW babies:
(A) 500 babies
(B) 1000 babies
(C) 10000 babies
(D) 100000 babies
10. The term 'family size' refers to:
(A) Total number of female children born to a woman
(B) Total number of persons in a family
(C) Total number of children a woman has born at a point in time
(D) Average No of children a woman has born at a point in time in a community
11. Which of the following method is used for contraceptive efficacy:
(A) PERT Analysis
(C) NPP
(B) Life Table Analysis
(D) PPV
12. Which one is incorrect regarding, angle of the needle to the injection site in different route of injection during immunization:
(A) Intra-dermal injection: $15^{\circ}$ (BCG)
(B) Sub-cutaneous: $45^{\circ}$ (Measles, MMR)
(C) Sub- cutaneous: $45^{\circ}$ (DPT, DT, TT, Hepatitis)
(D) Intra-muscular: $90^{\circ}$ (DPT, DT, TT, Hepatitis
13. Which of the following is correct explanation of Gestational Age (SGA):
(A) As the $5 \%$ of Birth Weight at any given Gestational age
(B) As the $10 \%$ of Birth Weight at any given Gestational age
(C) As the $15 \%$ of Birth Weight at any given Gestational age
(D) As the $20 \%$ of Birth Weight at any given Gestational age
14. What is Cycle Beads?
(A) Emergency Contraceptive Pills (ECP)
(B) Statistical Method used in MCH
(C) Hormonal Contraceptive
(D) Natural family planning method
15. Pomeroy Technique is widely used in?
(A) Mini Pills
(B) IUD
(C) Tubal ligation
(D) Coitus interrupts
16. What is Catch-Up Growth?
(A) Under growth, according to WHO growth curve
(B) Higher than normal growth to recover in previous growth curve
(C) Onset of growth in a growth curve
(D) Birth injury
17. What are the recommendations of school health services for a school?
(A) 1 Privy for 50 Children and 1 Urinal/40 Children
(B) 1 Privy for 100 Children and 1 Urinal/60 Children
(C) 1 Privy for 50 Children and 1 Urinal/100 Children
(D) 1 Privy for 150 Children and 1 Urinal/80 Children
18. What is meant by 'Ring Vaccination'?
(A) Vaccine given around 100 meters of a detected case
(B) Vaccine given around 100 yards of a detected case
(C) Vaccine given around 200 meters of a detected case
(D) Vaccine given around 200 yards of a detected case
19. Which is the major principle of medical ethics that states that physicians and other medical professionals must act in the best interest of the patient?
(A) Justice
(B) Autonomy
(C) Non-malfeasance
(D) Beneficence
20. Xenograft is
(A) Transplant across species barriers
(B) A transplant of tissue from one to oneself
(C) Transplant between genetically identical (monozygotic) twins
(D) Transplant from one individual to another with a different genetic make-up
21. According to COTPA-2003 (Cigarette or any other tobacco product) no person shall sell, offer for sale, or permit the sale of Cigarette or any other tobacco product in any educational institution in an area within a radius of
(A) One hundred yards
(B) One hundred meters
(C) One hundred feet
(D) One hundredth of a mile
22. Which of the following is a nominal variable?
(A) Blood sugar level
(B) Blood Type (A, B, AB \& O)
(C) Hemoglobin level
(D) Severity (mild, moderate and severe)
23. Whenever the median is reported as the measure of central tendency of a continuous variable, the associated appropriate measure of dispersion is
(A) Range
(B) Variance
(C) Interquartile Range
(D) Standard Deviation
24. Which of the following properties is characteristic of a variable following normal distribution?
(A) High Skewness
(B) Small variance
(C) Large mean
(D) Symmetry about mean
25. An investigator wants to draw samples from a group of men and women separately at random. What sampling method would he choose?
(A) Stratified random sampling
(B) Simple random sampling
(C) Systematic Sampling
(D) Quota sampling
26. One way ANOVA is used
(A) To compare the means of more than two groups
(B) To compare several proportions
(C) To test for linear trend
(D) To compare ratio of two variances
27. The association between two categorical variables is best shown in a
(A) Scatter plot
(B) Bar chart
(C) Dot plot
(D) Cross - tabulation of two factors
28. The equivalent nonparametric test for paired ' $t$ ' test is
(A) Kruskal-Wallis test
(B) Wilcoxon-signed rank test
(C) Median test
(D) Sign test
29. The correlation between two continuous variables (r, pearson's correlation coefficient measures:
(A) The nonlinear relationship
(B) How mutually exclusive
(C) The strength of linear relationship
(D) The extent to which they are clustered
30. The aim of simple linear regression analysis is to
(A) Investigate a non linear relationship between two continuous variables
(B) Replace dots in scatter diagram by a straight line
(C) Measure extent of relationship between two variables
(D) Describe the relationship in straight line form that best describe it and enables prediction of one variable from the other
31. Which among the following study designs is effective, when the frequency of disease is low?
(A) Cohort
(B) Case control
(C) Ecological
(D) Descriptive
32. Which of the following sets of four numbers has the smallest possible standard deviation?
(A) $7,8,9,10$
(B) $5,5,5,5$
(C) $0,0,10,10$
(D) $0,1,2,3$
33. If your grade on a recent test was 72 th percentile. If 90 students write the test then approximately how many students received a higher grade than you?
(A) 18
(B) 65
(C) 5
(D) 72
34. Which of the following statement is NOT true about random sampling?
(A) It is reasonably accurate
(B) Free from personal biases
(C) Economical method of sampling
(D) Can be applied for all types of data collection
35. The best sampling method for sampling a population finite size:
(A) Area sampling
(B) Systematic sampling
(C) Purposive sampling
(D) Quota sampling
36. Which of the following best suit for 'Action Research'?
(A) It is an applied research
(B) It is a quantity research
(C) It is a survey research
(D) It is a population research
37. The 'write once read many' concept in the article is synonymous to which of James Martins concepts?
(A) Subject centered systems
(B) Database centered systems
(C) Task centered systems
(D) Multimedia centered systems
38. Which scale measures the psychological meanings of an attitude object using bipolar objectives?
(A) Likert
(B) Summated rating
(C) Semantic differential
(D) Constant sums
39. The research study carried out with the help of data points (either primary or secondary) is termed as:
(A) Conceptual
(B) Observational (C) Empirical
(D) Action
40. The primary objective of $\qquad$ is to provide insights into, and an understanding of, the problem confronting the researcher.
(A) Exploratory research
(B) Conclusive research
(C) Causal research
(D) Descriptive research
41. Population value is called $\qquad$ .
(A) Statistic
(B) Parameter
(C) Variable
(D) Core value
42. The report submitted when there is a time lag between data collection and presentation of result is called $\qquad$ .
(A) Thesis
(B) Interim report
(C) Summary report
(D) Article
43. The goal for universal EHR adoption is $\qquad$ _.
(A) 2011
(B) 2020
(C) 2015
(D) 2014
44. What are interactive computer programs designed to assist physicians and other health professionals with decision making tasks?
(A) Electronic health records
(B) Clinical decision support systems
(C) Computerized provider order entry
(D) Telemedicine
45. Health information management standards began with the establishment of the:
(A) HMMS
(B) AHIMA
(C) CAHIIM
(D) ASHIM
46. $\qquad$ is defined as the mobilization of healthcare information electronically across organizations within a region, community or hospital system?
(A) Health informatics
(B) Health information exchange
(C) Health information technology
(D) Health information management
47. $\qquad$ is a health information organization that brings together health care stakeholders within a defined geographic area and governs health information exchange among them for the purpose of improving health and care in that community?
(A) HIE
(B) NORC
(C) RHIO
(D) HIT
48. What does SNOMED-CT stand for?
(A) Systemic nomenclature of medicine and coding terms
(B) Systematised nomenclature of medicine and coding terms
(C) System novel of medicine and clinical terms
(D) Systematised nomenclature of medicine and clinical terms
49. To achieve effective integration of various electronic patient records in the NHS requires?
(A) Unique patient identification system
(B) Patient identification system
(C) Radical change to the way the NHS runs
(D) Unique patient billing system
50. Human metabonomics is the study of?
(A) Study of human metabolic responses to environmental changes
(B) Study of metabolic responses to drugs and diseases
(C) Study of metabolic responses to drugs, environmental changes and diseases
(D) Study of human metabolic responses to aging

## Biotechnology(Ph.D.))

## 1. Which activity defines Bioaugmentation

(A) Addition of nutrients at the site of pollution
(B) Degradation of toxic waste by resident microflora
(C) Sequestration of heavy metal ions by bacteria
(D) Introduction of specific bacterial species at the pollution site
2. Which of the following is not present in the cell wall of Gram negative bacteria
(A) Peptidoglycan (B)
Teichoic acid
(C) Lipid A
(D) Porins
3. Mark the incorrect combination
(A) Biosafety level 1 Bacillus subtilis
(B) Biosafety level 2 Mycobacterium tuberculosis
(C) Biosafety level 3 Bacillus anthracis
(D) Biosafety level 4 Ebola virus
4. Which of the following statements is incorrect
(A) Milk is pasteurized at $62.8^{\circ} \mathrm{C}$ for 30 minutes
(B) Pasteurization kills Coxiella burnetii
(C) Pasteurization temperature depends on the heat resistance of M.tuberculosis
(D) Pasteurisation increases the shelf life of milk
5. A polypeptide of 90 kDa will be coded by a gene of the size
(A) 0.245 kb
(B) 2.454 kb
(C) $\quad 24.54 \mathrm{~kb}$
(D) 1.5 kb
6. Which of the following form the apoptosome complex
(A) Cytochrome c, Apaf-1 and procaspase-8
(B) Cytochrome c, Bcl-2 and procaspase-9
(C) Cytochrome c, FasL and procaspase-9
(D) Cytochrome c Apaf-1, and procaspase-9
7. What does not hold true for Cre/loxP system used in generating transgenic animals
(A) It causes large scale deletions in the chromosomes
(B) It activates the transgene by removing DNA sequences flanked by loxP sites
(C) It can remove the selectable marker that is no longer needed
(D) It creates a more efficient way to integrate useful transgene into the chromosomes
8. Genetically modified papaya exhibits
(A) Herbicide tolerance
(B) Insect protection
(C) Virus resistance
(D) Early ripening
9. Bt brinjal cultivation has been banned in
(A) Canada
(B) Bangladesh
(C) India
(D) Philippines
10. Which of the following genetically modified crop is cultivated in maximum countries?
(A) Cotton
(B) Maize
(C) Rice
(D) Soyabean
11. National Green Tribunal has banned the use of plastic bag below the thickness of
(A) 40 micron
(B) 50 micron
(C) 60 micron
(D) 80 micron
12. A bacterial population growing exponentially at a specific growth rate of $1.25 / \mathrm{h}$ will have a generation time of
(A) 0.55 h
(B) 0.65 h
(C) 0.85 h
(D) 1.55 h
13. The promoter for heterologous gene hyperexpression in pET28a plasmid vector is
(A) $\mathrm{P}_{\mathrm{L}}$ promoter of lambda phage
(B) $\quad$ trp promoter
(C) T7 phage promoter
(D) lac promoter
14. In DH5alpha strain of E.coli which mutations allow better yields and stability of the recombinant plasmid
(A) endA1 and rec A1
(B) $\quad$ gyrA and $d n a \mathrm{~J}$
(C) rec A and dam
(D) supF 44 and rel E
15. Treatment of bacterial culture with Trizol and chloroform leads to separation of
(A) RNA and DNA in aqueous phase and proteins in organic phase
(B) RNA in aqueous phase,DNA in interphase and proteins and lipids in organic phase
(C) RNA in aqueous phase,DNA and proteins in interphase and lipids in organic phase
(D) RNA in the organic phase,DNA in interphase and proteins in aqueous phase
16. Craig Venter has not been associated with which of the following
(A) Sequencing of H.influenza genome
(B) Sequencing of C.elegans genome
(C) Developing Expressed Sequence Tags (ESTs)
(D) Synthesis of the minimal bacterial genome
17. In which of the following situations NGS would be most useful
(A) To genotype 100 DNA samples for a known SNP
(B) To study the differential gene expression of a tumor sample
(C) To detect common missense mutation in Sickle cell disease
(D) To detect gene for resistance to rifamycin in M.tuberculosis
18. Which pair of aminoacids has highest and lowest isoelectric point
(A) Glycine, serine
(B) Serine, glutamine
(C) Serine, lysine
(D) Lysine, glutamine
19. Which of the following pair is wrongly matched
(A) IFN $\alpha$
Leucocytes
(B) $\quad$ IFN $\beta$
Fibroblasts
(C) $\quad \mathrm{TNF} \alpha$
Activated T cells
(D) CEA Liver cells
20. What is not true for peroxisomes
(A) In plants and animal it is involved in converting fatty acids to sugars
(B) In liver cells it detoxifies toxic molecules
(C) It is self replicating membrane bound organelle
(D) It imports proteins and lipids from cytosol
21. Which of the following oncogenes is not involved in signal transduction
(A) $s r c$
(B) $k i$-ras
(C) $n-r a s$
(D) jun
22. Mark the correct statement
(A) Only helices are involved in DNA - protein interactions
(B) The many types of leucine zipper proteins can all form heterodimers with one another
(C) The strength and specificity of DNA protein interaction can be adjusted by changing the number of zinc fingers repeats
(D) Helix turn helix motif of bacterial gene regulatory proteins is often embedded in same structural context
23. Which of the following does not affect the stability of mRNA
(A) Reduction of polyA tail to 100 residues
(B) Removal of 3' tail leading to decapping at 5 'end
(C) Nucleotide sequence at $5^{\prime}$ UTR
(D) Pyrimidine rich residues at 3' UTR
24. Which technique would be suitable to determine the transcription binding sites on global scale
(A) Deletion mapping
(B) DNA foot printing
(C) Whole genome sequencing
(D) ChIP-seq
25. In E.coli errors left after proof reading are removed by
(A) MutS,MutL,MutH and DNA polymerase I
(B) MutS,MutL,MutH and DNA polymerase III
(C) MSH,MLH and DNA polymerase I
(D) Rec A,Rec BCD,UMC and DNA polymerase III
26. Puromycin which is commonly used in the studies of protein translation, causes
(A) Nonspecific binding of aminoacyl tRNA to A site
(B) Release of EF-G-GDP from ribosome
(C) Inhibition of peptidyl tranferase activity
(D) Exit of growing polypeptide chain from ribosome

## 27. Which statement is true for $X$ Chromosome

(A) Paternally derived X Chromosome is heterochromatized
(B) Maternally derived X Chromosome is heterochromatized
(C) Heterochromatized X Chromosome is reactivated during oogenesis
(D) Genes responsible for pigmentation are located on X Chromosome
28. If a set of four mice is immunized with high dose of an antigen,in the secondary response they will produce
(A) Low affinity,heterogenous antibodies mainly IgG
(B) Low affinity,heterogenous antibodies mainly $\operatorname{IgM}$
(C) High affinity,homoogenous antibodies mainly IgG
(D) High affinity,homogenous antibodies mainly IgM
29. Which of the following can activate classical complement cascade
(A) A single molecule of IgM bound to RBC
(B) IgM molecules circulating in the serum
(C) A single molecules of IgG bound to the surface of RBC
(D) Two molecules of IgG bound at a distance on the surface of RBC
30. In 2019, the UN climate summit is scheduled to be held in
(A) Germany
(B) India
(C) Brazil
(D) Chile
31. Kagali Amendment of the Montreal protocol relates to the regulation of global warming by phasing out
(A) CFCs
(B) HCFCs
(C) HFCs
(D) Methyl bromide
32. Highly acid frits are canned in which type of steel cans
(A) Type L
(B)
Type M
(C) Type MC
(D) Type MR
33. Bacterial spores can be inactivated by all except
(A) Ethanol
(B) Steam under pressure
(C) Formaldehyde
(D) Glutaraldehyde
34. Which type of microbial contamination is difficult to detect in continuous cell line
(A) Bacterial
(B) Fungal
(C) Yeast
(D) Mycoplasma
35. Serum in cell culture medium provides hormones which stimulate proliferation except one which inhibits growth and promotes differentiation
(A) VEGF
(B) FGF
(C) TGFß
(D) IGF II
36. In sauerkraut production, the sequence in which different bacteria do desirable fermentation is
(A) Leuconostoc mesenteroides followed by Lactobacillus curcumeris and then Lactobacillus pentoaceticus
(B) Lactobacillus curcumeris followed by Lactobacillus pentoaceticus and then Leuconostoc mesenteroides
(C) Lactobacillus lactis followed by Lactobacillus curcumeris and then Leuconostoc mesenteroides
(D) Leuconostoc mesenteroides followed by Lactobacillus pentoaceticus and then Lactobacillus curcumeris
37. Patients with celiac disease should not eat
(A) Peanuts
(B) Corn
(C) Rice
(D) Wheat
38. Which of the following food pathogens can survive at cold temperature
(A) Clostridium botulinum
(B) Campylobacter jejuni
(C) Listeria monocytogenes
(D) Bacillus cereus
39. Mark the incorrect statement
(A) Corn syrups add viscosity to confection due to its dextrin content
(B) Invert sugar retards crystallization of sucrose
(C) Sorbitol is less sweet than sucrose and adds no calories
(D) Aspartame is sweeter then sucrose and adds no calories
40. Diphtheria toxin which is tagged to monoclonal antibody for targeted cancer therapy acts by
(A) ADP-ribosylation of eEF2
(B) Adenylate cyclase cleavage
(C) Uncoupling of electron transport chain
(D) Blocking of transmission in peripheral nerves
41. Mark the incorrect statement
(A) Tandem mass spectroscopy is used to determine peptide sequence
(B) Glycosylation or phosphorylation of protein can be identified by MALDI
(C) Mass spectroscopy cannot analyze metabolome
(D) The time of flight (TOF) for ions is directly correlated with mass to charge ratio
42. Mark the incorrect statement
(A) Pyrosequencing involves sequencing by synthesis
(B) Next generation sequencing can be used for quantification of gene expression
(C) Deep sequencing of bisulfite treated DNA can be used to study global methylation in cancer
(D) Exome-sequencing is not suitable to detect mutations associated with tumor progression
43. Most proteins have net positive charge at
(A) High Ph
(B) Low pH
(C) In the presence of SDS
(D) In the presence of mercaptoethanol
44. Which statement regarding yeast replicative plasmid is not true
(A) It is a high copy number plasmid
(B) It does not contains $2 \mu$ plasmid origin of replication
(C) It does not show mendelian segregation
(D) It gives the most stable maintenance of cloned gene
45. What is best option to express heterologous proteins efficiently in E.coli when codon bias can be a limitation
(A) Using genetically modified host strain that expresses rare tRNAs
(B) Genetically engineer the foreign gene to incorporate codons for more abundant tRNAs
(C) Genetically engineer the gene to remove the codons for rare tRNAs
(D) Grow the genetically modified host cells for longer time to make protein with rare codons
46. Which of the following techniques is based on Forster Resonance Energy transfer (FRET)
(A) Multiplex PCR
(B) Taq man PCR
(C) Surface Plasmon Resonance
(D) Fourier transform infrared spectroscopy
47. Which statement is not true for CRISPR Technology
(A) It is a kind of adaptive immunity in bacteria
(B) It works exclusively in bacteria
(C) It is based on RNA guided sequence specific DNA cleavage
(D) It is a powerful tool for making knock outs
48. In reverse phase HPLC sample is purified by using $\qquad$ stationary phase and
$\qquad$
(A) Hydrophilic, non polar
(B) Hydrophilic, polar
(C) Hydrophobic, non polar
(D) Hydrophobic, polar
49. All the statements given below are correct except
(A) Sec pathway transports proteins from the cytoplasm to outside cell in two steps
(B) Transportation of inner membrane proteins require sequences specific for signal recognition particle (SRP)
(C) Only unfolded protein containing two arginines in their signal peptide go through TAT pathway
(D) TAT pathway is important for virulence in pathogens
50. Yeast two hybrid system cannot be used to study
(A) Protein - Protein interaction
(B) Protein - RNA interaction
(C) Loss of Protein - Protein interaction
(D) Loss of DNA- protein interaction $\boldsymbol{x}-\boldsymbol{x}-\boldsymbol{x}$

## Forensic Science(Ph.D.)

1. In forensic science the most reliable form of evidence is
(A) Blood
(B) Weapon
(C) Clothes
(D) Eye witness
2. Which of the following individuals is known as the "Father of Forensic Toxicology"?
(A) R. A. Riess
(B) Edmond Locard
(C) Calvin Goddard
(D) Mathieu Orfila
3. Who was first to advocate the application of science to criminal investigation?
(A) Hans Gross
(B) Alphonse Bertillon
(C) Paul Kirk
(D) August Vollmer
4. In India the first State Forensic Science laboratory was established at
(A) Calcutta, 1952
(B) Mumbai, 1955
(C) Delhi, 1952
(D) Hyderbabad, 1955
5. Which of the following is the preferred way to identify a deceased person?
(A) Visual inspection
(B) Tattoos
(C) Examination of teeth, surgical records
(D) Surgical scars
6. The state forensic science laboratory was established in Shimla in year
(A) 1986
(B) 1987
(C) 1988
(D) 1989
7. The term 'self-loading pistol' is strictly synonymous with which one of the following?
(A) Semi-automatic pistol
(B) Double-action revolver
(C) Single-action revolver
(D) Automatic pistol
8. Which one of the following is used to analyse for the presence of gunshot residues? (A) Microspectrophotometry
(B) Refractive index measurements
(C) Polarized light microscopy
(D) Scanning electron microscopy used in conjunction with energy dispersive X-ray analysis
9. Which of the following is characteristic of genuine signatures (as opposed to forged signatures)?
(A) Pen strokes with blunt ends
(B) Pen strokes with tapering ends
(C) Evidence of retouching
(D) Unnatural pen lifts
10. If the probability of being blood type A is $1 / 8$ and the probability of blood type O is what is the probability of being either blood type A or O ?
(A) $5 / 8$
(B) $1 / 8$
(C) $1 / 2$
(D) $1 / 16$
11. The phase in the growth cycle of an individual hair in which the hair is actively growing is known as the:
(A) Anaphase
(B) Catagen phase
(C) Telogen phase
(D) Anagen phase
12. The final breakthrough for the fingerprint method of personal identification was made by:
(A) Sir Francis Galton
(B) Joseph Faurot
(C) William Herschel
(D) Edward Henry
13. For most fingerprint examiners, the chemical method of choice for visualizing latent prints is:
(A) Silver nitrate
(B) Iodine
(C) Chlorate
(D) Ninhydrin
14. The three basic types of fingerprint pattern are:
(A) Aarches, loops and rings
(B) Loops, arches and whorls
(C) Whorls, arches and accidentals
(D) Whorls, accidentals and loops
15. Quadrant zone method of crime scene investigation is suitable for:
(A) Indoor crime scene
(B) Outdoor crime scene
(C) Automobile crime scene
(D) Blast site
16. Blood or buccal swabs for DNA analysis are to be taken from any consensual partner having sex with the victim within $\qquad$ hours following intercourse.
(A) 24 hours
(B) 36 hours
(C) 48 hours
(D) 72 hours
17. The acid phosphatase screening test is used to detect which physiological fluid?
(A) Sweat
(B) Semen
(C) Blood
(D) Saliva
18. Which of the following best defines rigor mortis?
(A) This is the body cooling after death
(B) Livor mortis
(C) Postmortem rigidity
(D) Postmortem medical examination
19. Which of the following physiological fluids would be expected to have a high concentration of the enzyme amylase?
(A) Urine
(B) Sweat
(C) Semen
(D) Saliva
20. Which of the following is not a presumptive test for blood?
(A) Acid phosphatase test
(B) Luminol test
(C) Kastle-Meyer test
(D) Leuco-malachite green test
21. Luminol reagent reacts with blood in the presence of peroxide to emit light by a chemical process known as:
(A) Fluorescence
(B) Chemiluminescence
(C) Coagulation
(D) Illumination
22. The condition characterised by the absence of sperm cells in the seminal fluid is known as:
(A) Hypospermia
(B) Hyperspermia
(C) Oligospermia
(D) Azoospermia
23. The determination of whether or not a substance is blood is best made by means of a preliminary color test such as the Kastle-Meyer color test, which uses the chemical:
(A) p 30
(B) Benzidine
(C) Precipitin
(D) Phenolphthalein
24. Corpora-basal index is used to determine:
(A) Age
(B) $\operatorname{Sex}$
(C) Race
(D) Stature
25. Properties of evidence that can be attributed to a common source with an extremely high degree of certainty is/are:
(A) Individual characteristics
(B) Referent characteristics
(C) Comparison characteristics
(D) Class characteristics
26. How much seminal fluid does the normal male release during an ejaculation?
(A) 2.5 ounces
(B) 6 pints
(C) 1 milliliter
(D) 2.5 to 6 milliliters
27. Cheiloscopy is the study of
(A) Lips
(B) Eyes
(C) Ears
(D) Nose
28. In the ABO system, blood group ' O ' is characterised by the:
(A) Presence of antigen $A$ and the absence of antigen $B$
(B) Presence of both antigen A and antigen B
(C) Presence of antigen O
(D) Absence of both antigen A and antigen B
29. The ideal place to record the temperature of dead body is
(A) Rectum
(B) Axilla
(C) Mouth
(D) Groin
30. Diatoms in bone barrow are seen in death due to:
(A) Strangulation
(B) Drowning
(C) Electrocution
(D) Asphyxia
31. Specimens for toxicological evaluation should be preserved in
(A) $10 \%$ Formaldehyde
(B) Alcohol
(C) Normal Saline
(D) Saturated solution of common salt
32. All of the following are items to be collected from a deceased's body and sent to the forensic laboratory except:
(A) Head and pubic hairs
(B) Ocular fluid
(C) Blood
(D) Fingernail scrapings
33. Based on a mathematical calculation by Victor Balthazard, the probability of two individuals having the same fingerprints is one out of:
(A) $1 \times 10$ to the 30th power
(B) $1 \times 10$ to the 60 th power
(C) $1 \times 10$ to the 100th power
(D) $1 \times 10$ to the 90th power
34. The common habit forming drug is
(A) Nicotine
(B) Alcohol
(C) Opium
(D) Heroin
35. Huffing is common in:
(A) Psychedelic abuse
(B) Solvent abuse
(C) Ether abuse
(D) Smack dependence
36. Horizontal ligature mark in the neck is seen in
(A) Throttling
(B) Hanging
(C) Choking
(D) Strangulation by ligature
37. Adipocere is formed due to:
(A) Putrefaction of soft tissues
(B) Dehydration of muscles and subcutaneous fat
(C) Hydrogenation of subcutaneous tissues
(D) Saponification of subcutaneous fat
38. The child would have $\qquad$ number of teeth at the age of 3 :
(A) 20
(B) 16
(C) 24
(D) 12
39. The following bones are useful in determination of stature:
(A) Fibula and Pisiform
(B) Radius and Talus
(C) Femur and Ulna
(D) Humerus and Hyoid
40. The technique commonly used for comparison of ink sample is:
(A) Gas chromatography
(B) Thin layer chromatography
(C) X-ray diffraction
(D) Infra red spectroscopy
41. The report of fake currency note is admissible as evidence in court under:
(A) IPC292
(B) IEA-292
(C) $\quad \mathrm{CrPC}-292$
(D) CPC-292
42. A file placed on a computer's hard disk drive by Websites the user has visited are called:
(A) Firewall
(B) Bookmarks
(C) Caches
(D) Cookies
43. Which of the following toxins comes from the castor oil plant?
(A) Strychnine
(B) Ricin
(C) Atropine
(D) Digitalin
44. Who propounded the 'principle of exchange'?
(A) Francis Galton
(B) Edmond Locard
(C) Hans Gross
(D) Alphonse Bertillon
45. A particular SNP locus can be A, C or T. How many genotypes can be expected to be found in the population?
(A) 3
(B) 6
(C) 12
(D) 9
46. A PCR reaction that continues for 30 cycles will produce approximately how many PCR products from a single template DNA molecule?
(A) 64
(B) 128,000
(C) Approximately 1 million
(D) Approximately 1 billion
47. The Y-STR's utility in the forensic sciences is that:
(A) The frequency of occurrence in the general population is very small
(B) It is shorter by six bases on the Y chromosome then the X chromosome
(C) Replication of the DNA takes less then one hour
(D) It originates only from a male donor of DNA
48. A genetic locus that is analyzed in many forensic and paternity testing laboratories is the human leukocyte antigen locus known as HLA-DQ alpha. There are four major alleles at this locus known as $1,2,3$, and 4 . How many different genotypes are possible for these four alleles?
(A) 8
(B) 10
(C) 12
(D) 16
49. Using the equation given below,

$$
W \approx \frac{d^{3}}{16}
$$

calculate the approximate diameter of the crater that would be expected to be produced by a bomb containing 210 kg of explosive. where W is the weight of the charge in kilograms and $d$ is the diameter of the crater in meters. Which is the correct answer?
(A) 5 m
(B) 20 m
(C) 15 m
(D) 10 m
50. The main technique used to analyse samples suspected of containing liquid fire accelerants is:
(A) Atomic absorption spectroscopy
(B) Gas chromatography
(C) X-ray diffraction
(D) Ultraviolet-visible spectroscopy

## Microbial Biotechnology(Ph.D.)

1. A bacterium weighs one femtogram and has a doubling time of 20 min . If this organism grows continuously for 24 hrs , with same doubling time, its biomass will be
A) $\quad 1.00 \mathrm{~kg}$
B) $\quad 100.00 \mathrm{~kg}$
C) $\quad 1000.00$ Tons
D) More than the weight of the earth
2. Use of agar as a solidifying agent for nutrient media was given by
A) F. E.Hesse
B) R. Koch
C) L. Pasteur
D) J. Lister
3. In S.aureus cell wall, penta-glycine joins two adjacent tetra peptides. This joining is between
A) D-Ala of one peptide and D-Lys of second peptide
B) D-Ala of one peptide and L-Lys of second peptide
C) L-Ala of one peptide and L-Lys of second peptide
D) D-Ala of one peptide and meso-DAP of second peptide
4. One of the components involved in flagellar movement is called Rotor. It is made up of
A) MS and C rings
B) MS and L rings
C) MS and P rings
D) $\quad \mathrm{P}, \mathrm{L}$ and C rings
5. Most microbes die if the internal pH drops/rises
A) Below 5.0-5.5
B) $\quad 6.0-6.20$
C) Above 7.5-.8.0
D) Above 7.6-8.3
6. Approximate conditions for moist heat killing of yeasts and molds are
A) 10 minutes at $80^{\circ} \mathrm{C}$
B) 30 minutes at $80^{\circ} \mathrm{C}$ minutes
C) 10 minutes at $70^{\circ} \mathrm{C}$
D) 30 minutes at $70^{\circ} \mathrm{C}$.
7. Using glyoxylate cycle, bacteria converts acetylCoA into an end product known as
A) Oxaloacetate
B) Succinate
C) Malate
D) Pyruvate
8. Pasteurization is a process of killing
A) Pathogenic microbes in milk
B) Undesirable microbes in milk
C) All vegetative forms of microbes in a given sample
D) Undesirable microbes in a given sample
9. The starting material for biosynthesis of Penicillin by Penicilliumchrysogenum is
A) D-alpha-Aminoadipic acid, L-Cysteine, L-Valine
B) D-alpha-Aminoadipic acid, L-Cysteine, D-Valine
C) L-alpha-Aminoadipic acid, D-Cysteine, L-Valine
D) L-alpha-Aminoadipic acid, L-Cysteine, L-Valine
10. Which of the product is synthesized with the simultaneous involvement of two microbial strains (Plus and Minus)
A) Monosodium glutamate
B) Glutamic acid
C) Beta-Carotene
D) Vitamin $\mathrm{B}_{12}$
11. Coagulation of milk is an important step in the production of Cheese. Coagulum is formed by the degradation of which type of casein.
A) Kарра
B) Beta
C) Alpha
D) Epsilon
12. One of the important SCPs is Spirulina maxima. For its biomass production, the source of Carbon as nutrient is
A) $\quad \mathrm{CO}_{2}$
B) Whey
C) Molasses
D) Rice straw
13. Which of the following molecule does not have enzymatic activity
A) Streptokinase
B) Urokinase
C) Tissue plasminogen activator
D) Nonase
14. For the synthesis of Riboflavin from GTP as the staring material, the sequence of enzymes is
A) Cyclohydrolase, Reductase, Deaminase, and Synthase
B) Cyclohydrolase, Deaminase, Reductase and Synthase
C) Oxidase, Deaminase, Reductase and Synthase
D) Oxygenase, Deaminase, Reductase and Synthase
15. What type of bonds in Starch are cleaved by Glucoamylase (EC 3.2.1.3)
A) Alpha 1,4 and alpha 1,6
B) Beta 1,4 and beta 1,6
C) Alpha 1,4 and beta 1,6
D) Beta 1,4 and alpha 1,6
16. Vanillin can be synthesized from Capsaicin using following enzymes
A) Pen G Acylase and Vanillyl alcohol oxidase
B) Pen G Acylase and Vanillyl alcohol reductase
C) Capinase and Vanillyl alcohol oxidase
D) Capinase and Vanillyl alcohol reducatse
17. Tick the most appropriate statement
A) Secondary Metabolites arise from majority of precursors from Primary Metabolites
B) Secondary Metabolites arise from limited number of precursors from Primary Metabolites
C) Secondary Metabolites arise from only 2-Carbon Primary Metabolites
D) Secondary Metabolites arise from only 6-Carbon Primary Metabolites
18. Indigo dye used for coloring jeans can be synthesized from Tryptophan using following enzymes:
A) Glycinase and Xylene oxidase
B) Tryptophanase and Phenol oxidase
C) Glycinase and Indi oxidase
D) Tryptophanase and Xylene oxidase
19. Which of the following bacteria is the most crucial for sustenance of life on Earth
A) Nitrogen fixing bacteria
B) Carbon fixing bacteria,
C) Phosphorus fixing bacteria
D) ATP generating bacteria
20. Which of the following organism has fastest speed of movement.
A) Mycobacterium tuberculosis
B) Escherichia coli
C) Man
D) Leopard
21. Shine-Dalgarno sequence is
A) Found in 23 S rRNA, that binds to purine rich sequence in mRNA
B) Found in 16 S rRNA, that binds to pyrimidine rich sequence in mRNA
C) Found in 5SrRNA, that binds to pyrimidine rich sequence in mRNA
D) Found in 28 S rRNA, that binds to purine rich sequence in mRNA
22. Type II Restriction endonucleases are enzymes that
A) Recognize specific sequences, cleave DNA within or near recognized DNA sequence and require $\mathrm{Mg}^{2+}$ as a cofactor
B) Restrict/destroy the endonucleases present in cytoplasm
C) Remove the bases from the ends of a linear DNA molecule to produce blunt ends; means to allow ligation of two DNA molecules
D) Degrade ssRNA genome of retroviruses
23. A 1500bp DNA can code for $\qquad$ amino acid long polypeptide chain.
A) 300
B) 333
C) 500
D)
433
24. Most common modification found in nucleotide bases of DNA
A) O-glycosylation
B) Methylation
C) Sulfation
D) Nitrosylation
25. RNA is chemically $\qquad$ reactive as compared to DNA due to $\qquad$
A) More, presence of extra hydroxyl group
B) Less, presence of extra hydrogen atom
C) More, presence of extra nitrosyl
D) Less, presence of benzene ring
26. $\beta_{2}$ microglobulin is associated with
A) MHC I
B) MHC II
C) $\quad \mathrm{TCR}$
D) $\operatorname{IgG}$
27. Clostridium difficileis often associated with
A) Antibiotic-associated diarrhoea
B) Tetanus
C) Endocarditis
D) Catheterinfections
28. If a microbial pathogen inhibits the expression of MHC-I on the surface of infected cells, the binding of which of the following receptors will be inhibited?
A) $\quad$ CD 4
B) $\quad$ CD 8
C) $\quad$ CD 16
D) $\quad$ CD 32
29. Which of the following diseases is an intoxication and not infection?
A) Staphylococcal food poisoning
B) Tetanus
C) Shigellosis
D) Diphtheria
30. A germline B lymphocyte possesses 250 distinct V region genes, 4 J region genes and 2 isotypic options for assortment of its light chain. How many distinct idiotypes can be produced when this sequence combines with one heavy chain?
A) 2000
B) 500
C) 8
D) 1000
31. Bacillary dysentery is commonly caused by
A) Clostridium spp.
B) Salmonella spp.
C) Entamoeba histolytica
D) Shigella spp.
32. Phagocytosis involves all these steps except
A) Cytoskeletal rearrangements
B) Binding of TCR-CD3
C) Oxidative killing
D) Recognition of PAMPs
33. Which of the following cells does not exhibit antibody-dependent cell-mediated cytotoxicity?
A) B cells
B) NK cells
C) Macrophages D) Neutrophils
34. Administration of anti-tetanus serum is an example of $\qquad$ immunity
A) Natural, passive
B) Natural, active
C) Artificial, passive
D) Artificial, active
35. Rheumatic fever is an autoimmune disease associated with $\qquad$ antigens
A) Staphylococcal cell wall
B) Mycobacterial cell wall
C) Fungal cell wall
D) Streptococcal cell wall
36. Which of the following microbes is an obligate intracellular parasite?
A) Mycobacteriumleprae
B) Mycobacterium tuberculosis
C) Shigellaflexneri
D) Clostridiumperfringens
37. The only carbohydrate which is not having any chiral carbon atom is
A) Glyceraldehyde
B) Erythose
C) Dihydroxyacetone
D) Erythrulose
38. Histones are rich in
A) Lysine
B) Alanine
C) Histidine
D) Lysine and Arginine
39. Which of the following methods would be most appropriate for sterilizing an antibiotics solution?
A) Dry heat sterilization
B) Microfiltration
C) Autoclaving
D) Desiccation
40. The reserved food materials of green algae is
A) Laminaria
B) Chrysolaminaria
C) Floridian starch
D) Starch
41. The magnitude of BOD of wastewater is related to
A) Bacterial count
B) Amount of organic materials
C) Amount of inorganic materials
D) Fungal count
42. Which were the investigators live at the same times
A) Koch and Pasteur
B) Darwin and Woese
C) Van Leeuwenhoek and Ricketts
D) Berg and Hooke
43. Yeasts cells are good source of
A) Vitamins A and B
B) Vitamins A and D
C) Vitamins B and D
D) Vitamins A and C
44. The floating and drifting microbes are called
A) Zooplanktons
B) Benthos
C) Planktons
D) Limnos
45. The major carrier of Salmonellosis are
A) Meat and eggs
B) Meat and fish
C) Eggs and fruits
D) Eggs and fruits
46. In which year "The Biological Diversity Act" was enacted in India
A) 2004
B) 2002
C) 2005
D) 2016
47. Which of the following is not a type of copyright work?
A) Engine Design
B) Musical Work
C) Literary
D) Art
48. The term for patent protection under Indian Patent Act is:
A) 15 Years
B) 60 Years
C) 20 Years
D) 35 Years
49. What prominent amendment was made in The Patent Act, in 2005
A) Parallel importation
B) Establishment of appellate board
C) Simplification of procedures
D) Product patent for inventions in all fields of technology
50. Which of the following category does not fall under industrial property rights
A) Patents
B) Trademarks
C) Trade secrets
D) Copyrights

## Microbiology

1. Histozoic are the organisms residing in
A) Colon
C) Tissues
B) Blood
D) Small Intestine
2. India was declared polio free country in
A) 2013
B) 2015
C) 2011
D) 2014
3. Microparasites are
A) Algae
C) Protozoa
B) Fungi
D) Insects
4. Protozoa belong to kingdom
A) Animalia
C) Monera
B) Protista
D) Plantae
5. The parasite eradicated from India is
A) Medinaworm
C) Threadworm
B) Hookworm
D) Pinworm
6. GRAS stands for
A) Generally regarded as safe
C) Generally registered as safe
B) Generally required safe
D) Generally recognized as safe
7. Probiotics are generally
A) Yeast
C) Enterobacter
B) Streptococci
D) Lactic acid bacteria
8. HSC represents
A) Human stem cells
C) Haem stem cells
B) Haemopoietic stem cells
D) Human secretory cells
9. Class I MHC molecules are expressed in
A) Antigen presenting cells
C) B cells
B) Nucleated cells
D) T cells
10. Respiratory bursts is a
A) Biochemical mechanism
C) Physiological mechanism
B) Microbicidal mechanism
D) Pathogenic mechanism
11. Emil Von Behring got nobel prize for his research work in
A) Anaphylaxis
C) Antitoxins
B) Toxins
D) Antibodies
12. Multiple Sclerosis occurs due to
A) Hypersensitivity
C) Autoimmune disorder
B) Sedentary life
D) Malnutrition
13. Urease is the potent enzyme produced by
A) Campylobacter
C) Helicobacter
B) Escherichia
D) Enterobacter
14. Late lactose fermenters are deficient in enzyme
A) $\quad \beta$ galactosidase
B) $\quad \beta$ galactosidase permease
C) $\quad \beta$ galactopermease
D) $\quad \beta$ galactoglucopermease
15. Toxoid is a
A) Polysaccharide
C) Proteinaceous compound
B) Glycoprotein
D) Glycolipid
16. ISCOM is a
A) Protein antigen + Quil A
C) Protein antigen + detergent
B) Protein antigen + Quil A + detergent D) Protein antigen + detergent + MF 59
17. Cider Vinegar is produced from
A) Molasses
C) Apple juice
B) Wheat
D) Barley
18. Enzyme saccharase acts on
A) Saccharin
C) Sucrose
B) Maltose
D) Fructose
19. Kefir is produced by
A) Lactic acid bacteria (LAB)
C) LAB + yeast
B) Yeast
D) Bifidobacteria
20. Curing is the process of preservation of meat by
A) Refrigeration
C) Salts
B) Lyophilization
D) Salts + refrigeration
21. Eukaryotic mRNAs are
A) Polycistronic
C) Monocistronic
B) Tricistronic
D) Bicistronic
22. In homofermentative process, the end product is always
A) Lactic acid + ethanol
C) Lactic acid alone
B) Lactic acid + water + carbon dioxide D) Lactic acid + traces of other products
23. Genetic material in viruses may be
A) RNA
C) RNA and DNA
B) DNA
D) RNA or DNA
24. Ribosome 55 s is found in
A) Eukaryotes
B) Prokaryotes
C) Mitochondria of vertebrates
D) Mitochondria in golgi bodies of eukaryotes
25. mRNA of prokaryotes are
A) Stable and long shelf life
C) Unstable and polycistronic
B) Stable and polycistronic
D) Stable and short shelf life
26. RNA polymerase I is located in
A) Nucleoplasm
C) Nucleolus
B) Cytoplasm
D) Nuclear membrane
27. Autoclave and porcelain filters were discovered by
A) Robert Koch
C) T Needham
B) Charles Chamberlain
D) Elie Metchnikoff
28. Importance of microorganisms in carbon and nitrogen cycle was discussed by
A) Louis Pasteur
C) Beijerinck
B) Lister
D) S N Winogradsky
29. In bioterrorism, the most lethal organism is
A) Yersinia
C) Bacillus
B) Vibrio
D) Varicella
30. World Health Day is celebrated every year in the month of
A) $\quad 7^{\text {th }}$ January
B) $\quad 7^{\text {th }}$ June
C) $7^{\text {th }}$ March
D) $\quad 7^{\text {th }}$ April
31. 16s rRNA has classified microbes into three domain and is given by
A) Carl Woese
C) Holben
B) Olsen
D) Car Louis
32. Second edition of Bergey's Manual of systematic bacteriology has
A) Four volumes
C) Three volumes
B) Five volumes
D) Six volumes
33. Genetic material in TMV is
A) ssDNA
C) dsRNA
B) $\quad \mathrm{dsDNA}$
D) $\operatorname{ssRNA}$
34. Intermediate host for filaria is
A) $\quad R a t$
C) Human
B) Culex
D) Anopheles
35. Fastest moving parasite is
A) Giardia
C) Trichomonas
B) Paramaecium
D) Trypanosome
36. Malaria is caused by
A) Tachyzoites
C) Sporozoites
B) Gametocytes
D) Merozoites
37. Saprozoic microorganism takes
A) Solid foods
C) Semi solid foods
B) Soluble foods
D) Solid and liquid foods
38. Amoebic cyst are formed in
A) Tissues
C) Soil
B) Colon
D) Ileum
39. The antibodies that crosses placental barriers is
A) $\operatorname{IgM}$
B) $\operatorname{IgG}$
C) $\quad \operatorname{Ig} A$
D) $\quad \operatorname{IgD}$
40. IgM is most effective antibody in neutralizing viruses because of its
A) High valency
C) Half life
B) High amount
D) Avidity
41. CDC stands for
A) Centre for Disease Control \& prevention
B) Centre for Disease Control \& protection
C) Centre for Disease control \& Progress
D) Centre for Disease control \& Program
42. Smallpox is eradicated even then virus is preserved in
A) Baltimore, USA
C) Rockfellar Institute, USA
B) Atlanta, USA
D) Texas University, USA
43. Peritrichous flagella is present in
A) E. coli
C) Pseudomonas
B) Vibrio cholera
D) Clostridium
44. Dienes phenomena is exhibited by
A) Klebsiella
C) Proteus
B) Escherichia
D) Serratia
45. Among different species of Klebsiella, which species is indole producing
A) K. pneumonae
C) K. oxytoca
B) K. aerogenes
D) K. ozaenae
46. Capsule of Klebsiella can be seen with
A) Simple staining
C) Zeil Neilson staining
B) Gram staining
D) Negative staining
47. Red/pink colonies are seen due to production of pigment by
A) E. coli
C) Salmonella
B) Shigella
D) Serratia
48. Pure culture concept was introduced by
A) Lister
C) Louis Pasteur
B) Robert Koch
D) John Tyndall
49. Black centered colonies are observed on DCA by
A) Shigella
C) Salmonella
B) Proteus
D) Serratia
50. Mission Indradhanush was launched on
A) $\quad 25^{\text {th }}$ December 2014
B) $\quad 25^{\text {th }}$ December 2016
C) $\quad 25^{\text {th }}$ December 2013
D) $\quad 25^{\text {th }}$ December 2012
